

Optical properties and sulfate scattering efficiency of boundary layer aerosol at coastal Neumayer Station, Antarctica

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Abstract

We measured optical properties and ionic composition of the aerosol at Neumayer Station from 2004 through 2006 by an integrating nephelometer and chemical analysis of daily aerosol samples, respectively. From this unique data set, we discussed the seasonality of optical parameters along with the chemical composition of the aerosol. Austral summer (November through March) was characterized by mean particle number concentrations of $472 \pm 260 \text{ cm}^{-3}$ compared to $168 \pm 160 \text{ cm}^{-3}$ during winter (April through October), mean scattering Ångström exponents of 1.5 ± 0.6 compared to 1.2 ± 0.5 during winter, and mean hemispheric backscattering ratios at 700 nm of 0.21 ± 0.13 compared to 0.17 ± 0.08 during winter. In contrast, light scattering coefficients (σ_{sp}) showed a broad maximum during winter ($4.8 \pm 5.3 \text{ Mm}^{-1}$ for $\sigma_{\text{sp}}(550)$). The mean single scattering albedo was $0.99 \pm_{0.02}^{0.01}$ at 550 nm. We further derived mass scattering and mass backscattering efficiencies for biogenic sulfate aerosol (BSA) at 450 nm, 550 nm, and 700 nm for relative humidities between 5% and 11%. At 550 nm the scattering efficiency for biogenic sulfate aerosol $\alpha_{\text{BSA}}^{\text{s}}(550)$ was $8.9 \pm 0.7 \text{ m}^2 \text{ g}^{-1}$ with a corresponding backscattering efficiency $\alpha_{\text{BSA}}^{\text{bs}}(550)$ of $1.0 \pm 0.08 \text{ m}^2 \text{ g}^{-1}$. From the seasonality of the aerosol composition, we inferred a dominant contribution of sulfate aerosol regarding radiative forcing in the lower troposphere from December through January, while the impact of sea salt aerosol prevailed for the rest of the year at Neumayer.

29 1. Introduction

30 Investigations of optical properties of atmospheric particles and their relation to physical
31 condition (liquid, solid), size distribution, number density, and chemical composition are
32 crucial to understand and assess radiative forcing entailed by aerosol (e.g. *Hatzianastassiou et*
33 *al.*, [2004]; *McComiskey et al.*, [2008]). Concerning marine aerosol, such kind of measure-
34 ments have been frequently conducted e.g. within the Aerosol Characterization Experiments
35 ACE-1, ACE-2, ACE-3, and the Indian Ocean Experiment (INDOEX) [*Quinn et al.*, 1996,
36 1998, 2002, and 2004], but only rarely at high southern latitudes and Antarctica [*Bodhaine et*
37 *al.*, 1986; *Bergin et al.*, 1998; *Quinn et al.*, 1996], which is the region least affected by civili-
38 zation. Polar regions are characterized by high surface albedo and large solar zenith angles.
39 Especially here, a detailed knowledge of the physico-chemical properties of the aerosol and
40 their seasonality is required to assess the role of aerosol in radiative forcing [*Wuttke and*
41 *Seckmeyer*, 2006]. Therefore one objective of our investigation was to document for the first
42 time the seasonality of optical properties along with the chemical composition of boundary
43 layer aerosol at coastal Antarctica. In addition, special emphasis was placed on determining
44 mass scattering efficiencies of biogenic sulfate aerosol because marine phytoplankton is
45 believed to be the major natural source for sulfuric acid particles and thus playing an impor-
46 tant role in the Earth's energy balance [*Andreae and Crutzen*, 1997; *Gondwe et al.*, 2003].
47 Specific mass scattering efficiencies allow a direct and calculational easy conversion of aero-
48 sol mass to aerosol optical properties and are widely used in global circulation models assess-
49 ing radiative forcing of a specific aerosol component [*Kinne et al.*, 2006]. Notational, for each
50 aerosol component j the mass scattering efficiency α_j is defined by the partial derivative of the
51 scattering coefficient σ_j of that component $\partial\sigma_j/\partial m_j$ [*Anderson et al.*, 1994]. In practice, spe-
52 cific mass scattering efficiencies were essentially derived from light scattering measurements
53 of natural aerosol mixtures in combination with chemical analyses of simultaneously sampled

54 aerosol [Quinn *et al.*, 1996, 1998, 2002, and 2004; Hand and Malm, 2007]. Usually the ob-
55 tained scattering efficiencies refer to dry aerosol (relative humidity, RH, typically below 40%)
56 and have to be adjusted to the corresponding ambient RH for appliance in climate models
57 [McInnes *et al.*, 1998]. Values for the sulfate scattering efficiency α_{sulfate} determined so far in
58 the remote marine troposphere are within a wide range between 1.5 and 7.7 m² g⁻¹ [Hegg *et*
59 *al.*, 1993; Quinn *et al.*, 1996, 1998, 2002, and 2004]. In all these previous assessments, the
60 mass fraction of sulfate aerosol in the sub-micron range was typically well below 50% which
61 may have hampered an unambiguous apportionment of the light scattering to sulfate aerosol
62 [Hegg *et al.*, 1993]. At Neumayer, however, the aerosol budget during austral summer is
63 frequently dominated by biogenic non sea salt sulfate (nss-SO₄²⁻) and methane sulfonic acid
64 (MSA) aerosol [Minikin *et al.*, 1998], allowing a more reliable apportionment of the light
65 scattering to this portion of the aerosol.

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68 2. Measurement Techniques and Methodology

69 Measurements were conducted at the Air Chemistry Observatory, Neumayer Station (70°39'
70 S, 8°15'W, http://www.awi.de/en/go/air_chemistry_observatory) which participates in the
71 Global Atmosphere Watch (GAW) programme. König-Langlo *et al.* [1998] described the
72 climatology of this site and in Weller *et al.* [2002] aspects of contamination control of the
73 sampling were reported. Concerning the representativeness of our measurements, one has to
74 consider the characteristics of the boundary layer at this site. As typical for Antarctica, surface
75 inversions are common. During winter the thickness of the surface inversion is up to about
76 2 km while from November to February inversions are rare and confined to heights less than
77 1 km [König-Langlo *et al.*, 1998]. Frequent stormy weather conditions usually destroy the
78 stratification of the lower troposphere. The stable atmospheric boundary layer (SBL) at Neu-

79 mayer has been investigated during two extended boundary layer experiments from 1983-
80 1987 and 1994. Due to the stability of the free atmosphere above the boundary layer the typi-
81 cal height of the SBL, defined as the lowest altitude above ground where turbulent mixing
82 ceased, ranged between 10 m and 50 m [*Handorf*, 1996].

83 From January 2004 through December 2006, aerosols were sampled for 24-hour time peri-
84 ods using a 2-stage PFA (polyfluoroalkoxy-copolymer) filter holder system, including a teflon
85 and a nylon (Nylasorb) filter (all 1 μm pore size). The optical aerosol properties were meas-
86 ured by means of a three wavelength integrating nephelometer (TSI, type 3563). A condensa-
87 tion particle counter (CPC, TSI, type 3022A) provided data on particle number concentrations
88 (condensation particles, CP) for particles above a diameter of 10 nm. Black carbon (BC)
89 concentrations were monitored by an aethalometer (Magee Scientific, type AE10). Aerosol
90 measurements were further supported by an aerodynamic particle sizer (TSI, type APS 3321)
91 providing data on aerosol size distributions from 0.542 μm to 10.4 μm aerodynamic diameter
92 (D_p). All experiments installed in the Air Chemistry Observatory were under daily control and
93 daily performance protocols are available. A ventilated stainless steel inlet stack (total height
94 about 8 m above the snow surface) with a 50% aerodynamic cut-off diameter around 7-10 μm
95 at wind velocities between 4-10 m s^{-1} (determined with the APS 3321) supplied the experi-
96 ments with ambient air.

97 The operation cycle of the nephelometer was set to 10 min average time and each hour a 10
98 min zero calibration with particle free air was automatically executed. The standard deviation
99 of the zero calibration signal constituted the instrumental noise level. The detection limit was
100 defined by a signal to noise ratio of two and ranged from 0.1 Mm^{-1} to 0.3 Mm^{-1} for the 10 min
101 averages ($1 \text{ Mm}^{-1} = 10^{-6} \text{ m}^{-1}$). Each year in January and after serious instrumental failure, we
102 calibrated the nephelometer with particle free air as low span gas and CO_2 as high span gas.
103 The performance of the operated nephelometer was estimated according to *Anderson et al.*

104 [1996]. Total scattering coefficients $\sigma_{sp}(\lambda)$ and hemispheric back scattering coefficients
 105 $\sigma_{bsp}(\lambda)$ were corrected for non-Lambertian illumination and truncation errors (so called angu-
 106 lar nonidealities) according to *Anderson and Ogren* [1998] (Table 4a therein, “no-cut” case
 107 for back scattering and Table 4b therein, “no cut” case for total scattering). The impact of
 108 truncation error increases with increasing particle size especially for forward scattering. Ac-
 109 cording to detailed investigations by *Anderson et al.*, [1996], the uncertainty for $\sigma_{sp}(\lambda)$ and
 110 $\sigma_{bsp}(\lambda)$ caused by nephelometer nonidealities is around 10% for sub-micron particles, increas-
 111 ing considerably (up to 20-50%) for total scatter in case of super-micron (coarse mode) aero-
 112 sol. The used corrections for total scattering took into account the Ångström exponents from
 113 the uncorrected scattering coefficients, which were a rough estimate for the size distribution
 114 of the aerosol. This enabled a more adequate truncation correction [*Anderson and Ogren*,
 115 1998]. Values below the detection limit (39 out of 6516 values) were excluded and not con-
 116 sidered in determining averages as well as data points measured within potential impact of
 117 local contamination, indicated by elevated CP number concentrations (in summary less than
 118 2% of the data). The scattering Ångström exponent \hat{a} and the backscattering ratio $b(\lambda)$ were
 119 calculated according to

$$120 \quad \hat{a}(450-700) = - \frac{\log(\sigma_{sp}(450)/\sigma_{sp}(700))}{\log(450/700)}$$

$$121 \quad b(\lambda) = \sigma_{bsp}(\lambda)/\sigma_{sp}(\lambda)$$

122 where $\hat{a}(450-700)$ refers to the wavelength pair 450 nm and 700 nm.

123 Technical details about aerosol absorption measurements with an aethalometer can be found
 124 in *Hansen et al.* [1984] and *Hansen and McMurry* [1990]. Per se, the aethalometer measures
 125 (white) light absorption by aerosol sampled on a filter (in our case Pallflex, type T60A20).
 126 We used the calibration factor recommended by the manufacturer to first calculate the BC

127 mass concentration (in ng/m^3). From this the aerosol absorption coefficient $\sigma_{\text{ap}}(550)$ can be
 128 calculated by using the specific cross section for BC aerosol suspended in air ($10 \text{ m}^2 \text{ g}^{-1}$) at a
 129 wavelength of 550 nm [Clarke *et al.*, 1986; Bodhaine, 1995]. The accuracy of absorption
 130 measurements by this method largely depends on the specific BC absorption, which may vary
 131 by a factor of 2 [Lioussé *et al.*, 1993], and instrumental problems concerning light scattering
 132 effects on the used filter material [Petzold and Schönlinner, 2004]. As a consequence, the
 133 aerosol absorption coefficients σ_{ap} presented here may have a relatively wide error margin of
 134 around $\pm 100\%$.

135 For details concerning aerosol filter handling and analyses see Piel *et al.* [2006]. Generally,
 136 samples were analyzed by ion chromatography for methane sulfonate (MS), Cl^- , Br^- , NO_3^- ,
 137 SO_4^{2-} , Na^+ , NH_4^+ , K^+ , Mg^{2+} , and Ca^{2+} . In short, the combined uncertainty was approximately
 138 $\pm 10\%$ to $\pm 15\%$ for the main components MS, Cl^- , NO_3^- , SO_4^{2-} , Na^+ , and between $\pm 20\%$ and
 139 $\pm 30\%$ for the minor species NH_4^+ , K^+ , Mg^{2+} , and Ca^{2+} . Non-sea salt sulfate (nss-SO_4^{2-}) mass
 140 concentrations were calculated by subtracting the mass concentration of the sea salt derived
 141 sulfate from the total SO_4^{2-} mass concentration (unit: ng m^{-3}). We used Na^+ as sea salt refer-
 142 ence species and the sulfate to sodium ratio in bulk sea water of $\alpha_{\text{sulfate}} = 0.252$ from Novem-
 143 ber through February and $\alpha_{\text{sulfate}} = 0.07$ from March to October due to the potential impact of
 144 sea salt fractionation by frost flower formation [Wagenbach *et al.*, 1998] i.e.:

$$145 \quad (\text{nss-SO}_4^{2-}) = (\text{SO}_4^{2-}) - \alpha_{\text{sulfate}} \times (\text{Na}^+)$$

146 We selected days where the ratio of the ($\text{nss-SO}_4^{2-} + \text{MS}$) mass concentration to the total
 147 mass concentration of all ions determined by IC, i.e. $R_{\text{sulfate}} = c(\text{nss-SO}_4^{2-} + \text{MS})/c(\text{ions})$, was
 148 higher than 0.75 for calculation of the scattering efficiency of biogenic sulfate aerosol. This
 149 was given on 62 days during polar summers within the observation period between January
 150 2004 and end of December 2006. During these particular events the mean \pm standard devia-

151 tion (std) of the ambient temperature, ambient relative humidity (RH), and wind velocity were
152 $-5.4 \pm 4.6^\circ\text{C}$, $82 \pm 28\%$, and $6.8 \pm 4.0 \text{ m s}^{-1}$, respectively.

153

154

155 **3. Results and Discussion**

156 **3.1. General Characterization of the Aerosol During the Observation Period**

157 Time series of the measured scattering coefficients $\sigma_{\text{sp}}(550)$, hemispheric back scattering
158 ratios $b(700) = \sigma_{\text{bsp}}(700)/\sigma_{\text{sp}}(700)$, scattering Ångström exponents $\alpha(450-700)$, and aerosol
159 absorption $\sigma_{\text{ap}}(550)$ are presented in Figure 1. Mass concentrations of nss-SO_4^{2-} and Na^+ , as
160 well as particle number concentrations are shown in Figure 2. Obviously all these parameters
161 exhibited a more or less pronounced seasonality. Table 1 contrasts the mean summer
162 (November through March) and winter (April through October) values. Figure 3 gives an
163 overview on the ionic composition of the aerosol during summer and winter for the relevant
164 measuring period. The results presented in Figures 2 and 3 are consistent with findings from
165 our continuous long-term aerosol sampling program established since 1983 [*Wagenbach et*
166 *al.*, 1998; *Minikin et al.*, 1998]. While sea salt aerosol constituted the main aerosol component
167 all over the year, austral summer at Neumayer was characterized by a sharp maximum of
168 biogenic sulfur aerosol (i.e. MS and nss-SO_4^{2-}) between December and February, associated
169 with more than an order of magnitude higher particle number concentrations compared to
170 mid-winter (May through August).

171 On the whole the hourly averaged $\sigma_{\text{sp}}(\lambda)$ and $\sigma_{\text{bsp}}(\lambda)$ coefficients and their variability are
172 comparable to those measured at remote Arctic sites like Pallas [*Aaltonen et al.*, 2006] and
173 Barrow [*Delene and Ogren*, 2002] or the high-alpine site Jungfrauoch [*Collaud Coen et al.*,
174 2007], but nearly an order of magnitude higher compared to South Pole [*Bergin et al.*, 1998].

175 The low scattering coefficients at South Pole are mainly caused by significantly lower aerosol
176 mass concentrations, especially for sea salt particles [Arimoto *et al.*, 2004; Harder *et al.*,
177 2000].

178 The annual mean BC mass concentration was $1.3 \pm 2 \text{ ng m}^{-3}$ with a vague maximum around
179 $2\text{-}3 \text{ ng m}^{-3}$ between September and November. Both, annual mean and seasonality of the
180 atmospheric BC burden were comparable with analogous measurements performed at the
181 Halley 5 station on the Brunt Ice Shelf [Wolff and Cachier, 1998]. The mean single scattering
182 albedo, $\omega = \sigma_{\text{sp}} / (\sigma_{\text{sp}} + \sigma_{\text{ap}})$ of the aerosol was derived from σ_{ap} and σ_{sp} measurements (daily
183 means in each case) at 550 nm. During the whole measuring period, >95% of the $\omega(550)$
184 values ranged between 1.00 and 0.97. Even considering the high error margin of σ_{ap} , (conser-
185 vatively estimated to be $\pm 100\%$), $\omega(550)$ values were never below 0.94 under clean air condi-
186 tions, i.e. if no impact of local contamination indicated by enhanced CP number concentra-
187 tions occurred.

188

189 **3.2. Seasonality of Aerosol Optical Properties**

190 Concerning optical properties, amplitudes and timing of the seasonal cycles were less
191 distinct compared to nss-SO_4^{2-} and CP. For $b(700)$ a seasonality was virtually absent during
192 2004, while for $b(450)$ as well as for $b(550)$ no seasonal cycle was discernible throughout (not
193 shown). Aerosol optical properties discussed here are primarily defined by complex refraction
194 index, shape, and size distribution of the particle ensemble and not by its chemical composi-
195 tion. Nevertheless, these physical parameters of a particle are inherently linked with its
196 chemical composition. We can assume that sulfate aerosol should be within the sub-micron
197 size range, while sea salt particles are representative for the super-micron range, albeit with a
198 considerable fraction in the accumulation mode [Murphy *et al.*, 1998]. According to Mie
199 theory, a and $b(\lambda)$ are completely defined by size distribution and complex refraction index

200 and both, \hat{a} and $b(\lambda)$ are higher when the size distribution of the aerosol is dominated by sub-
201 micron (i.e. accumulation-, Aitken-, and nucleation mode) particles [e.g. *Seinfeld and Pandis*,
202 1998, chapter 22]. Concordantly, observed $\hat{a}(450-700)$ and at least $b(700)$ values were highest
203 during summer at Neumayer and we found a significant correlation between scattering Ång-
204 ström exponents and hemispheric back scattering ratios for all wavelengths, exemplarily
205 shown for $b(700)$ in Figure 4.

206 The back scattering ratios for 450 nm listed in Table 1 exceeded those of the longer wave-
207 lengths, which appeared inconsistent because for spherical particles, back scattering ratios
208 should essentially decrease with increasing D_p/λ ratio due to more pronounced forward scat-
209 tering. We believe that this discrepancy and the lack of a seasonality for $b(450)$ and $b(550)$
210 were artefacts largely caused by the vastly differing truncation correction factors for total
211 scatter (but not for backscatter) at these wavelengths for sub-micron and super-micron aerosol
212 [*Anderson and Ogren*, 1998]. In our case, truncation corrections seemed to underestimate
213 total scatter, especially during winter. The applicability of the mentioned correction factors is
214 restricted to spherical particles and refers to a lognormal size distribution with a geometric std
215 of 1.8 [*Anderson and Ogren*, 1998]. Hence, especially during winter when the dominance of
216 sea salt aerosol caused a shift in the size distribution to higher D_p , overall correction factors
217 are not adequate. In addition sea salt particles should be non-spherical inside the
218 nephelometer due to low RH. In this case, a separate determination of the back scattering ratio
219 for sub-micron and super-micron particles would be reasonable [*Anderson and Ogren*, 1998],
220 which was not possible with our experimental set-up.

221

222 **3.3. Aerosol properties during $R_{\text{sulfate}} > 0.75$**

223 Table 2 presents a summary of the chemical composition of the aerosol during $R_{\text{sulfate}} >$
224 0.75. Due to the lack of size-segregated aerosol chemistry investigations at Neumayer, we

225 rely on relevant measurements from the Finnish station Aboa [Teinilä *et al.*, 2000], where the
226 summertime aerosol showed ionic concentrations comparable to Neumayer. Those studies
227 revealed a comparable mass size distribution for the aerosol compounds nss-SO_4^{2-} , MS and
228 NH_4^+ with a dominant mode in the accumulation size range at about $0.3 \mu\text{m}$ (RH between
229 47% and 75%), indicating that these ions were internally mixed. Nitrate was found nearly
230 exclusively in the super micron size range, most probably associated with sea salt particles
231 [Teinilä *et al.*, 2000]. Indeed, results from combined FTIR spectrometer and sun photometer
232 measurements at Neumayer in January/February 2000 supported the prevalence of an acidic
233 internal $\text{H}_2\text{O}(\text{ice})\text{-(NH}_4)_2\text{SO}_4\text{-H}_2\text{SO}_4$ aerosol mixture [Rathke *et al.*, 2002]. Consequently, we
234 are confident that the aerosol fraction determining the light scattering was an internal mixture
235 of $\text{H}_2\text{O}\text{-(NH}_4)_2\text{HSO}_4\text{-H}_2\text{SO}_4\text{-MSA}$ in the sub-micron (accumulation) mode with a negligible
236 portion of sub-micron sea salt aerosol.

237 The main advantage determining optical properties at low RH (typically below 40%) is that
238 variations can largely be attributed to changes of concentration and/or nature of aerosol parti-
239 cles and not to variations in relative humidity. Unfortunately, sulfuric acid particles show
240 hygroscopic growth under virtually all atmospheric humidity conditions [Tang and
241 Munkelwitz, 1994]. In our case, the RH was generally between 5% and 11% in the nephelom-
242 eters' measuring volume. According to the Aerosol Inorganic Model (AIM,
243 <http://www.hpcl.uea.ac.uk/~e770/aim.html>, [Clegg *et al.* 1998]) and neglecting the minor
244 sea salt portion, the mean aerosol composition inside the nephelometer under these conditions
245 should be around 60%-75% H_2SO_4 , 25%-40% H_2O and 2.6% NH_3 . As input parameters we
246 used $\text{H}_2\text{SO}_4 = 2.85 \text{ nmol}$, $\text{NH}_3 = 0.6 \text{ nmol}$, $\text{H}^+ = 5.1 \text{ nmol}$, $p(\text{H}_2\text{O}) = 380 \text{ Pa}$ at 300 K and
247 $p(\text{H}_2\text{O}) = 396 \text{ Pa}$ at 313 K (MSA could not be considered by the model and was treated as
248 sulfuric acid).

249 Concerning the optical aerosol properties under $R_{\text{sulfate}} > 0.75$, we derived scattering Ång-
250 ström exponents $\text{Å}(450-700) = 2.3 \pm 0.3$ (mean \pm std) and mean backscattering ratios above 0.2
251 for all wavelengths (Table 3). These findings supported again the dominance of sub-micron
252 aerosol in light scattering during $R_{\text{sulfate}} > 0.75$.

253

254 **3.4. Scattering Efficiency of Biogenic Sulfate Aerosol (BSA)**

255 Due to the fact that we have to suppose BSA being an internal mixture of H_2SO_4 -MSA- H_2O
256 (with a minor NH_3 contingent) at Neumayer, an apportionment of light extinction to the indi-
257 vidual components is not legitimated. For internal aerosol mixtures specific scattering effi-
258 ciencies are not additive but dependent on the particular contingent of each component in the
259 mixture which is usually impossible to disentangle [White, 1986]. Nevertheless we referred to
260 nss-SO_4^{2-} as a indicator ion for biogenic sulfate aerosol. In order to derive the sulfate scatter-
261 ing efficiency we correlated the total scattering coefficients $\sigma_{\text{sp}}(\lambda)$ and backscattering coeffi-
262 cients $\sigma_{\text{bsp}}(\lambda)$ with the biogenic sulfate aerosol mass concentration $c(\text{nss-SO}_4^{2-})$ during R_{sulfate}
263 > 0.75 (Figures 5 and 6). The parameters of reduced major axis (RMA) regression fits are
264 summarized in Tables 3 and 4. Our results for the scattering efficiency of BSA at 550 nm
265 ($\alpha_{\text{BSA}}^{\text{s}}(550) = 8.9 \pm 0.7 \text{ m}^2 \text{ g}^{-1}$) appeared somewhat high compared to $\alpha_{\text{sulfate}}^{\text{s}}(550)$ values ob-
266 tained so far in the remote marine troposphere. Hegg *et al.* [1993] determined $\alpha_{\text{sulfate}}^{\text{s}}(550)$ to
267 be $2.84 \pm 0.14 \text{ m}^2 \text{ g}^{-1}$ in the northeast Atlantic, while values between 1.5 and $7.7 \text{ m}^2 \text{ g}^{-1}$ were
268 found in the MBL of the Pacific and Indian Ocean [Quinn *et al.*, 1996, 1998, 2002, and 2004].
269 The latter authors determined backscattering efficiency $\alpha_{\text{sulfate}}^{\text{bs}}(550)$ to be between 0.40 and
270 $0.48 \text{ m}^2 \text{ g}^{-1}$ which are roughly a factor of two lower compared to our result (Table 4). In these
271 previous investigations sulfate was most commonly a minor constituent of the total sub-
272 micron aerosol mass concentration and the results applied to higher RH between 20% and
273 45% inside the instrument and thus higher water content of the aerosol in the measuring vol-

274 ume, while our scattering coefficients refer to an internal mixture of H₂SO₄-MSA-H₂O (with a
275 minor NH₃ but high MSA contingent) at RH between 5% and 11% inside the nephelometer.

276 Alternatively we correlated our measured light scattering coefficients with the sum of the
277 MS and nss-SO₄²⁻ mass concentrations $c(\text{nss-SO}_4^{2-}+\text{MS})$ explicitly considering MS, but impli-
278 cating the same scattering efficiency for nss-SO₄²⁻ and MS. This approach resulted in
279 $\alpha_{\text{sulfate+MS}}^{\text{s}}(550) = 5.1 \pm 0.5 \text{ m}^2 \text{ g}^{-1}$ (but with a poorer regression coefficient $r^2 = 0.66$), in good
280 agreement with the commonly adopted value of $\alpha_{\text{sulfate}}^{\text{s}}(550) = 5 \pm 2 \text{ m}^2 \text{ g}^{-1}$ [Charlson *et al.*,
281 1992]. Note that the hitherto derived $\alpha_{\text{sulfate}}^{\text{s}}(550)$ values mentioned above refer to marine
282 sulfate at lower latitudes where the MS portion is typically much lower compared to polar
283 regions [Gondwe *et al.*, 2004]. The high $\alpha_{\text{BSA}}^{\text{s}}$ values derived here refer to nss-SO₄²⁻ mass
284 concentrations implicitly including the high MS contingent of the internally mixed BSA
285 aerosol.

286 Statistically, (75-77)% of the variance of $\sigma_{\text{sp}}(\lambda)$ and (81-86)% of the variance of $\sigma_{\text{bsp}}(\lambda)$
287 were finally explained by variations of $c(\text{nss-SO}_4^{2-})$ (Tables 3 and 4). Generally all factors
288 determining size distribution and refraction index of the aerosol have an impact on the scatter-
289 ing coefficients $\sigma_{\text{sp,bsp}}(\lambda)$. Consequently, a significant part of the observed variance was
290 probably caused by a combination of: (i) the presence of an aerosol fraction in the sub-micron
291 range with different scattering efficiency, most probably sea salt aerosol, (ii) a mutable size
292 distribution (e.g. caused by varying relative humidity), (iii) changing refractive index. The
293 impact of these factors on $\sigma_{\text{sp,bsp}}(\lambda)$ could be antipodal (e.g. higher humidity should result in a
294 larger mean modal diameter but lower refractive index) and could not be assessed on the basis
295 of the available data.

3.5. Implications for light scattering by sulfate and sea salt aerosol over Antarctica

In the following we will compare the importance of biogenic sulfate aerosol in terms of radiative forcing with sea salt aerosol at Neumayer. From Figure 3 it is obvious that the total aerosol budget at Neumayer is dominated by sea salt particles. Especially during winter, sea salt aerosol constituted typically about 90% of the total aerosol mass. Here, low Ångström exponents and high aerosol mass concentration at low particle number densities (Figures 1 and 2) indicated the presence of super-micron sea salt particles. Due to the fact that we could not differentiate between sub- and super-micron mode of the aerosol in our measurements, a corresponding correlation of the scattering coefficients versus $c(\text{sea salt})$ resulted in a poor regression ($r^2 < 0.25$) even when more than 90% of the aerosol mass consisted of sea salt particles. Notwithstanding, from the individual $\sigma_s(550)/c(\text{sea salt})$ ratios under the condition that $c(\text{sea salt})/c(\text{ions}) > 0.9$ ($n = 178$ data points) we calculated an overall mean scattering efficiency for sea salt particles of $\alpha_{ss}^s(550) = 5 \pm 3 \text{ m}^2 \text{ g}^{-1}$. For comparison, *Quinn et al.* [1996] determined for high latitude Pacific regions $\alpha_{ss}^s(550) = 5.5 \pm 0.22 \text{ m}^2 \text{ g}^{-1}$ for sub-micron and $0.68 \pm 0.08 \text{ m}^2 \text{ g}^{-1}$ for super-micron sea salt aerosol, respectively. Overall, the scattering efficiency for sea salt aerosol appeared roughly a factor of 1.7 lower compared to biogenic sulfate aerosol. Nevertheless, except the period from December through January, scattering of solar radiation at Neumayer was dominated by sea salt particles. In continental Antarctica, on the other hand, the role of biogenic sulfate aerosol in terms of light scattering seems to be much more pronounced. Recent year-round aerosol measurements at Kohnen Station (75°S , 0°E) [*Weller and Wagenbach, 2007*] revealed an order of magnitude lower sea salt aerosol load accompanied by roughly comparable nss-SO_4^{2-} mass concentrations (Figure 7). Under these conditions, scattering of clear-sky solar radiation in the lower atmosphere can primarily be ascribed to biogenic sulfate aerosols.

320

321

322 **4. Conclusion**

323 Aerosol optical properties at Neumayer are largely characterized by sea salt and biogenic
324 sulfate particles, the latter showing a distinct seasonality with a strong summer maximum
325 which may be succinctly termed as biogenic haze (although the term haze is exaggerated
326 considering that σ_{sp} never exceeded 10 Mm^{-1} in such cases). In particular from December
327 through January, optical properties of boundary layer aerosol were most frequently deter-
328 mined by sulfate aerosol, while for the rest of the year the impact of sea salt aerosol domi-
329 nated. There is a large variability in sulfate and sea salt scattering efficiencies measured so far
330 in the remote marine troposphere, caused by the natural variability in aerosol properties.
331 Accordingly, an unique (i.e. global) sulfate mass scattering efficiency can not be expected.
332 Nevertheless the presented scattering efficiencies should be especially reliable for the marine
333 atmosphere at high southern latitudes, where the budget of sulfate aerosol is dominated by
334 biogenic sources with a relatively high amount of MS. The obtained high mass scattering
335 efficiencies and back scattering ratios for biogenic sulfate aerosol in combination with high
336 summertime mass concentrations (when incident solar radiation is maximal) emphasize the
337 significance of dimethyl sulfide (DMS) derived aerosol on the tropospheric radiative balance
338 over the Southern Ocean which has recently been demonstrated by remote sensing data [*Gab-*
339 *ric et al.*, 2005; *Meskhidze and Nenes*, 2006]. For the Pacific from 55°N to 70°S , however,
340 sea salt seems to dominate the aerosol mass in the marine boundary layer, even in the sub-
341 micron size range and should be the main contributor to light scattering by aerosol [*Quinn*
342 *and Coffman*, 1999].

343 While over the ice free Southern Ocean a net cooling effect of the aerosol layer can be ex-
344 pected due to low surface albedo [e.g. *Hatzianastassiou et al.*, 2004], the situation is more
345 subtle over regions with high surface albedo. Daily mean spectral albedo data for a flat snow

346 surface at Neumayer showed a broad maximum of 0.98 at wavelengths between 420 nm and
347 500 nm [Wuttke *et al.*, 2006], suggesting here a delicate dependence of the radiative balance
348 on the single scattering albedo ω . Clearly, assessing the impact of the aerosol considered here
349 on clear-sky radiative balance for ice covered Antarctica needs comprehensive model calcula-
350 tions. In addition, due to the lack of corresponding data, our results refer to surface observa-
351 tions primarily representative for the stable boundary layer. They discount the vertical struc-
352 ture of the aerosol layer and the changing physico-chemical properties of particles with alti-
353 tude, emphasizing the need for corresponding field measurements.

354

355

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501

502 **Table 1.** Seasonality of optical and chemical properties of tropospheric aerosol at Neumayer
 503 measured from 2004 through 2006; summer is from November through March, winter from
 504 April through October, $c(\text{seasalt}) = c(\text{Na}^+ + \text{Mg}^{2+} + \text{Cl}^- + \text{Br}^- + \text{ss-SO}_4^{2-})$.
 505

parameter	summer (mean±std)	winter (mean±std)
nss-SO ₄ ²⁻	256±141 ng m ⁻³	68±44 ng m ⁻³
MS	102±120 ng m ⁻³	9±18 ng m ⁻³
m(sea salt)	597±830 ng m ⁻³	844±1100 ng m ⁻³
σ _{sp} (450)	4.2±3.3 Mm ⁻¹	6.1±7.4 Mm ⁻¹
σ _{sp} (550)	3.2±2.9 Mm ⁻¹	4.8±5.3 Mm ⁻¹
σ _{sp} (700)	2.4±2.5 Mm ⁻¹	4.1±5.5 Mm ⁻¹
b(450)*	0.21±0.07	0.2±0.09
b(550)*	0.19±0.05	0.17±0.07
b(700)	0.21±0.13	0.17±0.08
ã(450-700)	1.5±0.6	1.2±0.5
σ _{ap} (550)	0.015±0.014 Mm ⁻¹	0.017±0.018 Mm ⁻¹
particle concentration	472±261 cm ⁻³	168±162 cm ⁻³

506 * values potentially biased by inadequate truncation correction.

507

508

509 **Table 2.** Summary of the aerosol composition during the relevant days with
 510 $c(\text{nss-SO}_4^{2-}+\text{MS})/c(\text{ions})$ ratios above 0.75. The H^+ concentration was calculated from the
 511 measured ion balance assuming that the difference between the anion and the cation equiva-
 512 lents corresponds roughly to the H^+ equivalents. The H^+ fraction is given in ion equivalent
 513 percent (eq%); $c(\text{seasalt}) = c(\text{Na}^++\text{Mg}^{2+}+\text{Cl}^-+\text{Br}^-+\text{ss-SO}_4^{2-})$.

514

Component	value (mean±std)	fraction (mean±std)
nss-SO ₄ ²⁻	351±160 ng m ⁻³	(59.0±8.7)%
MS	147±110 ng m ⁻³	(21.8±7.8)%
Cl ⁻	44.1±29 ng m ⁻³	(7.4±3.1)%
NO ₃ ⁻	29.1±15 ng m ⁻³	(5.1±2.4)%
Na ⁺	19±20 ng m ⁻³	(3.2±2.8)%
NH ₄ ⁺ [ng m ⁻³]	17.6±15 ng m ⁻³	(2.9±2.3)%
m(ions)	613±306 ng m ⁻³	100%
m(nss-SO ₄ ²⁻ +MS)	515±267 ng m ⁻³	(83.6±5.1)%
m(sea salt)	68.5±52 ng m ⁻³	(11.3±5.5)
H ⁺	8.1±4.2 ng m ⁻³	(80.3±12) eq%
particle concentration	498±230 cm ⁻³	-
black carbon	1.4±2.1 ng m ⁻³	(0.4±0.6)%

515

516

517 **Table 3.** Scattering efficiency of nss-SO₄²⁻ aerosol $\alpha_{\text{BSA}}^{\text{s}}(\lambda)$ derived from RMA regression of
 518 $\sigma_{\text{sp}}(\lambda)$ vs. $c(\text{nss-SO}_4^{2-})$ shown in Figure 1. Y-axis intercept = ε^{s} , regression coefficient = r^2 ,
 519 and backscattering ratio $b(\lambda) = \sigma_{\text{bsp}}(\lambda)/\sigma_{\text{sp}}(\lambda)$.

520

Wavelength λ [nm]	$\alpha_{\text{BSA}}^{\text{s}}(\lambda)$ [m ² g ⁻¹]	ε^{s} [m ² g ⁻¹]	r^2	$b(\lambda)$
450	13.9±1.1	-1.1±1.0	0.77	0.202±0.074
550	8.9±0.7	-0.78±0.66	0.77	0.201±0.052
700	5.2±0.4	-0.47±0.04	0.75	0.244±0.074

521

522

523

524 **Table 4.** Back-scattering efficiency of biogenic sulfate aerosol $\alpha_{\text{BSA}}^{\text{bs}}(\lambda)$ derived from RMA
 525 regression of $\sigma_{\text{bsp}}(\lambda)$ vs. $c(\text{nss-SO}_4^{2-})$ shown in Figure 2 (abbreviations see Table 3).

526

Wavelength λ [nm]	$\alpha_{\text{BSA}}^{\text{bs}}(\lambda)$ [m ² g ⁻¹]	ε^{bs} [m ² g ⁻¹]	r^2
450	1.2±0.09	0.18±0.08	0.81
550	1.0±0.08	0.034±0.06	0.85
700	0.83±0.05	0.004±0.05	0.86

527

528

529

530 **FIGURES**

531 **Figure 1:** Time series of (a) scattering coefficients $\sigma_{sp}(550)$, (b) hemispheric back-scattering
532 ratios $b(700)$, scattering Ångström exponents $\alpha(450-700)$, and (d) absorption coefficients
533 $\sigma_{ap}(550)$. Daily means are displayed in light grey and 31-days running means by the bold line
534 in dark grey.

535

536 **Figure 2:** Time series of (a) $nss-SO_4^{2-}$ and (b) Na^+ mass concentrations and (c) condensation
537 particle (CP) number concentrations. Daily means are displayed in light grey and 31-days
538 running means by the bold line in dark grey.

539

540 **Figure 3:** Mean ionic composition of the aerosol at Neumayer for (a) summer and (b) winter
541 during the total measuring period (Jan. 2004 to Dec. 2006).

542

543 **Figure 4:** Back scattering ratios $b(700)$ versus scattering Ångström exponent $\alpha(450-700)$ for
544 the total measuring period (Jan. 2004 through Dec. 2006); circles: summer data, dots: winter
545 data.

546

547 **Figure 5:** Dependence of scattering coefficients $\sigma_{sp}(\lambda)$ on $nss-SO_4^{2-}$ mass concentrations
548 during days with $c(nss-SO_4^{2-}+MS)/c(ions) > 0.75$. The lines represent RMA regression fits to
549 the measuring points (see Table 3).

550

551 **Figure 6:** Dependence of hemispheric backscattering coefficients $\sigma_{\text{bsp}}(\lambda)$ on nss-SO₄²⁻ mass
552 concentrations during days with $c(\text{nss-SO}_4^{2-}+\text{MS})/c(\text{ions}) > 0.75$. The lines represent RMA
553 regression fits to the measuring points (see Table 4).

554

555 **Figure 7:** Monthly mean values of nss-SO₄²⁻ and sea salt mass concentrations measured (a)
556 at Neumayer, and (b) at Kohnen. All monthly mean values from Neumayer comprise meas-
557 urements between January 2004 and December 2006, while the Kohnen data were obtained
558 between January 2003 and December 2005 [*Weller and Wagenbach, 2007*].

559

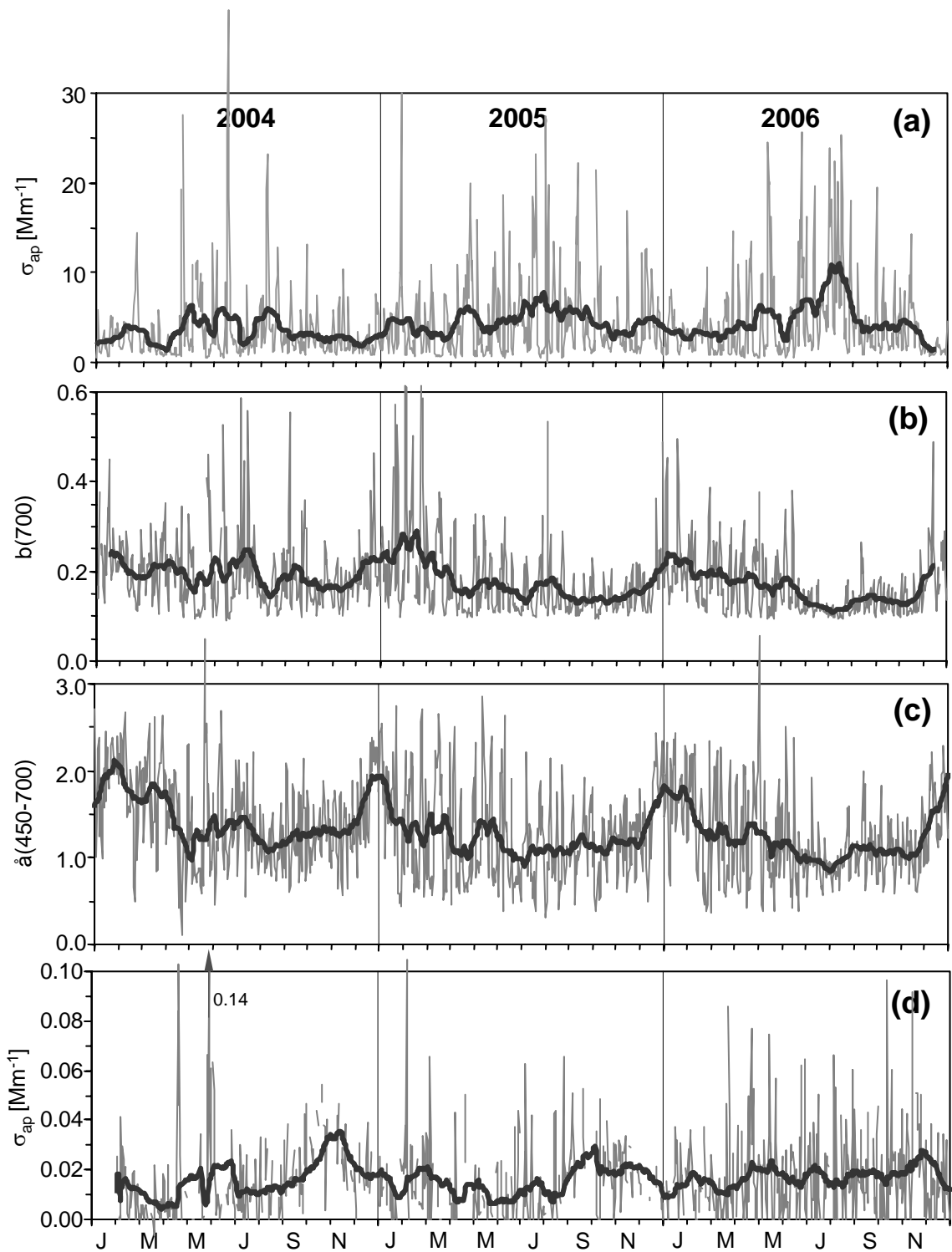


Figure 1: Time series of (a) scattering coefficients $\sigma_s(550)$. (b) hemispheric back-scattering ratios $b(700)$, (c) scattering Ångström exponents $\hat{a}(450-700)$, and (d) absorption coefficients $\sigma_{ap}(550)$. Daily means are displayed in light grey and 31-days running means by the bold line in dark grey.

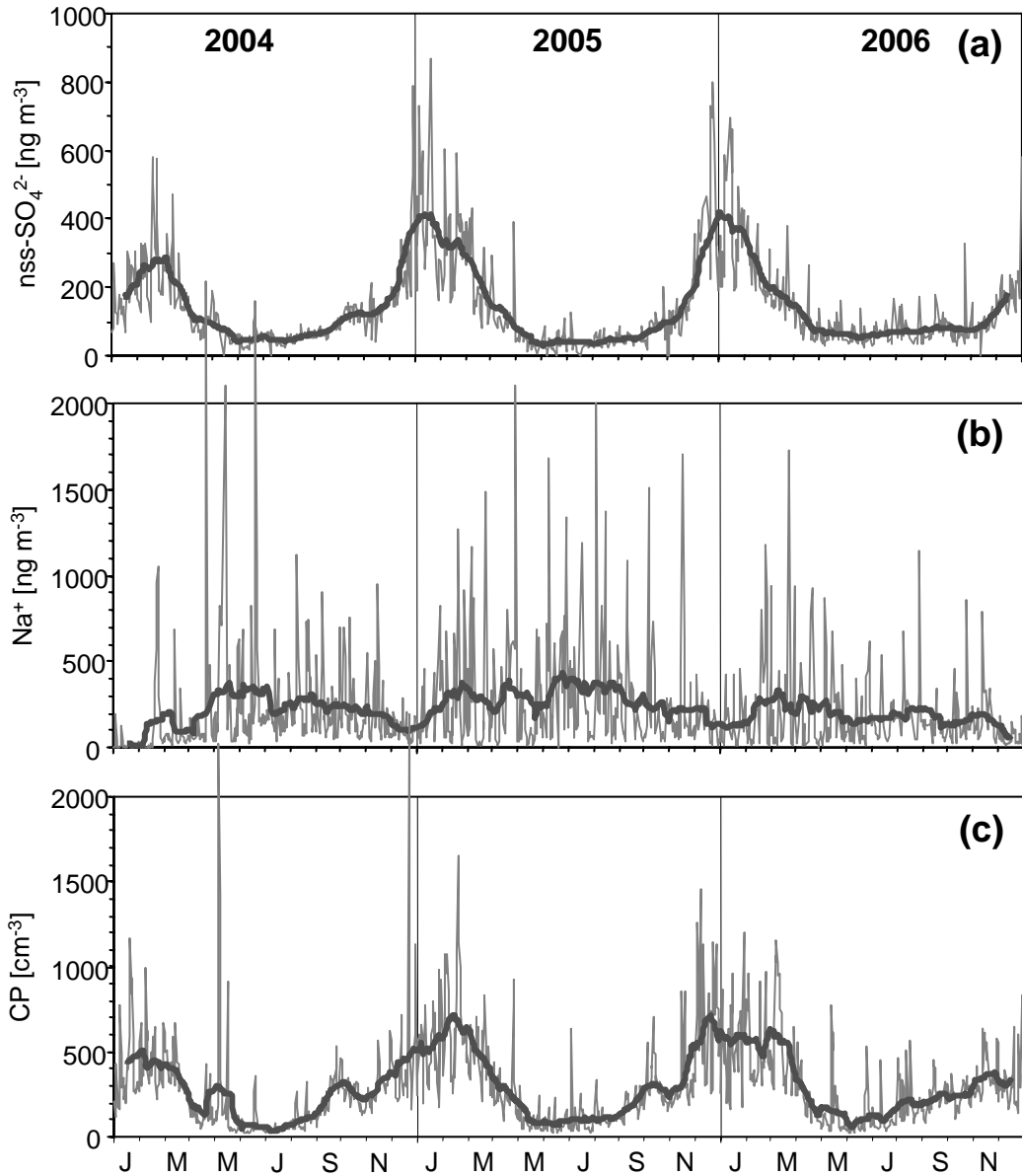


Figure 2: Time series of (a) nss-SO_4^{2-} and (b) Na^+ mass concentrations and (c) condensation particle (CP) number concentrations. Daily means are displayed in light grey and 31-days running means by the bold line in dark grey.

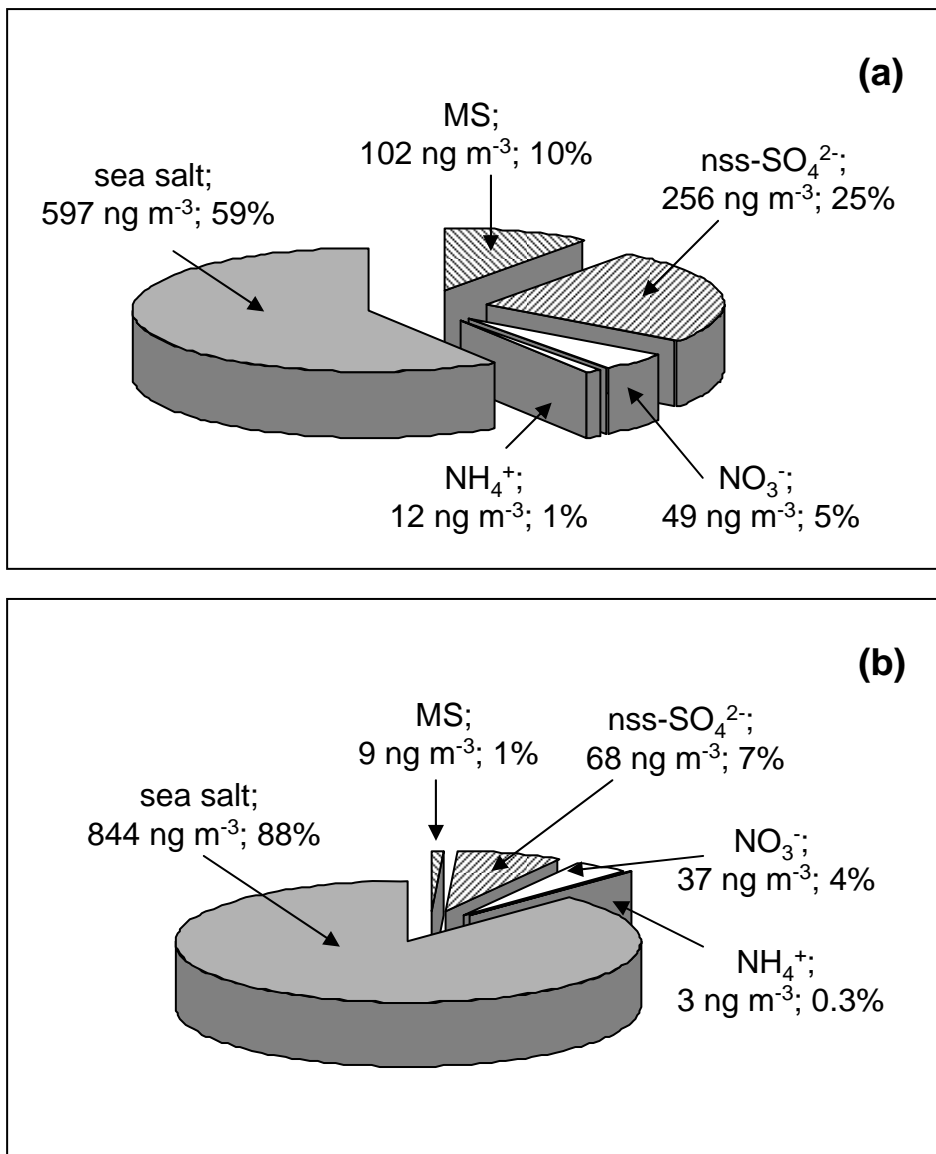


Figure 3: Mean ionic composition of the aerosol at Neumayer for (a) summer and (b) winter during the total measuring period (Jan. 2004 through Dec. 2006).

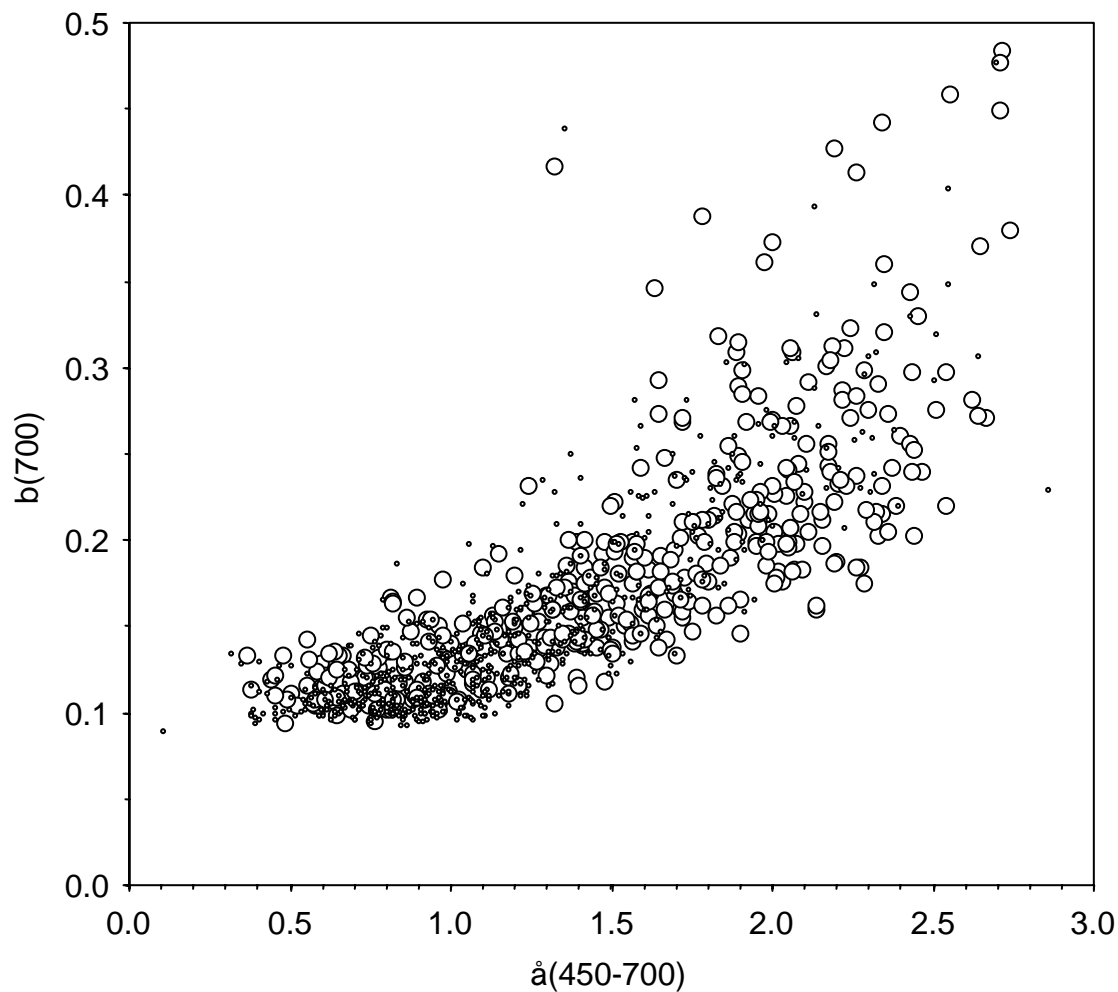


Figure 4: Hemispheric back scattering ratios $b(700)$ versus scattering Ångström exponent $\hat{a}(450-700)$ for the total measuring period (Jan. 2004 through Dec. 2006); circles: summer data, dots: winter data.

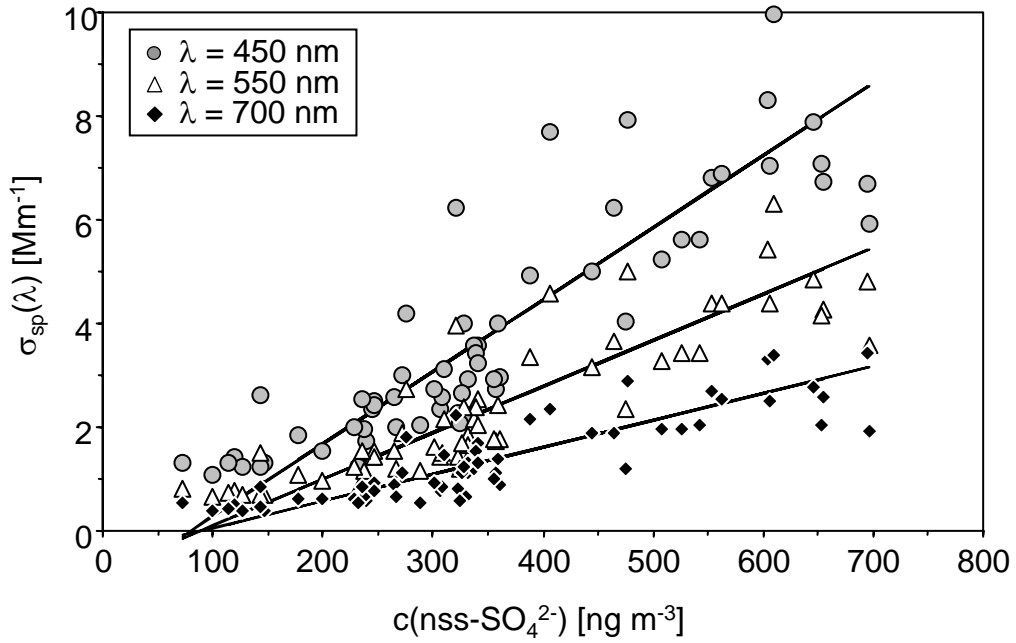


Figure 5: Dependence of scattering coefficients $\sigma_{\text{sp}}(\lambda)$ on nss-SO_4^{2-} concentrations during days with $c(\text{nss-SO}_4^{2-}+\text{MS})/c(\text{ions}) > 0.75$. The lines represent RMA fits to the measuring points (see Table 3).

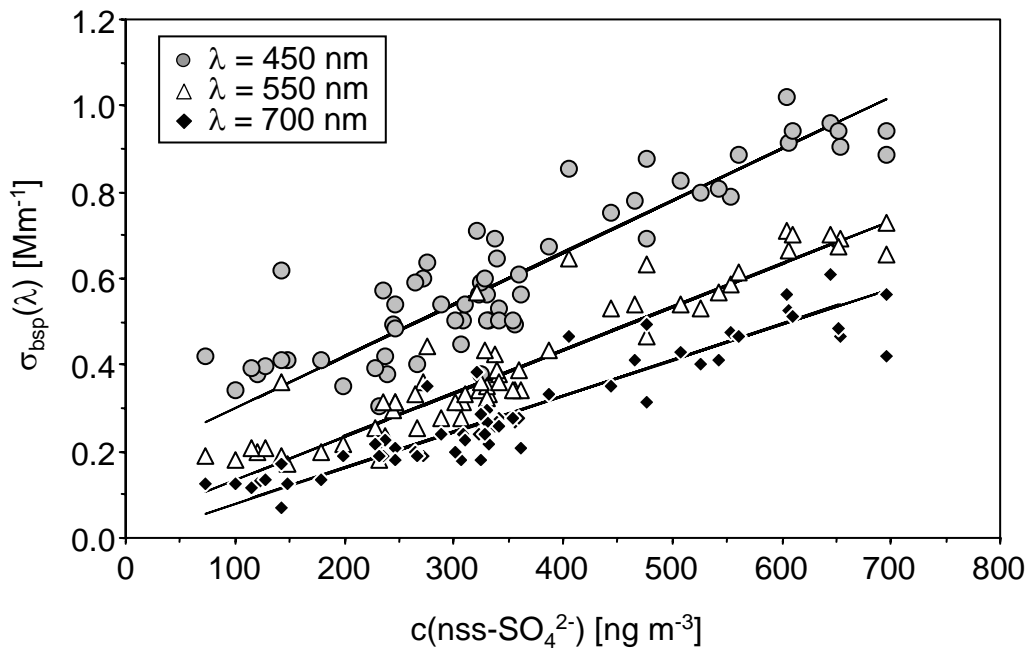


Figure 6: Dependence of backscattering coefficients $\sigma_{\text{bsp}}(\lambda)$ on nss-SO_4^{2-} concentrations during days with $c(\text{nss-SO}_4^{2-}+\text{MS})/c(\text{ions}) > 0.75$. The lines represent RMA regression fits to the measuring points (see Table 4).

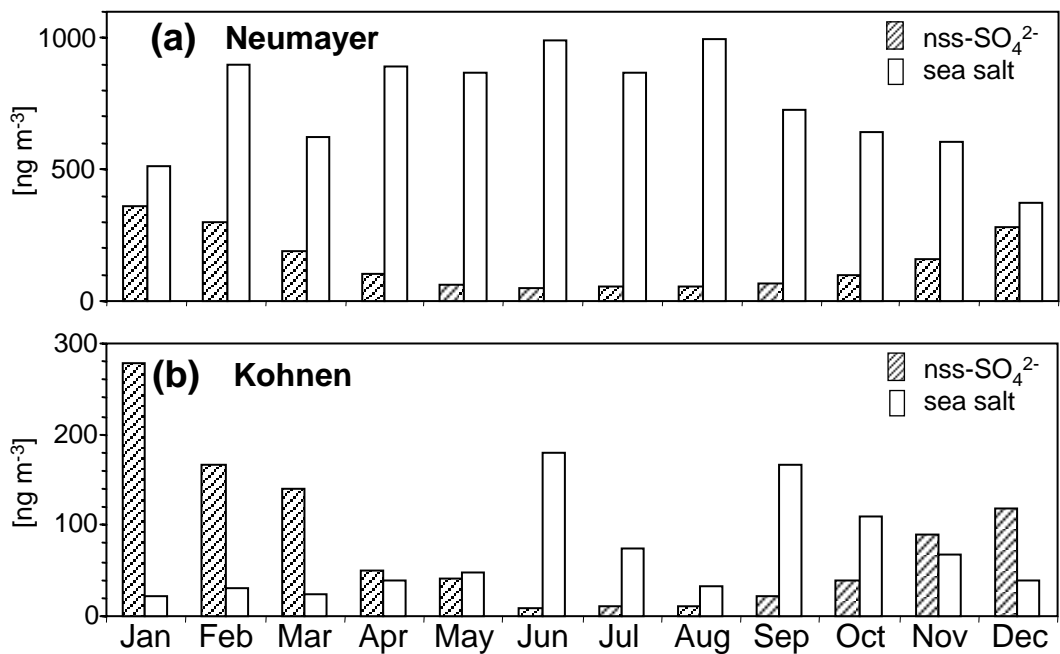


Figure 7: Monthly mean values of nss-SO₄²⁻ and sea salt concentrations measured (a) at Neumayer, and (b) at Kohnen. All monthly mean values from Neumayer comprise measurements between January 2004 and December 2006, while the Kohnen data were obtained between January 2003 and December 2005 [Weller and Wagenbach, 2007].