Dissolved organic carbon (DOC) in Arctic ground ice

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Abstract. Thermal permafrost degradation and coastal erosion in the Arctic remobilize substantial amounts of organic carbon (OC) and nutrients which have accumulated in late Pleistocene and Holocene unconsolidated deposits. Permafrost vulnerability to thaw subsidence, collapsing coastlines and irreversible landscape change are largely due to the presence of large amounts of massive ground ice such as ice wedges. However, ground ice has not, until now, been considered to be a source of dissolved organic carbon (DOC), dissolved inorganic carbon (DIC) and other elements which are important for ecosystems and carbon cycling. Here we show, using biogeochemical data from a large number of different ice bodies throughout the Arctic, that ice wedges have the greatest potential for DOC storage, with a maximum of 28.6 mg L⁻¹ (mean: 9.6 mg L⁻¹). Variation in DOC concentration is positively correlated with and explained by the concentrations and relative amounts of typically terrestrial cations such as Mg²⁺ and K⁺. DOC sequestration into ground ice was more effective during the late Pleistocene than during the Holocene, which can be explained by rapid sediment and OC accumulation, the prevalence of more easily degradable vegetation and immediate incorporation into permafrost. We assume that pristine snowmelt is able to leach considerable amounts of well-preserved and highly bioavailable DOC as well as other elements from surface sediments, which are rapidly frozen and stored in ground ice, especially in ice wedges, even before further degradation. We found that ice wedges in the Yedoma region represent a significant DOC (45.2 Tg) and DIC (33.6 Tg) pool in permafrost areas and a freshwater reservoir of 4200 km². This study underlines the need to discriminate between particulate OC and DOC to assess the availability and vulnerability of the permafrost carbon pool for ecosystems and climate feedback upon mobilization.

1 Introduction

Vast parts of the coastal lowlands of Siberia, Alaska and Canada consist of unconsolidated organic-rich, fine-grained deposits. These sediments, which occur as glacigenic and Yedoma-type sediments (including their degradation forms as thermokarst), are characterized by high ground-ice contents, both on a volumetric (vol %) and gravimetric (wt %) basis (Brown et al., 1997; Zhang et al., 1999; Grosse et al., 2013; Schirrmeister et al., 2013). Yedoma deposits, which formed during the late Pleistocene cold stages in unglaciated Beringia (Schirrmeister et al., 2013), for instance, are characterized by absolute ground-ice contents, excluding ice wedges, of 40–60 wt % (Schirrmeister et al., 2011c). Ice wedges are one of the most common types of ground ice in permafrost. They form when thermal contraction cracks open in winter, which are periodically filled with snow meltwater in spring that quickly (re)freezes at negative ground temperatures to form ice veins and finally vertically foliated ice wedges. The ice wedges are themselves characterized by volumetric ice contents approaching 100 vol % and make up much of the subsurface in these Yedoma deposits. Recent calculations of ice-wedge volumes in east Siberian Pleistocene Yedoma and Holocene thermokarst deposits show contents of 48 and 7 vol %, respectively (Strauss et al., 2013). Combining ice wedges and other ice types in Yedoma deposits gives a mean volumetric ground-ice content for those regions between 60 and 82 vol % (Zimov et al., 2006a, b; Schirrmeister et al., 2011b, c; Strauss et al., 2013). High ground-ice
contents are also typical for coastal Alaska (43–89 vol%; Kanevskiy et al., 2011, 2013) and the western Canadian Arctic (50–60 vol%; French, 1998). The presence of massive ice (i.e., gravimetric ice content > 250% on dry soil weight basis; cf. van Everdingen, 1998) and excess ice, which is visible ice that exceeds the pore space, is the key factor for the vulnerability of permafrost to warmer temperatures and mechanical disturbance, as ice melt will initiate surface subsidence and thermal collapse, also known as thermokarst (Czudek and Demek, 1970).

Permafrost soils hold approximately 50% of the global soil carbon pool (Tarnocai et al., 2009; Hugelius et al., 2014), mostly as particulate organic carbon (POC). These calculations of permafrost OC stocks, however, subtract the ground-ice content (Zimov et al., 2006a, b; Tarnocai et al., 2009; Strauss et al., 2013; Hugelius et al., 2013, 2014) and therefore disregard the OC, especially the amount of dissolved organic carbon (DOC), contained in large ground-ice bodies such as ice wedges and other types of massive ice. Although these numbers might be small compared to the POC stocks in peat and mineral soils, DOC from permafrost is chemically labile (Dou et al., 2008; Vonk et al., 2013a, b) and may directly enter local food webs. Due to its lability, DOC can become quickly mineralized by microbial communities and photochemical reactions (Battin et al., 2008; Vonk et al., 2013a, b; Cory et al., 2014) and returned to the atmosphere when released due to permafrost degradation (Schuur et al., 2009; Schuur and Abbot, 2011).

Several studies have shed light on the POC stocks contained in permafrost (e.g., Zimov et al., 2006a; Tarnocai et al., 2009; Schirrmeister et al., 2011b; Strauss et al., 2013; Hugelius et al., 2013, 2014; Walter Anthony et al., 2014) and how much of these stocks is potentially mobilized due to thermal permafrost degradation and coastal erosion (Rachold et al., 2004; Jorgenson and Brown, 2005; Lantuit et al., 2009; McGuire et al., 2009; Ping et al., 2011; Schneider von Deimling et al., 2012; Vonk et al., 2012; Günther et al., 2013, 2015; Wegner et al., 2015). DOC fluxes have also been quantified in western Siberian catchments (Frey and Smith, 2005), and monitoring efforts of the large rivers draining permafrost areas and entering into the Arctic Ocean have provided robust estimations of the riverine DOC export (Raymond et al., 2007; McGuire et al., 2009). However, DOC stocks in permafrost ground ice and the resulting potential DOC fluxes in response to coastal erosion and thermal degradation are still unknown (Guo et al., 2007; Duo et al., 2008). At this moment, any inference about DOC stocks in permafrost and fluxes from permafrost is derived from measurements in secondary systems such as lake (e.g., Kling et al., 1991; Walter Anthony et al., 2014), river (e.g., Benner et al., 2004; Finlay et al., 2006; Guo et al., 2007; Raymond et al., 2007; Holmes et al., 2012) and ocean waters (e.g., Opsahl and Benner, 1997; Dittmar and Kattner, 2003; Cooper et al., 2005) or from laboratory experiments (Dou et al., 2008). In contrast, the purpose of this study was to sample and measure DOC at the source (i.e., ground ice in permafrost) directly, before it gets altered by natural processes such as exposure to the atmosphere, lithosphere and hydrosphere.

Here, we present an Arctic-wide study on DOC stocks in ground ice, aiming at incorporating massive ground ice into the Arctic permafrost carbon budget. The specific objectives of our study are

- to quantify DOC contents in different massive ground-ice types;
- to calculate DOC stocks in massive ground ice at the Arctic level;
- to put ground-ice-related DOC stocks into the context of the terrestrial Arctic OC pools and fluxes;
- to introduce relationships between organic and inorganic geochemical parameters, stable water isotopes, stratigraphy, and genetic and spatial characteristics to shed light on the origin of DOC and the processes of carbon sequestration in ground ice.

2 Study area and study sites

This study was carried out along the coastal lowlands of east Siberia, Alaska and northwest Canada (Fig. 1). All study sites, except for the Fairbanks area, are located within the zone of continuous permafrost. The sites cover a wide and representative range of geomorphological settings, terrain units and ground-ice conditions (Table 1). All studied ground-ice bodies were found in ice-rich unconsolidated
Table 1. Summary of study areas, study sites, stratigraphy of the host sediments, ground-ice inventory and the studied ice types.

<table>
<thead>
<tr>
<th>Region</th>
<th>Location</th>
<th>Longitude</th>
<th>Latitude</th>
<th>Stratigraphy and host sediments</th>
<th>Ground-ice conditions (inventory, ground-ice types, sampled ice types marked in italic)</th>
<th>Reference</th>
</tr>
</thead>
<tbody>
<tr>
<td>Western Laptev Sea</td>
<td>Cape Mamonrøy Klyk</td>
<td>117.2</td>
<td>73.6</td>
<td>– Fluvial bottom sands&lt;br&gt;– Late Weichselian Ice Complex&lt;br&gt;– Lateglacial to Holocene thermokarst deposits&lt;br&gt;– Holocene valley deposits&lt;br&gt;– Holocene cover deposits&lt;br&gt;– Yedoma hills (20–40 m a.s.l.) of ice-rich permanent sequences with wide and deep syngenetic ice wedges&lt;br&gt;– Holocene ice wedges separated by thermoerosional valleys and thermokarst depressions</td>
<td>Ice-rich permafrost sequences with wide and deep syngenetic late Pleistocene ice wedges&lt;br&gt;Boereboom et al. (2013)</td>
<td>Schirrmeister et al. (2008, 2011b); Boereboom et al. (2013)</td>
</tr>
<tr>
<td>Lena Delta</td>
<td>Samoylov Island</td>
<td>126.4</td>
<td>72.4</td>
<td>– First terrace (0–10 m a.s.l.): early to late Holocene delta floodplain, along the main river channels in the central and eastern parts of the delta; fluvial facies from organic-rich sands to silty-sandy peats bottom-up&lt;br&gt;– Modern to late Holocene floodplain; alluvial facies from peaty sands to silty-sandy peats bottom-up</td>
<td>Ice-rich permafrost with active and buried syngenetic Holocene ice wedges&lt;br&gt;Ice-rich permafrost with epigenetic Holocene ice wedges</td>
<td>Schwamborn et al. (2002); Schirrmeister et al. (2011a); Meyer et al. (2015)</td>
</tr>
<tr>
<td>Eastern Laptev Sea</td>
<td>Muostakh Island</td>
<td>129.9</td>
<td>71.6</td>
<td>– Lateglacial and Holocene cover deposits on top of Ice Complex&lt;br&gt;– Middle to Late Weichselian Ice Complex</td>
<td>Very ice-rich permafrost, late Pleistocene ice wedges, Holocene ice wedges</td>
<td>Schirrmeister et al. (2011b, c); Günther et al. (2015)</td>
</tr>
<tr>
<td>Dmitry Laptev Strait</td>
<td>Oyogos Yar coast</td>
<td>143.5</td>
<td>72.7</td>
<td>– Alteration of wide thermokarst depressions (alases) and hills representing remnants of Ice Complex deposits (Yedoma)&lt;br&gt;– Lateglacial to Holocene thermokarst deposits and on top of Ice Complex&lt;br&gt;– Euteric formed during Weichselian-to-Holocene transition&lt;br&gt;– Late Weichselian Ice Complex&lt;br&gt;– Middle Weichselian Ice Complex</td>
<td>Late Pleistocene and Holocene ice wedges; all ice wedges were sampled at a coastal bluff at an elevation of about 10 m a.s.l. in a central alas depression</td>
<td>Wetterich et al. (2009); Opel et al. (2011); Schirrmeister et al. (2011b)</td>
</tr>
<tr>
<td>New Siberian Islands</td>
<td>Bol'shoy Lyakhovsky Island</td>
<td>143.9</td>
<td>73.0</td>
<td>– Late Holocene cover deposits and Holocene valley deposits&lt;br&gt;– Lateglacial to Holocene thermokarst deposits&lt;br&gt;– Euteric formed during Weichselian-to-Holocene transition&lt;br&gt;– Middle Weichselian Ice Complex</td>
<td>Late Pleistocene ice wedges</td>
<td>Meyer et al. (2002); Andreev et al. (2004, 2009); Schirrmeister et al. (2011b); Wetterich et al. (2011, 2014)</td>
</tr>
<tr>
<td>Northern Alaska</td>
<td>Barrow CRREL Permafrost Tunnel</td>
<td>-156.7</td>
<td>71.3</td>
<td>Buried ice wedge system under about 3 m of Lateglacial to early Holocene ice-rich sediments</td>
<td>Late Pleistocene ice wedges, Holocene ice wedges</td>
<td>Sellman and Brown (1973); Meyer et al. (2010a, b)</td>
</tr>
<tr>
<td>Interior Alaska</td>
<td>Fairbanks Vault Creek Tunnel</td>
<td>-147.7</td>
<td>65.0</td>
<td>Discontinuous permafrost. Late Pleistocene ice-rich silty, loess-like organic-rich sediments 12–15 m thick with large intersecting ice wedges</td>
<td>Late Pleistocene ice wedges, Holocene ice wedges</td>
<td>Shur et al. (2004); Meyer et al. (2008)</td>
</tr>
<tr>
<td>Yukon coast</td>
<td>Koomskok Beach</td>
<td>-140.5</td>
<td>69.6</td>
<td>– Middle and late Holocene ice-rich peat, polygonal tundra&lt;br&gt;– Early Holocene thaw lake sediments, peat, ice wedge casts&lt;br&gt;– Late Wisconsin (i.e., Late Weichselian) proluvial, alluvial, eolian deposits</td>
<td>Holocene ice wedges, Holocene snowpack ice (fossil snowpack)</td>
<td>Rampton (1982); Fritz et al. (2012)</td>
</tr>
<tr>
<td>Yukon coast</td>
<td>Herschel Island</td>
<td>-139.1</td>
<td>69.6</td>
<td>– Retrogressive thaw slumps along the coast exposing massive ground ice and ice-rich sediments&lt;br&gt;– Holocene cover deposits and slope material along steep coastal bluffs&lt;br&gt;– Mixed origin of marine, near-shore and terrestrial deposits&lt;br&gt;– Push-end moraine of Late Wisconsin age</td>
<td>Buried glacier ice of ≥ 20 m thickness within Late Wisconsin diamicton, Late Wisconsin ice wedges truncated by mass movement and early Holocene thaw unconformity, epigenetic and anti-syngenetic Holocene ice wedges, buried lake ice, fossil snowpack ice</td>
<td>Mackay (1959); Rampton (1982); Fritz et al. (2011, 2012)</td>
</tr>
<tr>
<td>Yukon coast</td>
<td>Roland Bay</td>
<td>-139.0</td>
<td>69.4</td>
<td>– Retrogressive thaw slumps along the coast exposing massive ground ice and ice-rich sediments&lt;br&gt;– Holocene cover deposits and slope material along steep coastal bluffs&lt;br&gt;– Late Wisconsin diamicton</td>
<td>Late Wisconsin and Holocene ice wedges</td>
<td>Rampton (1982)</td>
</tr>
<tr>
<td>Yukon coast</td>
<td>Kay Point</td>
<td>-138.2</td>
<td>69.2</td>
<td>– Retrogressive thaw slumps along the coast exposing massive ground ice and ice-rich sediments&lt;br&gt;– Holocene cover deposits and slope material along steep coastal bluffs&lt;br&gt;– Moraine (ridge) of Late Wisconsin age</td>
<td>Presumably Late Wisconsin buried glacier ice, Holocene ice wedges</td>
<td>Rampton (1982); Harry et al. (1985)</td>
</tr>
</tbody>
</table>
Holocene and late Pleistocene (Marine Isotope Stages 2–5) deposits. Outcrops in permafrost either were accessible due to strong rates of coastal erosion along the ice-rich coasts forming steep exposures (Forbes, 2011) or were technically constructed for research purposes, such as the CRREL (Cold Regions Research and Engineering Laboratory) Permafrost Tunnel in Barrow, or for mining, such as the Vault Creek Tunnel near Fairbanks, Alaska.

Coastal outcrops in Siberia were dominated by large late Pleistocene ice wedges reaching up to 20 m in depth and up to 6 m in width (Schirrmeister et al., 2011c). They formed syngenetically during periods of rapid sedimentation of Ice Complex deposits, also known as Yedoma (Schirrmeister et al., 2013). Holocene epigenetic and syngenetic ice wedges of 1–6 m in depth and < 1.0–3.5 m in width were encountered in exposed thermokarst depressions of Lateglacial\(^1\) to Holocene origin and within the Holocene peaty cover deposits. Besides ice wedges, other types of massive ground ice were sampled, such as buried/basal glacier ice, buried lake ice and a fossil snow patch (Fig. 2). In some cases, massive ground ice occupied as much as 90 vol% of 40 m coastal exposures, erod-

\[^1\]We refer to the Lateglacial as a stratigraphic and geochronological period at the transition between the Pleistocene and the Holocene. The Lateglacial spans the latest part of the Late Weichselian/Late Wisconsin glacial period. It includes the Bolling, the Older Dryas, the Allerød and the Younger Dryas, from ca. 14 700 to 11 600 years before present (cf. de Klerk, 2004).
ing up to 10 m a\(^{-1}\) (Lantuit et al., 2012). The focus of this paper is on massive ground ice; non-massive ice (in particular pore ice and intrasedimental ice such as ice lenses) was excluded from this first attempt to calculate DOC stocks in ground ice, because of the complex genetic processes associated with the interaction with enclosing sediment and the relatively small amount of ice relative to massive ice bodies. DOC in intrasedimental ice is, however, not considered to be insignificant.

3 Material and methods

3.1 Laboratory analyses

A total number of 101 ice samples from 29 ice bodies and 3 surface water samples from 3 thermokarst lakes were studied. Ice blocks were cut with a chain saw in the field and kept frozen until further processing with a band saw in a cold lab at \(-15\,^\circ\text{C}\) for removal of partially melted margins and cleaning of the edges. Samples ≥ 50 mL were thawed at 4 \(^\circ\text{C}\) in pre-cleaned (purified water) glass beakers covered with pre-combusted aluminum foil (550 \(^\circ\text{C}\)). Meltwater was filtered with gum-free syringes equipped with glass fiber filters (Whatman™ GF/F; pore size: 0.7 \(\mu\text{m}\)) and acidified with 20 \(\mu\text{L}\) HCl\(_{\text{suprapur}}\) (30 \%) to pH < 2 in order to prevent microbial conversion. DOC concentrations (mg L\(^{-1}\)) were measured with a high-temperature (680 \(^\circ\text{C}\)) combustion total organic carbon analyzer (Shimadzu TOC-VPCH). Internal acidification is used to convert inorganic carbon into CO\(_2\), which is stripped out of solution. Non-purgeable organic carbon compounds are combusted and converted to CO\(_2\) and measured by a non-dispersive infrared detector (NDIR). The device-specific detection limit is 0.4 \(\mu\text{g L}^{-1}\). For each sample, one measurement with three to five repetitions was performed and results were averaged.

Further analyses for hydrochemical characterization included pH, electrical conductivity, major anions and cations, and stable water isotopes (\(\delta^{18}\text{O}, \delta D\)). Stratigraphic investigations and stable water isotopes were used to differentiate between genetic ice types and to assess their approximate age (i.e., Holocene and late Pleistocene). Analyses of \(\delta^{18}\text{O}\) and \(\delta D\) were carried out with a mass spectrometer (Finnigan MAT Delta-S) using the water–gas equilibration technique (for further information see Horita et al., 1989; Meyer et al., 2000). The isotopic composition is expressed in delta per mil notation (\(\delta, \%e\)) relative to the Vienna Standard Mean Ocean Water standard. The reproducibility derived from long-term standard measurements is established with 1σ better than ±0.1 \(\%e\) for \(\delta^{18}\text{O}\) and ±0.8 \(\%e\) for \(\delta D\) (Meyer et al., 2000). Samples for ion analysis were passed through cellulose-acetate filters (Whatman™ CA; pore size 0.4 \(\mu\text{m}\)). Afterwards, samples for the cation analyses were acidified with HNO\(_3\)\(_{\text{suprapur}}\) (65 \%) to prevent microbial conversion processes and adsorptive accretion, whereas samples for anion analyses were kept cool. The cation content was analyzed by inductively coupled plasma–optical emission spectrometry (ICP-OES, Perkin-Elmer Optima 3000 XL), while the anion content was determined by ion chromatography (IC, Dionex DX-320). Hydrogen carbonate concentrations were measured by titration with 0.01 M HCl using an automatic titrator (Metrohm 794 Basic Titrino). Based on HCO\(_3^{-}\) concentrations we approximated the dissolved inorganic carbon (DIC) concentrations using the molecular weights.

3.2 Statistical methods

3.2.1 Principal component analysis (PCA)

Principal component analysis (PCA) was used to summarize the variation in a biplot by reducing dimensionality of the data while retaining most of the variation in the data set (Jolliffe, 2002). Ordinally scaled variables (i.e., chemical data set) were log-transformed, centered and standardized, except for pH, \(\delta^{18}\text{O}, \delta D\), latitude and longitude not being log-transformed due the intersample invariance. Ice types (ice wedge, buried lake ice, basal glacier ice, snowpack ice, surface water) and relative age (Pleistocene, Holocene, recent) were coded with dummy variables and were superimposed as inactive supplementary variables on the ordination plot to enable rough assumptions about the relationship between chemical composition, ground-ice formation and age. The whole data set was reduced to 92 samples and 23 variables by removing those containing missing values. PCA was performed with a focus on interspecies correlation and was implemented using CANOCO 4.5 software for Windows (ter Braak and Šmilauer, 2002).

3.2.2 Univariate tree model (UTM)

A powerful tool to explore the relationship between a single continuous response variable (DOC concentration) and multiple explanatory variables is a regression tree (Zuur et al., 2007). Tree models perform well with nonlinearity and interaction between explanatory variables. UTM are used to find interactions missed by other methods and also indicate the relative importance of different explanatory variables. Univariate tree modeling was performed using the computing environment R and Brodgar 2.6.5 software for Windows (ter Braak and Šmilauer, 2002; R Core Team, 2014).

4 Results

4.1 DOC and DIC concentrations

Table 2 provides an overview of mean DOC and DIC concentrations and range for each ground-ice type. We found strong variations of DOC concentrations within and across individual ground-ice types. The highest DOC concentrations were found in ice wedges with a mean of 9.6 mg L\(^{-1}\).
and a maximum of 28.6 mg L\(^{-1}\). Late Pleistocene ice wedges were characterized by higher mean DOC concentrations than Holocene ones, with 11.1 and 7.3 mg L\(^{-1}\), respectively. Other ice types had average DOC concentrations between 1.8 and 3.0 mg L\(^{-1}\), and their range was narrower than in ice wedges (Table 2, Fig. 3). Modern surface water gave DOC values between 5.5 and 5.8 mg L\(^{-1}\).

The highest DIC concentrations were found in modern surface water with on average 22.6 mg L\(^{-1}\) and a maximum of 40.2 mg L\(^{-1}\) (Table 2, Fig. 3). DIC concentrations were lower in ground ice but varied strongly across ice types. With 8.5 mg L\(^{-1}\), late Pleistocene ice wedges were characterized by almost 4-times-higher mean DIC concentrations than Holocene ones (2.2 mg L\(^{-1}\); Fig. 3). Buried glacier ice and lake ice had similar mean DIC concentrations (around 9 mg L\(^{-1}\)) but showed large ranges: from values around 0 up to 25 mg L\(^{-1}\). Basal glacier ice, buried lake ice and snowpack ice show mean DOC concentrations between 1.8 and 3.0 mg L\(^{-1}\). For individual sample values see Supplement Table S1.

### 4.3 Principal components

The first two axes of the PCA explain 43.9 % of the variation in the data (Fig. 5). Cl\(^-\) and Na\(^+\) ions are positively correlated with the first axis in descending order of correlation, whereas Ca\(^{2+}\), Mg\(^{2+}\) and HCO\(_3\)\(^-\) ions and pH are negatively correlated. Parameters positively correlated with PCA axis 2 include information on the ice origin of Pleistocene and basal glacier ice. In contrast, \(\delta D\), \(\delta^{18}O\), DOC concentration and information on the ice origin such as ice wedges and Holocene ground ice are negatively correlated with PCA axis 2. Variations in SO\(_4^{2-}\) and NO\(_3^-\) concentration as well as information on latitude and longitude are not correlated with the first two PCA axes. The separation of ice samples in the PCA ordination plot leads to three distinct groups: (1) Holocene ice wedges and recent surface water samples are entirely negatively related to the second axis, whereas (2) Pleistocene ice wedges are entirely negatively related to the first axis. (3) Pleistocene basal glacier ice and buried lake ice are positively related to the second axis. This separation might be related to the different processes of ice formation and climate variation.

Na\(^+\) - and Cl\(^-\) -dominated samples represent Holocene ice wedges from coastal cliffs in east Siberia (Muostakh Island and Oyogos Yar). The majority of ice wedges with a terrestrial ion composition (Mg\(^{2+}\), Ca\(^{2+}\), HCO\(_3\)\(^-\)) are of late Pleistocene age in areas such as Mamontov Klyk, Bol’shoy Lyakhovsky Island, Yukon coast and the Fairbanks area. The first axis probably separates samples with a strong marine impact at its upper end from those with more of a continental background. The second axis might represent climate conditions of formation. The majority of Pleistocene ice samples with a depleted stable water isotope composition show positive sample scores, whereas Holocene ground ice, being enriched in heavy stable water isotopes, mostly shows negative sample scores and therefore plots in the lower part of the PCA (Fig. 5).

### 4.2 Correlation matrix

With the help of a correlation matrix (Corrgram Package v1.6 in R version 3.1.2, R Core Team, 2014), environmental processes and chemical relationships can be visualized that may help to explain the sequestration of DOC into ground ice. Pearson’s correlation coefficients were calculated and plotted in a correlation matrix in order to assess the degree of association between DOC, chemical properties, stable water isotopes and spatial variables (Fig. 4). A strong positive correlation suggests a mutual driving mechanism, whereas negative values imply an inverse association. Most importantly, DOC is positively related to the relative proportion of Mg\(^{2+}\) in the cation spectrum \((R = 0.65)\). Further positive relations between DOC and other parameters, although less pronounced, involve K\(^+\) \((R = 0.36)\), HCO\(_3\)\(^-\) \((R = 0.36)\) and latitude \((R = 0.38)\). The only significantly negative relationship with regard to DOC exists together with Na\(^+\) \((R = -0.44)\) (Fig. 4). Climate-driven parameters such as \(\delta^{18}O\), \(\delta D\) and \(D\) excess do not explain DOC concentrations.

### Table 2. Summarized DOC and DIC concentrations of different massive ground-ice types. For individual sample values see Table S1.

<table>
<thead>
<tr>
<th>Ice type</th>
<th>DOC mean concentration [mg L(^{-1}]]</th>
<th>DOC concentration range [mg L(^{-1}]]</th>
<th>No. of ice bodies</th>
<th>No. of samples</th>
<th>DIC mean concentration [mg L(^{-1}]]</th>
<th>DIC concentration range [mg L(^{-1}]]</th>
<th>No. of ice bodies</th>
<th>No. of samples</th>
<th>Stratigraphic affiliation</th>
</tr>
</thead>
<tbody>
<tr>
<td>Ice wedge ice</td>
<td>9.6</td>
<td>1.6–28.6</td>
<td>22</td>
<td>72</td>
<td>4.7</td>
<td>0.3–19.8</td>
<td>21</td>
<td>66</td>
<td>Holocene, late Pleistocene</td>
</tr>
<tr>
<td>Buried/basal glacier ice</td>
<td>1.8</td>
<td>0.7–3.8</td>
<td>5</td>
<td>22</td>
<td>9.3</td>
<td>0.1–25.4</td>
<td>4</td>
<td>19</td>
<td>Late Pleistocene</td>
</tr>
<tr>
<td>Buried lake ice</td>
<td>2.0</td>
<td>0.3–5.2</td>
<td>1</td>
<td>6</td>
<td>8.8</td>
<td>0.3–22.9</td>
<td>1</td>
<td>6</td>
<td>Late Pleistocene</td>
</tr>
<tr>
<td>Snowpack ice</td>
<td>3.0</td>
<td>n.a.</td>
<td>1</td>
<td>1</td>
<td>n.a.</td>
<td>n.a.</td>
<td>n.a.</td>
<td>Holocene</td>
<td></td>
</tr>
<tr>
<td>Modern surface water*</td>
<td>5.6</td>
<td>5.5–5.7</td>
<td>3</td>
<td>3</td>
<td>22.6</td>
<td>5.0–40.2</td>
<td>3</td>
<td>3</td>
<td>Recent</td>
</tr>
</tbody>
</table>

* Three modern surface water samples are from three different water bodies representing thermokarst ponds along the Yukon coast.
Figure 3. Boxplots of (a) DOC and (b) DIC concentrations in different massive ground-ice types. Plots show minimum, maximum and median values (25th percentile and 75th percentile as edge of boxes) and the number of samples in each category. IW: ice wedges (all), IW-P: Pleistocene ice wedges; IW-H: Holocene ice wedges; BGI: buried glacier ice; BLI: buried lake ice; SPI: snowpack ice; SW: surface water. For individual sample values see Table S1.

Figure 4. Correlation matrix. Correlations mentioned in the text are printed in bold. Strong positive correlations of paired variables are indicated by dark bluish colors, while strong anti-correlations are depicted in red. Hatching from the upper right to the lower left depicts positive correlations, whereas negative correlations are reversely hatched for better perceptibility in a black-and-white print. (For interpretation of the references to color in this figure legend, the reader is referred to the web version of this article.)

4.4 Univariate tree model

The UTM (Fig. 6a) shows that differences in DOC concentrations can be explained according to inorganic geochemical properties. The first two nodes split on Mg$^{2+}$ with a threshold value of 16% of the cation spectrum. The next nodes split according to thresholds in K$^+$ of 2.30 and 2.65%, respectively (Fig. 6a). Threshold percentages presented here are based on the cation spectrum only. This means that all measured cations sum up to 100%. This is statistically more robust than using individual sample concentrations which can have different magnitudes. We end up with four statistically signif-
5 Discussion

5.1 DOC stocks in ground ice and relevance to carbon cycling

While the riverine DOC export to the Arctic Ocean has been estimated as 33–34 Tg a$^{-1}$ (McGuire et al. 2009; Holmes et al., 2012), comparable numbers for the DOC input by coastal erosion and thermal permafrost degradation (also known as thermokarst) are not available yet. This knowledge gap includes the DOC contribution derived from melting ground ice from ice-rich permafrost. Table 2 provides an overview of DOC contents in different massive ground-ice types. Ice wedges hold the greatest potential for DOC storage with a maximum of 28.6 mg L$^{-1}$. This is in good agreement with DOC measurements in a so-far limited number of ice wedges by Douglas et al. (2011) in Alaska and Vonk et al. (2013b) in east Siberia, who showed DOC concentrations of 18.4–68.5 mg L$^{-1}$ ($n = 5$) and 8.8–15 mg L$^{-1}$ ($n = 3$), respectively.

Ulrich et al. (2014) have calculated maximum wedge-ice volumes (WIVs), which range from 31.4 to 63.2 vol % for late Pleistocene Yedoma deposits and from 6.6 to 13.2 vol % for Holocene thermokarst deposits in east Siberia and Alaska. Strauss et al. (2013) have shown similar averages for WIVs of 48 vol % in late Pleistocene Yedoma and 7.0 vol % for Holocene thermokarst deposits. Together with average DOC concentrations of 11.1 mg L$^{-1}$ (max 28.6) this would lead to 5.3 g DOC m$^{-3}$ (max 18.1) for late Pleistocene ice wedges in the upper late Pleistocene permafrost column (Table 3) and a DOC pool of 43.0 Tg DOC based on 416 000 km$^2$ of undisturbed Yedoma in Beringia and a mean thickness of 19.4 m (Strauss et al., 2013). DOC stocks in ice wedges in Holocene thermokarst deposits are much lower with an average 0.51 g m$^{-3}$ and a maximum of 2.6 g m$^{-3}$ due to much lower WIVs (cf. Ulrich et al., 2014) and slightly lower DOC concentrations (Table 3). With an average 2.2 Tg the Holocene ice wedge DOC pool is much lower than the late Pleistocene pool, mainly because of lower WIVs and an average thickness of 5.5 m for thermokarst de-
Table 3. DOC stocks and pools in late Pleistocene and Holocene permafrost containing ice wedges (IW) based on calculated wedge-ice volumes (WIVs) in Yedoma and thermokarst basin deposits. All other ground-ice types, especially non-massive intrasedimental ice, are not included.

<table>
<thead>
<tr>
<th></th>
<th>DOC concentration in Pleistocene IW mg L(^{-1})</th>
<th>DOC concentration in Holocene IW mg L(^{-1})</th>
<th>WIV in Pleistocene Yedoma deposits vol%</th>
<th>WIV in Holocene thermokarst deposits vol%</th>
<th>DOC stocks in Pleistocene permafrost(^c) g m(^{-3})</th>
<th>DOC stocks in Holocene permafrost(^c) g m(^{-3})</th>
<th>DOC pools in Pleistocene permafrost(^c,d) Tg</th>
<th>DOC pools in Holocene permafrost(^c,d) Tg</th>
</tr>
</thead>
<tbody>
<tr>
<td>Min</td>
<td>2.4</td>
<td>1.6</td>
<td>16.7(^a)</td>
<td>1.0(^a)</td>
<td>0.4</td>
<td>0.02</td>
<td>3.2</td>
<td>0.07</td>
</tr>
<tr>
<td>Mean</td>
<td>11.1</td>
<td>7.3</td>
<td>48.0(^h)</td>
<td>7.0(^h)</td>
<td>5.3</td>
<td>0.51</td>
<td>43.0</td>
<td>2.2</td>
</tr>
<tr>
<td>Max</td>
<td>28.6</td>
<td>19.5</td>
<td>63.2(^a)</td>
<td>13.2(^a)</td>
<td>18.1</td>
<td>2.6</td>
<td>145.9</td>
<td>11.0</td>
</tr>
</tbody>
</table>

\(^a\) WIV data by Ulrich et al. (2014). \(^b\) Mean WIV data by Strauss et al. (2013). \(^c\) This includes ice wedges only. \(^d\) According to Strauss et al. (2013) undisturbed Pleistocene Yedoma covers 416 000 km\(^2\) with a mean thickness of 19.4 m, whereas Holocene thermokarst deposits cover 775 000 km\(^2\) with a mean thickness of 5.5 m.

Figure 6. Univariate tree model (UTM) explains variability pattern in DOC concentration. (a) The tree model focuses on DOC concentration as response variable. The UTM uses 92 observations and a set of 22 explanatory variables. Mg\(^2+\) and K\(^+\) ions are most important to explain differences in DOC concentrations. Mean DOC concentrations of each group in mg L\(^{-1}\). Number of observations in each group (n). (b) Cross validation determines the statistically significant size of the tree model. The dotted line is obtained by the mean value of the errors (x error) of the cross validations plus the standard deviation of the cross validations upon convergence. For individual sample values see Table S1.

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However, all types of non-massive intrasedimental ice, raising the total ground-ice volume to ~80 % (Schirrmeister et al., 2011b; Strauss et al., 2013), are still excluded. DOC concentrations in non-massive intrasedimental ice from Mustakh Island (Siberia) and the Yukon coast (Canada) have been found to be much higher (Fritz, unpublished data). Higher DOC concentrations in intrasedimental ice than in massive ice are certainly due to the long-term contact of soil moisture with soil organic matter prior to freezing. We therefore suggest that distinguishing DOC and POC also in analyses of non-massive ground-ice types (pore ice and intrasedimental ice) would lead to a significant rise in the proportion between DOC and POC in permafrost. However, a differentiation between particulate and dissolved OC in permafrost has not been done yet, although the technical means via rhizon soil moisture sampling is already available on a cost- and time-efficient basis. Nevertheless, we are aware of the fact that DOC makes up a limited proportion of the whole permafrost carbon stocks. A cautious estimation of the ratio of DOC and POC is on the order of ~1/2000 if we consider about 2 wt % total organic carbon (TOC) in sediments (e.g., Schirrmeister et a., 2011b, c; Strauss et al., 2013) and about 10 mg L\(^{-1}\) DOC in massive ground ice. This ratio will become much smaller if DOC and POC in the whole permafrost column were differentiated, because TOC comprises both POC and DOC.
5.2 Carbon sequestration and origin in relation to inorganic geochemistry

The origin and sequestration process into ground ice seems to play an important role in the magnitude and bioavailability of DOC. Sequestration of OC into ground ice is a complex process that is dependent on water source, freezing process, organic matter origin and inorganic geochemical signature of the ambient water to form ground ice.

Figures 4 and 6a show that the electrical conductivity (i.e., total ion content) of ground ice is unrelated to DOC but that the ion composition and therefore the ion source seem to be relevant. Mg\(^{2+}\) and K\(^+\) are the most significant parameters for explaining variations in DOC concentrations (Fig. 6a). Higher Mg\(^{2+}\) and K\(^+\) fractions of the cations spectrum are positively related to higher DOC concentrations (Fig. 4). We recognize that in the node (group) with the highest average DOC concentrations (\(\bar{\phi}=11.9\ \text{mg L}^{-1}, n=40\)) we find most of the Pleistocene ice wedges and to a lesser extent Holocene ice wedges (Fig. 6a). All study areas are represented here. Both Mg\(^{2+}\) and K\(^+\) have typically high shares in terrestrial water types because Mg and K are major elements in clay minerals and feldspars. In combination with terrestrial HCO\(_3\) and Ca\(^{2+}\) the mobility of Mg\(^{2+}\) is high in Mg / Ca(HCO\(_3\))\(_2\) solutions (Gransee and Führs, 2013).

Ice wedges are fed by meltwater from atmospheric sources that have been in contact with vegetation and sediments of the tundra surface before meltwater infiltrated the frost cracks in spring. By contrast, glacier ice, snowbank ice and lake ice are primarily fed by atmospheric waters having less interaction with carbon and ion sources. Yet, the yellowish brown to gray late Pleistocene and the milky-white Holocene ice wedges have incorporated sediments and organic matter that originate from surface soils and vegetation debris that was carried along with the meltwater into the frost crack (e.g., Opel et al., 2011). Spring snow meltwater interacts with the soil material leaching out carbon as it trickles downward toward the ice wedges. Also, since wedges may take thousands of years to form and the location of their upper surface changes with time, there are numerous spatial and temporal ways that deeper soil pore waters can get incorporated into the wedge ice. Leaching of DOC from relatively young surface organic matter takes place (Guo et al., 2007; Lachniet et al., 2012) as well as dissolution of ions from sediment particles. Snowmelt feeding ice wedges strongly attract leachable components because of its initial purity. This might be the reason why especially ice wedges contain relatively high amounts of bioavailable DOC with low-molecular-weight compounds that may be old but remained fresh over millennia (Vonk et al., 2013b).

Principal component analysis clusters ice wedges into two main groups along the first axis based on Na\(^+\) and Cl\(^-\) dominating Holocene ice wedges in modern coastal settings and Mg\(^{2+}\), Ca\(^{2+}\) and HCO\(_3\) for Pleistocene ice wedges and Holocene ones being far from coasts (Fig. 5). This pattern depicts the competing influence of maritime and terrestrial/continental conditions. A similar differentiation of ice wedges (and all ground-ice types) is done along the second PCA axis (Fig. 5). Differences in stable water isotopes indicate the predominant climate variations between the late Pleistocene and the Holocene which are also reflected in the landscape (i.e., distance to sea; maritime vs. continental). Distance from the coastline is crucial for the incorporation of marine-derived ions through aerosols such as NaCl via sea spray. While the Fairbanks area is the only site far inland, all other study sites except for Samoylov Island in the central Lena River delta are coastal areas today. However, during the late Pleistocene global sea level was lower and large parts of the shallow circum-Arctic shelves were subaerially exposed. Present-day coastal sites were located up to hundreds of kilometers inland. Marine ion transport via sea spray is not expected to have played a role on inland sites but indeed since the rapid marine transgression during the Holocene that changed far-inland sites into coastal ones. Input of sea spray is only relevant during the open-water season so that a prolonged ice cover during the late Pleistocene (Nørgaard-Pedersen et al., 2003; Bradley and England, 2008) should have further reduced the influx of sea salt. Additionally, sustained dry conditions (Carter, 1981; Alifimov and Berman, 2001; Murton, 2009) probably increased eolian input of terrestrial material into ice wedges, which is then directly mirrored in the hydrochemical signature.

So far we have shown that coastal/maritime and terrestrial environmental conditions can be differentiated based on inorganic hydrochemistry and that terrestrial surface OC sources feed the DOC signal in ice wedges. DOC sequestration into ground ice is also dependent on active-layer properties, vegetation cover, vegetation communities and deposition rates. Long-term stable surfaces and relatively constant active-layer depths will lead to substantially leached soil layers in terms of DOC (Guo and Macdonald, 2006) and inorganic solutes (Kokelj et al., 2002). Based on \(\Delta^{14}C\) values and \(\delta^{13}C\) ratios on DOC from soil-leaching experiments and natural river water samples, Guo et al. (2007) have shown that intensive leaching of DOC from young and fresh plant litter and upper soil horizons occurs during the snowmelt period. Later in the season, DOC yields decreased in rivers draining permafrost areas, indicating that deepening of the active layer and leaching of deeper seasonally frozen soil horizons were accompanied by much lower concentrations of DOC due to the refractory and insoluble character of the remaining organic matter compounds. In addition, dissolved organic matter compounds in runoff into lakes and rivers can become rapidly degraded by microbial communities and photochemical reactions (Striegl et al., 2005; Olefeldt and Roulet, 2012; Cory et al., 2014). One destination of the fresh, young and therefore most bioavailable DOC components will be ice wedges (Vonk et al., 2013b), where the chemical character is preserved because of immediate freezing. This highlights the
5.3 DOC mobility and quality upon permafrost degradation

The absolute numbers of DOC in permafrost might still be small compared to the POC. However, POC from both peat and mineral soil has a relatively slow decomposition rate after thaw compared to DOC (Schuur et al., 2008). Organic matter from melting ground ice was shown to be highly bioavailable and can even enhance organic matter degradation of the host material by increased enzyme activity in ice wedge meltwater (Vonk et al., 2013b). Bioavailability experiments with Yedoma DOC from thaw streams fed by ice wedge meltwater in NE Siberia illustrated the rapid decomposability of Yedoma OC, with OC losses of up to 33% in 14 days (Vonk et al., 2013a). Incubations with increasing amounts of ice wedge water in the Yedoma-water suspension enhanced DOC loss over time. Vonk et al. (2013b) concluded that ice wedges contain a DOM pool of reduced aromaticity and can therefore be regarded as an old but readily available carbon source with a high content of low-molecular-weight compounds. Additionally, a co-metabolizing effect through high potential enzyme activity in ice wedges upon thaw leads to enhanced degradation rates of organic matter of the host material. When studying organic matter cycling in permafrost areas, we have to abandon the paradigm, which holds true for temperate regions and Arctic oceanography, that old OC is refractory and that only young OC is fresh, bioavailable and therefore relevant for foods webs and greenhouse gas considerations.

We suggest that reduced organic matter degradation during cold periods is the main reason why late Pleistocene syngenetic ice wedges have incorporated more DOC on average than Holocene ice wedges. Incorporation of soluble organic matter into ground ice might have been more effective than today for various reasons. Ice Complex deposits in the coastal lowlands formed during the late Pleistocene cold period, when high accumulation rates of fine-grained sediments and organic matter were accompanied by rapidly aggrading permafrost (Hubberten et al., 2004). This means that organic matter is less decomposed because it was rapidly incorporated into perennally frozen ground and into the surrounding syngenetic ice wedges as the permafrost table rose together with the rising surface during deposition (Schirrmeister et al., 2011b). Also, colder annual air temperatures led to reduced decomposition rates of organic matter which originated from vegetation communities dominated by easily decomposable forbs (Willerslev et al., 2014) in contrast to resistant sedge–moss–shrub tundra vegetation since postglacial times (Andreev et al., 2011). Additionally, low precipitation and reduced runoff presumably retained more DOC in the landscape, ready to be transported into frost cracks.

Guo et al. (2007) concluded that most of the DOC in Arctic rivers is derived from young and fresh plant litter and upper soil horizons. Leaching of deeper seasonally frozen soil horizons is accompanied by much lower DOC concentrations due to the refractory and insoluble character of the remaining organic matter compounds (Guo et al., 2007). DOC impoverishment in the active layer is logical as it is leached each season over a long time under modern climate conditions, where permafrost aggradation is much slower than during cold stages – if it happens at all. The quantity and quality of DOC pools in deeper permafrost is probably much higher because of – so far – suppressed remobilization. Dou et al. (2008) studied the production of DOC as water-extractable organic carbon yields from organic-rich soil horizons in the active layer and permafrost from a coastal bluff near Barrow (Alaska) facing the Beaufort Sea. Besides high DOC yields in the uppermost horizon (0–5 cm below surface) the second-highest DOC yields derived from permafrost although the sampled horizon showed lower soil OC contents than others (Dou et al., 2008). Interestingly, higher fractions of low-molecular-weight DOC, which is regarded to be more bioavailable, were generally found at greater depths. This supports the view that permafrost deposits hold a great potential for mobilizing large quantities of highly bioavailable organic matter upon degradation. Coastal erosion and thermokarst often expose old and deep permafrost strata. Contained organic matter is directly exposed to the atmosphere and transferred into coastal and freshwater ecosystems without degradation because of short travel and residence times. Therefore, Arctic coastal zones are supposed to receive high loads of bioavailable dissolved and particulate organic matter. Dou et al. (2008) used pure water (presumably MilliQ) and natural sea water as a solvent for studying the production of DOC. It turned out that seawater extraction significantly reduced DOC yields which were attributed mainly to reduced solubility of humic substances due to the presence of polyvalent cations such as Ca$^{2+}$ and Mg$^{2+}$ in seawater (Aiken and Malcolm, 1987). On the one hand Dou et al. (2008) invoked that a laboratory setup using pure water and dried/rewetted soil samples would lead to an overestimation of DOC input to the Arctic Ocean during coastal erosion. On the other hand and based on the large ground-ice volumes in coastal cliffs (Lantuit et al., 2012), we suggest that ice wedge meltwater with a low ion content is probably able to leach greater amounts of DOC from permafrost upon thaw than other natural surface water.

An open question remains as to how much DOC can be found in intrasedimental ice and how much DOC is produced upon degradation of old permafrost (e.g., late Pleistocene Yedoma type), for example as a result of coastal erosion. To answer this question, it is crucial to follow the fate of permafrost organic matter upon remobilization. Additionally, robust estimations of carbon release are crucial for predicting the strength and timing of carbon-cycle feedback effects,
and thus how important permafrost thaw will be for climate change this century and beyond.

6 Conclusions and outlook

Ground ice in ice-rich permafrost deposits contains DOC, DIC and other nutrients which are relevant to the global carbon cycle, Arctic freshwater habitats and marine food webs upon release.

The following conclusions can be drawn from this study:

- Ice wedges represent a significant DOC (45.2 Tg) and DIC (33.6 Tg) pool in the studied permafrost areas and a considerable freshwater reservoir of 4200 km$^2$.

- Syngenetic late Pleistocene ice wedges have the greatest potential to host a large pool of presumably bioavailable DOC because of (i) highest measured average DOC concentrations in combination with (ii) their wide spatial (lateral, vertical) distribution in ice-rich permafrost areas and (iii) the sequestration of fresh and easily leachable OC compounds.

- Increased incorporation of DOC into ground ice is linked to relatively high proportions of terrestrial cations, especially Mg$^{2+}$ and K$^+$. This indicates that leaching of terrestrial organic matter is the most relevant process of DOC sequestration into ground ice.

Based on our results about the stocks and chemical behavior of DOC in massive ground-ice bodies we propose that further studies shall strive to

- quantify DOC fluxes in the Arctic from thawing permafrost, melting ground ice and coastal erosion;

- differentiate between DOC and POC in permafrost including non-massive intrasedimental ice;

- quantify DOC production from permafrost in different stratigraphic settings and with different natural solvents to answer the question of what fraction of soil OC will be leached as DOC;

- assess the age and lability of DOC versus POC in permafrost and the potential impact on coastal food webs and freshwater ecosystems.

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