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1 **Silica burial enhanced by iron limitation in oceanic upwelling margins**

2

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9 **Models suggest that large swathes of ocean can be silica limiting for primary producers**
10 **(diatoms) and this imposes a cap on the biological uptake of carbon dioxide into the**
11 **ocean. Sedimentary burial of biogenic silica (opal) is the main oceanic silicon sink and is**
12 **distributed equally between iron-limited open ocean provinces and upwelling margins.**
13 **However, the margin silicon sink is mainly localised in the eastern Pacific while**
14 **contributions from other upwelling systems are comparatively small. Such large inter-**
15 **margin differences remain unexplained. Here we measure biogenic fluxes, diatom-**
16 **bound iron concentrations and silicon isotope ratios in sedimentary material from the**
17 **Gulf of California, an archetypal margin silicon sink. We suggest that transient iron**
18 **limitation during intense upwelling periods dramatically increases silica relative to**
19 **organic carbon export resulting in elevated silica burial -a view also supported by long**
20 **sedimentary records from the same basin. Global compilations further show that**
21 **hotspots of silicon burial in the ocean are all characterised by high silica to organic**
22 **carbon export ratios, a diagnostic trait for diatoms growing under Fe stress. Therefore,**
23 **we propose that prevailing conditions of silica limitation in the ocean is largely caused**
24 **by Fe deficiency imposing an indirect constraint on oceanic carbon uptake. (199 words)**

25 Diatoms, a group of phytoplankton with amorphous silica cell walls (opal), account for up to
26 40% of the marine primary production¹. Because their frustules are siliceous and contain no
27 calcium carbonate, they export carbon to the deep ocean as organic matter. Hence, their
28 relative abundance determines the organic to calcium carbonate ratio of settling biogenic
29 particles from the sea surface: the C rain-rate ratio². This parameter regulates the efficiency
30 of the biological pump of carbon by pushing the marine carbon cycle towards increased (high
31 C rain-rate ratio) or decreased (low C rain-rate ratio) CO₂ drawdown from the atmosphere³.
32 In ideal conditions, diatoms need biologically available silicic acid and nitrate with a ratio of
33 1:1 in the surface waters⁴. However, while silicic acid is generally above biological
34 requirements in the deep ocean relative to other macronutrients such as nitrate and phosphate,
35 thermocline waters fuelling biological productivity in most part of the ocean are depleted in
36 silicic acid⁵. This lead to the yield limitation of diatom production by silicic acid in the
37 surface ocean⁶⁻⁸, restricting the amount of biogenic silica produced and hence the efficiency
38 of the biological pump of carbon. Here we investigate the causes of this potentially
39 widespread Si limitation in the ocean.

40 Silicon is introduced to the ocean via river runoff and is removed by sedimentary burial of
41 mainly diatom opaline skeletal remains^{2,9}. The ocean waters are undersaturated with respect
42 to opal, and hence its burial efficiency is only 3% because of dissolution in the water column
43 and superficial sediments. On an average Si is recycled ~25 times with in the ocean before
44 permanent removal through sediment burial⁹. However, areas of high opal export also
45 display higher sedimentary Si burial rates as pore waters become rapidly supersaturated in
46 silicic acid, preventing further dissolution and increasing Si burial¹⁰. Shallow water column
47 and high sedimentation rates in margins further aid the burial of biogenic Si by reducing its
48 exposure to undersaturated seawater².

49

50 **Concurrence of Marine Si sinks and Fe limitation**

51 Marine Si sinks are highly localised with burial mostly occurring in the Southern Ocean (SO)
52 and the North Pacific (NP) for the open ocean ($3 \text{ Tmol Si.yr}^{-1}$)¹¹ and in continental margins
53 ($3.7 \text{ Tmol Si.yr}^{-1}$) influenced by upwelling (Supplementary Material section 2; Table 1).
54 Despite covering a small surface area, coastal upwelling areas represent the main Si sink in
55 ocean margins. However, not all comparably productive marginal environments are Si sinks.
56 This is often overlooked and remains unexplained^{12,13}. For instance, Si burial rates in the
57 Eastern Pacific and Gulf of California (GoC) upwelling regions are one to two orders of
58 magnitude higher than in the Arabian sea despite similar productivity levels¹²⁻¹⁴. Therefore,
59 understanding why some oceanic margins are particularly efficient at removing Si is
60 important in explaining silicic acid limitation in the surface ocean in view of the constraint it
61 imposes on the biological C pump.

62 Areas of high opal production usually support significant organic carbon (C_{org}) export² but
63 there are notable exceptions to this¹⁵ (Fig.1a & 1c). In the SO, parts of the NP and to a lesser
64 extent the equatorial Pacific, C_{org} export is generally modest despite high biogenic silica
65 fluxes. This results in very high biogenic silica to organic carbon (Si: C_{org}) ratios in the
66 settling particles (up to 10), in contrast to the rest of the ocean where the Si: C_{org} ratio is
67 generally below 1 (Fig.1c, and Supplementary Material section 1 Fig.S1). The areas of high
68 water column Si and Si/ C_{org} export in Figure 1b are also regions that support relatively large
69 sedimentary Si burial (Supplementary Material section 2; Fig.S3.) suggesting the close link
70 between water column and sediment fluxes. These large open ocean regions with high Si
71 exports are High Nitrate-Low Chlorophyll (HNLC) areas characterised by micronutrient
72 limitation. In the ocean interior, dissolved iron concentration is generally below biological
73 requirement relative to essential macronutrients leading to iron limitation in parts of the sea
74 surface where additional continental supply of Fe is not sufficient^{16,17}. The dearth of bio-

75 available Fe from continental sources limits primary production and nitrate utilisation in
76 HNLC regions^{18,19}. One direct consequence of this Fe deficiency is the depletion of silicic
77 acid in these HNLC regions owing to the high uptake ratio of Si(OH)₄ relative to carbon and
78 other macronutrients by diatoms growing under Fe stress^{20,21}. This over consumption of
79 silicic acid during diatom growth in turn leads to a high Si:C_{org} ratio of the settling biogenic
80 particles, thereby enhancing sedimentary Si burial. Similarly, in coastal areas the Si:C_{org} ratio
81 of settling particles varies greatly between margins. Continental margins that support high
82 biogenic Si fluxes and burial such as the Eastern Pacific and the GoC show elevated Si:C_{org}
83 ratios (1-4.5), whereas biogenic fluxes in the Arabian Sea and the Benguela upwelling system
84 display Si:C_{org} ratios < 1 (Fig.1c). Such very large variations cannot be simply explained by
85 differences in silicic acid to nitrate and phosphate ratios of thermocline waters that upwell
86 and fuel biological production. Further analyses of sediment trap data¹⁵ presented in Fig.1c
87 indicate that such a large contrast between margins cannot be explained solely by variations
88 in siliceous versus calcareous primary producers or the differences in the ability of these
89 skeletal remains to ballast organic carbon. Instead, the high Si:C_{org} fluxes (>1) are related to
90 high elemental Si:C ratios of the siliceous producers themselves (see Supplementary Material
91 section 1; Fig.S2). Notionally, unlike open ocean HNLC provinces, coastal upwelling areas
92 receive additional iron inputs from the shelf and continental sources and are not expected to
93 be iron-limited. However, recent evidence suggests that iron limitation can arise even in
94 coastal systems during the course of intense upwelling when these additional terrestrial Fe
95 inputs are not sufficient to meet consumption by biota²²⁻²⁵. Here, using a suit of diatom-bound
96 geochemical tracers, we demonstrate that high Si:C_{org} export ratios and biogenic silica burial
97 in some upwelling margins can be attributed to iron limitation. We focus on the GoC which
98 constitutes a large Si sink despite its small size and is widely regarded as the archetypal area
99 for the formation of diatomaceous sediments²⁶.

100

101 **Opal exports in the Gulf of California**

102 Biogenic silica fluxes and Si:C_{org} ratio of the settling particles recorded between 1990 and
103 1997 in a sediment trap from the GoC are shown in Figure 2a¹⁴. The Si:C_{org} ratio increases
104 drastically during winters, when intense upwelling conditions prevail, leading to extremely
105 high silica export to the sediment. Diatoms, the main siliceous producers, ideally need a
106 silicic acid to nitrate ratio of about 1⁴. With a nutrient ratio of 1.8 in the subsurface water,
107 silicic acid supplied to the biota is in excess relative to nitrate in the GoC²⁷. Yet Si(OH)₄
108 becomes completely utilized by the phytoplankton at the sea surface over the course of the
109 upwelling season during ‘normal’ years²⁷ (Supplementary Material 3). This apparent surfeit
110 silicic acid consumption in the photic zone and the elevated Si:C_{org} rain-rate ratio,
111 comparable to HNLC regions such as the SO, together point to increased SiOH₄ uptake by
112 diatoms, greater export of biogenic silica relative to C_{org} during the productive season, and
113 are diagnostic characteristics of the effect of iron limitation on diatoms²³.

114 This inference is supported by the near-complete depletion of dissolved Fe (dFe) in the
115 surface waters of the central region of the GoC and the low dFe:PO₄ ratios (generally <10⁻³)
116 in subsurface waters²⁸ falling well below the lower limit of the optimum subsistence ratio for
117 coastal diatoms and close to the limit for other phytoplankton groups (Supplementary
118 Material 4)^{17,29}. This suggests that the GoC can be iron-limited during upwelling seasons
119 when diatom production and biological consumption of both macro- and micro-nutrients
120 increases^{23,24}.

121 In addition to Si:C_{org} rain-rate ratio, we track seasonal changes in the biological availability
122 of dissolved Fe in the GoC using trace metal composition of isolated and cleaned diatom
123 frustules from the sediment trap samples collected between March 1996 and February 1997¹⁴
124 (Fig.2a). In the Supplementary Material (Section 3.3; Fig.S5), we show that Fe

125 concentrations in diatom frustules track iron availability in natural seawater. In the GoC,
126 Fe:SiO₂ ratios are elevated during the summer months (July and August) when the water
127 column is stratified due to weak, southerly winds and productivity is at its lowest^{14,27}. Iron
128 measurements in the surface water of GoC corroborate this finding and reveal high dFe
129 concentrations and dFe:macronutrient ratio in excess relative to biological requirement under
130 stratified conditions due to terrestrial Fe inputs²⁸. After stratification breaks down, Fe
131 concentrations in the diatom frustules decrease by a factor of 5 on average (Fig.2b). The
132 lowest Fe:SiO₂ ratios are generally associated with the highest Si:C_{org} ratio in settling
133 particles and elevated biogenic silica fluxes, typically >1500 mmole.m⁻².yr⁻¹. The initial
134 decrease in Fe availability (October) occurs in conjunction with increased biogenic carbon
135 fluxes (carbonate and C_{org}) at the beginning of the upwelling season but prior to the biogenic
136 silica peak, further documenting the sequence of events in the basin: increased biological
137 consumption of iron in the surface and upwelling of Fe deficient water from the subsurface
138 trigger Fe limitation, which in turn causes a dramatic increase in Si export as a result of
139 increased Si:C uptake ratio by diatoms. Fe availability remains low during late winter-early
140 spring and is only replenished in the summer when the water column becomes stratified and
141 biological production reaches a minimum. Therefore, we suggest that conditions of transient
142 iron limitation in the GoC fostered by high biological productivity following intense
143 upwelling events result in high biogenic Si and Si:C_{org} fluxes^{17,20} causing the GoC to operate
144 as an enhanced Si sink²⁷.

145 It is important to recognize that in this scenario, Fe limitation is caused by intense upwelling
146 of waters that are deficient in Fe relative to other macronutrients such that the supply of Fe
147 from continental sources are insufficient to raise Fe levels to meet biological uptake. Thus,
148 dampened upwelling should reduce the amount of micro-nutrient required from terrestrial
149 sources to consume upwelled macronutrients. This in turn could lift Fe limitation. Such

150 conditions occur during El Niño episodes, such as the 1991 event, in the Gulf of California
151 (Fig.2a). As seen in Figure 2 (a & c), biogenic Si fluxes and the seasonal sharp increase in the
152 Si:Corg ratio of settling particles are much less pronounced during El Niño than during
153 intense upwelling years. In addition, excess silicic acid has been reported in the GoC during
154 El Niño events, once again reflecting reduced Si uptake relative to other macronutrients²⁷. As
155 a corollary, one can envision that climatic periods characterised by dampened winter
156 upwelling conditions will see the relaxation of Fe limitation in the basin, a significant
157 reduction in Si burial and the build up of unutilised silicic acid at the sea surface.

158

159 **Pleistocene opal burial in the Gulf of California**

160 We examine this scenario using high-resolution sedimentary records from the GoC
161 (Fig.3). Drastic declines in Si burials rates have been reported during Heinrich events-cold
162 climatic periods in the Northern Hemisphere when eastern Pacific margins including the GoC
163 experience reduced upwelling³⁰. In particular, we explore whether such reduction in biogenic
164 Si burial is matched by increased Fe biological availability and excess photic zone silicic acid
165 by combining estimates of diatom-bound Fe and Si isotope ratios in Core MD02-2515. These
166 long-term records suggest that the GoC has been operating in two modes with respect to iron
167 status (replete vs. limited) during the last 26 kyrs. Periods of high Si burial and Si/Corg ratios
168 at the core site correspond to periods of low diatom Fe/SiO₂ ratios and elevated $\delta^{30}\text{Si}$
169 (~1.2‰) On the other hand, the onsets of Heinrich events at 17 and 12.5 kys BP are
170 characterised by decrease in Si burial and Si:Corg ratios, low $\delta^{30}\text{Si}$ and high Fe/SiO₂. The
171 $\delta^{30}\text{Si}$ of dissolved silicic acid supplied to the GoC are estimated to be around 1.2‰^{31,32}. Using
172 a Raleigh fractionation model in a closed system (Supplementary Material section 4.2), we
173 calculate that episodes of high opal burial correspond with periods of complete silicic acid
174 utilisation in the GoC by biota²⁷. Given that silicic acid to nitrate ratio is about 1.8 in the

175 GoC²⁷, complete silicic acid utilisation in upwelled waters requires diatom uptake of these
176 nutrients in proportions >1 which is typical of iron limited conditions^{20,22,33}. This is evident
177 from the low Fe:SiO₂ recorded by diatoms during these periods. In contrast, Heinrich events
178 were matched by 2- to 3-fold higher diatom-bound Fe/SiO₂ than the rest of the record,
179 suggesting that Fe biological availability increased during diatom growth in these intervals.
180 Assuming silicate to nitrate ratio of upwelling waters remained at 1.8, to achieve ~50% silicic
181 acid relative utilisation diatoms should have consumed silicic acid and nitrate at a ratio close
182 to 1. This uptake ratio of 1 suggests diatom growth under Fe replete conditions and diatom
183 opal yield only limited by N. Therefore, the $\delta^{30}\text{Si}$ values measured during Heinrich events are
184 consistent with the absence of Fe limitation. Importantly, these results suggest that the
185 declines in Si burial during Heinrich events resulted from reduced Si uptake relative to other
186 macronutrients by diatoms growing under Fe replete conditions, leaving surplus silicic acid at
187 the surface of GoC. These conditions are similar to observations during modern El Niño
188 episodes in the GoC²⁷. This case study illustrates the switch from N limitation during
189 dampened upwelling (El Niño) to Si limitation during intense upwelling periods (eg. La
190 Niña) due to increased diatom Si(OH)₄:NO₃⁻ uptake ratios under Fe stress. Additionally, our
191 sedimentary records suggests that intense upwelling conditions increase the biological
192 consumption of Fe and the resultant transient Fe limitation is the key factor that enhances
193 biogenic silica burial rates in the GoC.

194

195 **Implications for the Marine Silicon Budget**

196 The episodic nature of the upwelling process and the strong biological requirement for dFe
197 imposed on restricted time and spatial scales make upwelling margins prone to transient Fe
198 limitation, although at variable frequency and intensity subject to local conditions causing
199 large differences in Si burial between upwelling margins. Upwelling margins such as the

200 eastern Pacific with high opal burial also show high Si/Corg export in the water column
201 (Fig.1a and also see Supplementary Material section 2; Fig.S3). This suggests Fe limitation is
202 a key factor enhancing opal burial in these margins as our case study in the GoC illustrates.
203 Eastern Pacific margins, where previous studies have reported transient Fe limitation, are an
204 important contributor to the marine Si sink and could possibly account for the burial of 1.5
205 Tmol Si /year or ~25% of the marine Si sink. This is comparable to the largest open ocean Si
206 sinks in the North Pacific and the SO (table 1, Supplementary Material section 2). We
207 estimate from our global compilation that between 60 to 75% of the total Si sink may occur
208 in open ocean and margins where Fe limitation is prevalent. Without these Si burial hotspots,
209 oceanic Si limitation^{5,6} would be much diminished. Thus our study reveals the intrinsic link
210 between Fe stress and Si removal from the ocean and the indirect constraint imposed by Fe
211 on the C rain-rate ratio and oceanic CO₂ uptake by fostering Si deficiency in the ocean.

212 (2499 words)

213 **Methods**

214 Silicon isotope determination has been conducted in ETH Zürich. A small amount of
215 biogenic opal (0.5 mg) was dried down with concentrated perchloric acid at ~180°C in
216 Teflon© vials, and then dissolved in 100 µl of 1 M NaOH, before being diluted to 5 ml with
217 0.01 M HCl after 24 hours. Equivalent to 10 µg of opal was loaded onto a pre-cleaned 1.8ml
218 DOWEX 50W-X12 cation exchange resin bed (in H⁺ form) and eluted with 5 ml of purified
219 water (Milli-Q element 18.2 MΩ.cm⁻¹). The Si isotope composition was determined on the
220 diluted solution (0.6ppm Si) on the Nu1700 high-resolution MC-ICPMS at ETH Zürich,
221 using a standard-sample-standard bracketing protocol^{34,35}. All results in this study were
222 calculated using the δ³⁰Si notation for deviations of the measured ³⁰Si/²⁸Si from the
223 international Si standard NBS28 in parts per thousand (‰). The long-term reproducibility

224 was better than 0.07‰ $\delta^{30}\text{Si}$ (1 s.d.). Samples were measured at least 5 times, which resulted
225 in a 95 % confidence level below 0.08‰.

226 The trace metal measurements on cleaned diatom frustules^{36,37} were performed on an Ion
227 Microprobe Cameca ims-4f at the NERC (National Environment Research Council)
228 Microprobe Facility located at the School of Geosciences, University of Edinburgh. The
229 clean diatom samples (2 mg) were pressed into an indium foil and analysed several (10-12)
230 times for up to 10 cycles with the ion microprobe. We used a 10nA primary beam, 25 micron
231 image field and energy filtering (75eV). The metal:SiO₂ values represent the average of the
232 10-12 measurements made for each sample. (253 words)

233

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243 diatom-bound trace metals, L.E.P. measured elemental composition and silicon isotopes,
244 W.G and R.T. provided samples for diatom-bound trace metal measurements, L.E.P. and
245 R.S.G. wrote the paper with the participation of W.G. All authors were involved in the
246 discussions of the results and commented on the manuscript.

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351 **Figure 1: The distribution of biogenic silica fluxes and molar biogenic silica (Si) to**
352 **organic carbon (Si:Corg) ratios shows the variability in Si:Corg ratio amongst a wide**
353 **range of oceanic provinces (a & b):** ODV maps and shading are derived from JGOFS¹⁵
354 (black dots) and the Guaymas basin, Gulf of California¹⁴ (GoC, black star) sediment trap
355 data. High-Nitrate, Low Chlorophyll regions (dotted white ellipses) and North East Pacific
356 margin display higher Si:Corg export ratio than the surrounding open ocean. **(c)** Corg
357 normalised biogenic Si and inorganic carbon (Cinorg) ratios in JGOFS sediment traps.

358 **Figure 2: Biogenic fluxes, Si:Corg ratio and diatom-bound Fe in the GoC sediment trap**
359 **(a)** Si fluxes and Si:Corg ratios drastically increase during winter upwelling (blue stripes).
360 **(b)** A close up on year 1996-1997 shows that Fe:SiO₂ (ppm, grey bars) is high during the
361 stratified period and decreases during the upwelling season when inorganic (green) and
362 organic carbon (black) fluxes increase. **(c)** 1990-1997 average biogenic flux changes in the
363 GoC. Winter and upwelling conditions show elevated Si fluxes and Si:Corg ratios relative to
364 summers and El Niño winter. Relatively constant inorganic carbon (Cinorg) suggests Si:Corg
365 variations are not related to calcareous production.

366 **Figure 3: Multiproxy reconstruction of silica cycling and iron biological availability in**
367 **the GoC over the last climatic cycle in core MD 02-2515 (27°53'N, 111°40'W) illustrates**
368 **the long-term impact of Fe limitation on Si burial.** Diatom-bound Fe **(a)** increases at the
369 onset of Heinrich event 1 and the Younger Dryas period (yellow stripes), in conjunction with
370 decreasing silicic acid relative utilisation ($\delta^{30}\text{Si}$, **(b)**)²⁷, biogenic silica concentration (weight
371 %) **(c)**, Si:Corg molar ratio **(d)**.