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Oxidation of sulfur, hydrogen, and iron by metabolically versatile Hydrogenovibrio from deep sea hydrothermal vents

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Abstract

Chemolithoautotrophic Hydrogenovibrio are ubiquitous and abundant at hydrothermal vents. They can oxidize sulfur, hydrogen, or iron, but none are known to use all three energy sources. This ability though would be advantageous in vents hallmarked by highly dynamic environmental conditions. We isolated three Hydrogenovibrio strains from vents along the Indian Ridge, which grow on all three electron donors. We present transcriptomic data from strains grown on iron, hydrogen, or thiosulfate with respective oxidation and autotrophic carbon dioxide (CO₂) fixation rates, RubisCO activity, SEM, and EDX. Maximum estimates of one strain's oxidation potential were 10, 24, and 952 mmol for iron, hydrogen, and thiosulfate oxidation and 0.3, 1, and 84 mmol CO₂ fixation, respectively, per vent per hour indicating their relevance for element cycling in-situ. Several genes were up- or downregulated depending on the inorganic electron donor provided. Although no known genes of iron-oxidation were detected, upregulated transcripts suggested iron-acquisition and so far unknown iron-oxidation-pathways.

Keywords: Hydrogenovibrio, iron oxidizer, hydrogen oxidizer, sulfur oxidizer, chemolithoautotrophy, autotrophic CO₂ fixation: Indian ridge, hydrothermal vent environment bacteria

Introduction

In deep-sea hydrothermal vent systems, hot and reduced hydrothermal fluids enriched in carbon dioxide (CO₂), hydrogen (H₂), methane, reduced sulfur (S) species, and metals (e.g., iron, Fe), rise from cracks in the basaltic ocean crust where seawater is circulating and transformed into hydrothermal fluids. They mix with entrained ambient, oxic, cold waters, creating thermal and chemical gradients along the fluid pathway [1]. In the mixing zone, intermediate inorganic S species like S⁰, thiosulfate (S₂O₃²⁻), and polysulfide can be found [2, 3]. This chemical disequilibrium can be exploited by chemolithotrophic microorganisms catalyzing the oxidation of these reduced substances to conserve energy that can be utilized for autotrophic CO₂ fixation [1, 4]. One group of chemolithoautotrophs that is often abundant in well mixed hydrothermal environments are Hydrogenovibrio spp. [5-9]. Members of this group were originally placed into the genus Thiomicrospira [10] and described as typical S-oxidizers capable of using H₂, hydrogen sulfide (H₂S), $S_2O_3^{2-}$, $[S_n(SO_3)_2]^{2-}$, and S^0

under aerobic or microaerobic conditions [7, 8, 11, 12]. Based on physiology, morphology, and phylogeny they were reclassified as *Hydrogenovibrio* [13].

Recent work showed that several Hydrogenovibrio spp. can harness energy from H₂ oxidation [5, 14]. The so far only strain capable of Fe(II) and S₂O₃²⁻ oxidation, Hydrogenovibrio sp. SC-1, was isolated from a non-hydrothermal marine habitat by microbial traps positioned in Catalina Island off California, USA [15]. To our current knowledge, Fe(II) oxidation has not been tested for other Hydrogenovibrio spp. However, in incubation experiments with hydrothermal fluids to which Fe(II) was added, CO₂ fixation rates were highly stimulated, while the relative abundance of Hydrogenovibrio/Thiomicrospira increased from 15% in the environment to >50% in the incubations [6], highlighting that this metabolic potential is likely more common in this group than thought. Until today it has not been demonstrated that a member of this group can grow with all three electron donors (Fe(II), H₂, S). Such metabolic versatility would be highly beneficial, given the temporal and spatial dynamics existing in hydrothermal vent

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environments and may explain why *Hydrogenovibrio* species are present at geographically distinct places with distinct vent chemistry [14, 16].

Here we report the isolation of three Hydrogenovibrio strains from three different vents along the Indian Ridge (IR). Members of the genus Hydrogenovibrio have previously been shown to dominate in hydrothermal vent communities of the South-West Indian Ridge (SWIR) [9, 17], and more recently a strain has been isolated from an active hydrothermal vent chimney at the SWIR that is capable of S and H₂ oxidation (Hydrogenovibrio thermophilus strain S5) [14]. We demonstrate that our isolates can oxidize Fe(II), H₂, and S₂O₃^{2–}, thereby gaining sufficient energy to synthesize biomass. We calculate the potential these strains have for element cycling in IR vents. Further, we link this physiological work with shifts in the transcriptomic data under the three incubation conditions and identify genes related to the different metabolisms.

Material and methods Field sites and sampling

Hydrothermal vent fluid samples were collected during the BGR led INDEX 2019 cruise on RV Sonne SO-271 along the Central Indian Ridge (CIR) and South-East Indian Ridge (SEIR). Field sites, sampling, and chemical parameters of the hydrothermal fluids are described in detail by Adam et al. [17]. Fluid samples used for the isolation of the three strains were 040 KIPS C/D (F1, VF4, CIR), 083 KIPS A/B (F2, VF1, SEIR), 104 KIPS C/D (F3, VF2, SEIR) (for more details see Table S4).

Quantification of cell numbers

For cell counts, a subsample of the hydrothermal fluids was fixed with 4% formaldehyde) for 24 h at 4°C. The fixed cells were then concentrated on polycarbonate filters (type: Nucleopore, 0.2 μ m pore size, Whatman, Buckinghamshire, United Kingdom), washed with sterile PBS and stored at -20° C. Filter sections were stained with DAPI (4′,6-Diamidin-2-phenylindol) and cells counted under an epifluorescence microscope.

Enrichment and isolation

Alongside the enrichment cultures that were described by Adam et al. [17], enrichments from the fluids 040 KIPS C/D, ROPOS 083 KIPS A/B, and 104 KIPS C/D were set up on ZVI (zero valent iron) plates under microoxic conditions to enrich for microaerophilic Fe(II)-oxidizers. The ZVI plates were prepared as previously described [18] with artificial seawater (ASW), N₂/CO₂ (80:20) as headspace and pH 6.8. Detailed information on the ZVI plates and medium composition can be found in the SI (Table S3). After the initial enrichment, cultures were subjected to dilution to extinction series for isolation. Growth was regularly checked by brightfield and fluorescence microscopy, where the cells were stained with LIVE/DEAD stain (BacLight, Invitrogen, Waltham, MS, USA). The purity of the culture was confirmed by microscopic examination and sequencing of the 16S rRNA gene. Strains 083 and 104 were deposited at the DSMZ under accession numbers DSM 117350 and DSM 117349. Strain 040 was physiologically very similar to strain 104 and therefore not deposited at the DSMZ.

DNA isolation and sequencing of the 16S rRNA gene

DNA was extracted with the NucleoSpin Soil kit (Macherey-Nagel, Düren, Germany) according to the manufacturer's instructions. The bacterial 16S rRNA gene was PCR amplified using the primers 27F/1492R [19] and sequenced by Sanger sequencing (Eurofins Genomics, Ebersberg, Germany). The phylogenetic tree of the strains and their closest relatives was constructed with MEGA X [20] based on the Maximum-likelihood method and Tamura-Nei model with 1000 bootstrap replications after multiple alignments using ClustalW [21].

Growth on various electron acceptors and donors

Growth was tested on FeS (in gradient tubes) and FeCl₂ under microoxic conditions, FeCl₂ and NO₃⁻, H₂S and NO₃⁻, H₂S, and O₂ (in gradient tubes), $S_2O_3^{2-}$ and NO₃⁻, and $S_2O_3^{2-}$ and O₂.

Growth on Fe(II) and quantification of Fe(II) oxidation rates

To confirm growth on Fe(II) cultures were also grown in gradient tubes with an FeS plug in the bottom and in liquid ASW medium containing FeCl₂. Gradient tubes contained gel-stabilized ASW with a bottom layer containing 1% (wt/vol) agarose with a 1:1 mixture of FeS and ASW medium overlain by ASW medium with 0.15% (wt/vol) agarose and air in the headspace [22, 23]. In gradient tubes, cultures were inoculated vertically over the whole length of the tube. Cultures in ASW with FeCl₂ were grown in 100 ml serum vials containing 50 ml ASW, 500 μ M FeCl₂ with a headspace of N₂/CO₂ (80:20) to which sterile air was added to reach an oxygen concentration of ~1%. The headspace was flushed daily with N₂/CO₂, and afterwards 500 μ M FeCl₂ and fresh sterile air (1% final O₂ concentration) was added.

Fe(II) oxidation rates were determined in cultures grown in ASW with FeCl₂. Each day samples from the culture were taken and fixed with 1 M HCl (final concentration). At the same time, a sample for cell counting was taken and fixed with 4% formaldehyde. The concentration of Fe(II) and total Fe was measured photometrically with the Ferrozine Assay [24], and for total Fe measurements all Fe(III) was reduced to Fe(II) with hydroxylamine hydrochloride. Fe(III) concentrations were calculated by subtracting Fe(II) from total Fe. Uninoculated replicates were taken as abiotic controls. Fe(II) oxidation rates were calculated from the difference between the biotic and abiotic incubations. Total cell counts were performed as described above.

Growth on H_2 and quantification of H_2 oxidation rates

Cultures growing on H₂ were cultivated in 16 ml Hungate tubes, containing MJ medium (detailed composition Table S3) with a H₂/CO₂/O₂ (79:20:1) headspace. Resazurin was added as redox indicator and the cultures were flushed with H₂/CO₂/O₂ (79:20:1) regularly once the medium became clear.

For quantification of H_2 -consumption rates, cultures were grown in 100 ml serum vials with 50 ml MJ medium and a $H_2/N_2/CO_2/O_2$ (2:77:20:1) headspace. H_2 concentrations in the headspace were measured with a Trace GC Ultra gas chromatograph (ThermoFisher Scientific, Waltham, MA, USA), using a ShinCarbon ST 100/120 column (Restek Corporation, Bellefonte, PA, USA) and a Pulsed Discharge Detector (Vici Valco Instruments, Houston, TX, USA) as previously described [5]. Uninoculated replicates were used as controls. Cell counts were performed as described above.

Growth on $S_2O_3{}^{2-}$ and quantification of $S_2O_3{}^{2-}$ oxidation rates

Cultures growing on $S_2O_3^{2-}$ were cultivated in 100 ml serum flasks containing 50 ml of T-ASW medium with 40 mM $S_2O_3^{2-}$ (detailed composition Table S3) and sterile air in the headspace. Cultures were also grown on T-ASW plates, which were prepared with T-ASW medium and 1.5% agar. For quantification of $S_2O_3^{2-}$ oxidation rates, samples were taken over time and the

 $\rm S_2O_3^{2-}$ concentration was measured via high-performance liquid chromatography (HPLC) as previously described [25, 26]. Uninoculated replicates were used as controls. Cell numbers were quantified as described above.

Quantification of ${}^{14}C$ -HCO₃⁻ incorporation rates during Fe(II), H₂, and S₂O₃²⁻ oxidation

For quantification of ${\rm ^{14}C\text{-}HCO_{3}^{-}}$ incorporation rates for strain 104 with the three different substrates, cultures were set up as described above for the rate measurements. The following treatments were prepared in triplicates for each of the electron donors in a ¹⁴C version and a no radioactivity version: (i) control without cells with electron donor, (ii) control with cells without electron donor, and (iii) a treatment with cells with electron donor. The ¹⁴C treatments were initially incubated and sampled in parallel with the other treatments and after 6 h for FeCl₂, 8 h for H₂ and 24 h for $S_2O_3^{2-}$ 1 μ Ci of ¹⁴C-NaHCO₃₋ (specific activity of 50-60 mCi mmol⁻¹) was injected to the vials and incubated for 18, 17, and 24 h for FeCl₂, H₂, and $S_2O_3^{2-}$, respectively. The ¹⁴C incubation was stopped by adding formaldehyde (final concentration 4% (v/v)). At the same time, the non-¹⁴C parallel treatments were continuously sampled to determine the consumption of FeCl₂, H₂ and $S_2O_3^{2-}$, and cell numbers during the ¹⁴C incubation.

For quantification of ¹⁴C incorporation, samples for DIC concentration measurements were taken before tracer addition, and fixed with HgCl₃. For cultures grown on H₂ and FeCl₂ DIC was quantified with a QuAAtro four-channel flow injection Analyzer (Seal Analytical) and the respective standard QuAAtro method (Q-067-05 Rev.1). For cultures grown on S₂O₃^{2–} DIC was quantified by flow-injection using a conductivity method with 30 mM HCl as carrier and 5 mM NaOH as receiver.

Just before the 14 C incubation was stopped, 100 μ l of supernatant from each vial was added to a scintillation vial containing scintillation cocktail (Ultima Gold XR, PerkinElmer) to determine the total radioactivity in the culture by liquid scintillation counting (TriCarb 291001, PerkinElmer). The remaining culture was concentrated on polycarbonate filters (pore size 0.22 μ m), washed with sterile PBS solution, and leftover bicarbonate was removed by acid fuming for 24 h with 2 M HCl in a desiccator. The FeCl₂ filters were additionally treated with oxalic acid/ammonium oxalate solution (100 mM/80 mM) and Fe(II)-EDAS (100 mM) while the filter was still mounted on the filtration tower to dissolve Feminerals, including siderite, and washed with sterile PBS again before acid fuming. The filters were added to scintillation vials, scintillation cocktail was added and the radioactivity quantified by liquid scintillation counting. The rate of C-fixation per ml culture per hour was calculated as described previously [27].

Quantification of RubisCO enzyme activity

Specific RubisCO activity was measured with an HPLC based enzyme assay as previously described [28, 29]. A 1 l batch of Hydrogenovibrio strain 104 was cultured under the same conditions as outlined above for the rate measurements with H₂. Crude extracts were prepared using a French pressure cell press (Thermo Spectronic) (for detailed information see SI). RubisCO activity was measured at 25°C in RubisCO assay buffer [100 mM Tris– HCl (pH 7.8), 10 mM MgCl₂, 1 mM EDTA, 25 mM NaHCO₃, and 1 mM DTT] with 0.2 mg μ l⁻¹ protein crude extract and 10 mM ribulose-1,5-bisphosphate (RuBP). The consumption of RuBP was quantified using the LaChrom Elite HPLC system (Hitachi, Tokio, Japan) with a Lichrospher 100 RP 18e column (VWR International GmbH, Darmstadt, Germany) (for further information see SI).

Transcriptomic analyses under different growth conditions

For transcriptomic analyses, strain 104 was grown with H₂, S₂O₃²⁻, and FeCl₂ as described above (45 \times 50 ml). After harvesting the cultures in the late exponential phase by centrifugation, RNA was extracted using the Marchery and Nagel NucleoBond RNA soil mini kit (Düren, Germany) according to manufacturer's protocol but with Chloroform:Isoamylalcohol (24:1) instead of Phenol:Chloroform:Isoamylalcohol (25:24:1). Residual genomic DNA was removed using the RapidOut DNA Removal Kit as specified by the manufacturer (ThermoFisher Scientific) and the absence of genomic DNA was verified via PCR. Library preparation for transcriptome sequencing was done with the Stranded Total RNA Prep kit with the Ribo-Zero plus rRNA depletion kit (Illumina, San Diego, USA) according to manufacturer's protocol. Sequencing was performed on a NovaSeq 6000 System (2×150 bp; Illumina) with the NovaSeq 6000 SP Reagent Kit v1.5 and NovaSeq XP 2-Lane Kit v1.5.

Sequence reads were processed with FASTP (v0.23.2) [X] to remove artificial sequences originating from adapters, low quality sequences (more than 40% of bases below a phread quality score below 15), and to correct bases in the overlap regions of the forward and reverse reads of each read pair. Reads were than aligned to the NCBI RefSeq assembly GCF_003991075.1 (strain S5, which had based on 16S rRNA gene 100% sequence similarity to our strain) with BWA MEM (v0.7.17-r1188). Samtools (v1.16) was used for sorting the alignments and featureCounts from the Subread package (v.2.0.6) was employed to obtain counts per gene based on the genomic coordinates provided with the aforementioned RefSeq assembly. Normalization and differential expression analysis were carried out with DESeq2 (v.1.40.1). Although the selection of the reference strain can have a significant impact on the results, we also carried out a complementary analysis without using any reference strain. For this, we performed a transcriptome assembly with Trinity (v2.15.1) and then annotated the assembled transcripts with Prokka (v1.14.6). By this approach we could also not detect any of the known genes involved in Fe oxidation (e.g., cyc2) as summarized elsewhere [30].

Results

Isolation of three Hydrogenovibrio strains from IR vent systems

Hydrothermal fluids from VF4 (CIR), VF1 (SEIR), and VF2 (SEIR) [17] were used for enrichment on ZVI-plates. Cells were uniform and were mostly attached to Fe-minerals, which appeared as bulbous structures. EDS analysis confirmed that the bulbous structures consist of elemental Fe (Fig. 1) and XRD analysis of the products confirmed existence of crystalline α -Fe. 16S rRNA gene sequence comparison showed that the strains were most closely related to Hydrogenovibrio thermophilus strains JR-2 and SS (Fig. 2) with 99.9, 100, and 100% sequence similarity for strain 040, 083, and 104, respectively. More information can be found in SI.

Growth of strains and quantification of electron donor oxidation and C-fixation rates

Growth was observed on FeS gradient tubes, FeCl₂ and O₂ (microoxic, 1% O₂), H₂ and O₂ (microoxic, 1% O₂), and S₂O₃ [2] and O₂ (fully oxic) (Table 1). On H₂ (MJ medium) and S₂O₃²⁻ (T-ASW medium), growth was confirmed by color change of the medium (blue to pink to colorless for MJ and red to yellow for



Fig. 1. Microscopy, and EDX of culture 104 grown on ZVI. (A and B) SEM images. In a cells attached to the bulbous mineral structures can be seen. (C) Overlay of brightfield and fluorescence microscopy. The Fe-minerals appear dark in brightfield microscopy, the cells are stained with LIVE/DEAD stain and visible in fluorescence microscopy. D: SEM image with overlain EDX spectrum measured along the x-axis of the EDX graph.

Table 1. Effect of redox substrates on growth of strains 040, 083,and 104.

	e ⁻ acceptor		
e ⁻ donor	O ₂	NO_3^-	
ZVI	+	n.d.	
FeCl ₂	+	-	
FeS gradient	+	n.d.	
H ₂	+	n.d.	
H ₂ S	_	-	
$S_2O_3^{2-}$	+	n.d.	

n.d. = not determined

T-ASW) as well as by microscopy. On T-ASW, growth was also visible from the formation of a white precipitate, likely ZVS (zero valent S).

Microbial Fe(II), H_2 , and $S_2O_3^{2-}$ oxidation rates

Cell numbers in the Fe(II)-amended cultures increased on average 4.8-, 3.9-, and 3.0-fold within 3 days for culture 040, 083, and 104. This is lower than what was reported for SC-1, for which cell numbers increased by two orders of magnitude within a week [15]. Maximum Fe(II) oxidation rates were 0.016 (± 0.002), 0.008 (± 0.004), and 0.010 (± 0.002) μ mol Fe(II) ml ⁻¹ h⁻¹ for culture 040, 083, and 104, respectively (± are SD). Cell-specific rates were 1.6, 0.9, and 1.7 fmol Fe(II) cell⁻¹ h⁻¹ for culture 040, 083, and 104, respectively. Cultures 040 and 104 completely oxidized all H₂ in the headspace within 48 h; culture 083 was slower but also completely oxidized all H₂ available within 96 h (Fig. 3). Maximum

H₂ oxidation rates were 145.2 (± 18.2), 75.4 (±3.0), and 140.3 (±6.2) nmol ml⁻¹ h⁻¹ for culture 040, 083, and 104, respectively. Per cell, the rates were 1.8, 9.2, and 1.7 fmol H₂ cell⁻¹ h⁻¹. This is very similar to what is reported for strain SP-41 (1.47–6.10 fmol H₂ cell⁻¹ h⁻¹) [5]. In all cultures S₂O₃²⁻ was completely oxidized within 120 h (Fig. 3). Maximum S₂O₃²⁻ oxidation rates were 1.16 (± 0.026), 1.05 (± 0.05), and 2.06 (± 0.014) μ mol S₂O₃²⁻ ml⁻¹ h⁻¹ for culture 040, 083, and 104, respectively. Cell specific rates were 53.6, 190.8, and 262.4 fmol S₂O₃²⁻ cell⁻¹ h⁻¹ for culture 040, 083, and 104, respectively. The oxidation rates are similar to those reported for Hydrogenovibrio thermophilus S5, which oxidized maximal 1.04 μ mol S₂O₃²⁻ ml⁻¹ h⁻¹ [14].

Autotrophic CO₂ fixation rates

Autotrophic CO₂ fixation rates were determined for strain 104 by measuring ¹⁴C-HCO₃⁻ incorporation during growth on FeCl₂, H₂, and S₂O₃²⁻. The highest rates of ¹⁴C-HCO₃⁻ fixation (per ml culture and per cell) were found when the culture was grown with S₂O₃²⁻, whereas the lowest C-fixation rates were found when the culture was grown with Fe(II) as electron donor (Table 1). ¹⁴C incorporation measurements with hydrothermal fluids conducted in other studies reported significantly lower cell-specific rates in the range of 0.0001–0.1 fmol cell⁻¹ h⁻¹ [31, 32]. In a previous study [6] rates quantified in incubation experiments with hydrothermal fluids to which Fe(II) was added were in the range of 4 × 10⁻⁷ to 20 × 10⁻⁷ mmol C ml⁻¹ h⁻¹. The rates in mmol C ml⁻¹ h⁻¹ we quantified for our strain with Fe(II) were in the lower range of this. Autotrophic CO₂ fixation in *Hydrogenovibrio* is operated by the Calvin-Benson-Bassham cycle with RubisCO as the key



Fig. 2. Phylogenetic tree. Phylogenetic tree based on 16S rRNA gene sequences, showing the phylogenetic relationship of the three isolates and closely related strains. * indicate which strains have been reclassified from *Thiomicrospira* to *Hydrogenovibrio* [13]. Colors indicate experimentally verified metabolic capabilities of the strains as indicated in the legend. Bootstrap values are only displayed if above 80%. The scale bar represents the expected number of changes per nucleotide position. *M. ferrooxydans* is added as an outgroup.

carboxylating enzyme [14, 33]. The specific RubisCO activity of *Hydrogenovibrio* strain 104 grown under H_2 :CO₂:O₂ (79:20:1) atmosphere was 41.5 ± 7 nmol RuBP per min and mg of protein crude extracts in the exponential growth phase which is 3.0 to 9.9 times lower then what has been measured for other *Hydrogenovibrio* isolates (Table 2).

This significant difference in RubisCO activities might reflect that RubisCO enzymes of strain 104 expressed under the given cultivation conditions actually have comparatively poor catalytic properties. However, the apparently reduced RubisCO activity could just as well be caused by methodological biases in the quantification of RubisCO activities (e.g., ¹⁴C incorporation versus HPLC assay, see Table 2). The transcriptomic data, though, suggest that genes associated with the carboxysome operon are down-regulated in H₂-treated cultures, which could also explain the rather low RubisCO activity (more information follow in section below).

Transcriptomic analyses under different growth conditions

Transcriptomic analyses were performed for strain 104 under the three growth conditions, namely with Fe(II), H₂, and S₂O₃^{2–} and revealed significant differences in the expression patterns (Fig. S6). Whereas over 500 genes are differentially expressed (FDR \leq 0.05 and $|log_2FC| \geq 1$) when comparing S₂O₃^{2–} against H₂, about twice as many are observed in comparisons including FeCl₂ (Table S4). Here we present the most important findings regarding Fe(II), H₂, and S₂O₃^{2–} oxidation and C-fixation. A more detailed description of the results from the transcriptome analysis can be found in the SI.

Transcriptomic shifts of genes related to energy metabolism.

There are only two known genes involved in neutrophilic microaerophilic Fe(II) oxidation (cyc2 and mtoAB) [30]. Neither is



Fig. 3. Rates of electron donor oxidation and cell numbers over time. Concentrations of electron donors and numbers of cells over time for all three cultures grown on (A) Fe(II), (B) H_2^- , and (C) $S_2O_3^{2-}$. Symbols are the mean (n = 3), error bars show standard deviation.

Table 2. CO₂ fixation rates for strain 104 and specific RubisCO activities for strain 104 and other Hydrogenovibrio species.

CO ₂ fixation rates	mmol C-fixation $ml^{-1} h^{-1}$	fmol C-fixation $cell^{-1} h^{-1}$	
S ₂ O ₃ ^{2–} oxidation strain 104 H ₂ oxidation strain 104 Fe(II) oxidation strain 104	$\begin{array}{c} 1.26 \mathrm{x} 10^{-4} \pm 9.31 \mathrm{x} 10^{-6} \\ 5.16 \mathrm{x} 10^{-6} \pm 3.27 \mathrm{x} 10^{-7} \\ 7.77 \mathrm{x} 10^{-7} \pm 7.52 \mathrm{x} 10^{-7} \end{array}$	23.30 ± 1.72 0.29 ± 0.019 0.09 ± 0.082	
RubisCO activities	nmol RuBP \min^{-1} mg $^{-1}$	reference	method
H. crunogenus TH-55 H. crunogenus XCL-2 H. thermophilus 178 Hydrogenovibrio 104	126 ± 8 410 ± 20 222 ± 40 41 ± 7	[28] [37] [76] this study	HPLC ¹⁴ C incorporation HPLC HPLC

included in the reference annotation. A number of genes encoding redox proteins (oxidoreductases) were induced with Fe(II) versus $S_2O_3^{2-}$ that likely relate to electron transport from iron to oxygen, most prominently genes of the sarcosine oxidase subunit delta (Fig. 4). A genetic association with formyltetrahydrofolate deformylase (purU) and a high induction of the system points to a highly important interrelation of C1-metabolism and sarcosine oxidase. The physiological function of this sarcosine oxidase and the importance of it specifically during growth with Fe(II) would be an interesting aspect for future research. Genes involved in Fe transport and storage were upregulated during growth on Fe(II), most prominently a TonB-dependent receptor. In Gramnegative bacteria the TonB-dependent receptor mediates the transport of siderophores into the periplasm (together with ExbB and ExbD, which are also in the transcriptome but were not significantly upregulated with Fe(II) (Fig. S4) [34-36]. Moreover, a Pirin family protein, 2Fe-2S iron-S cluster binding protein, and some hypothetical proteins were upregulated. As these are specifically induced under Fe(II)-oxidizing conditions, these enzymes likely are somehow involved in Fe oxidation or the subsequent electron transport pathway in this organism.

Hydrogenases (H₂-converting enzymes) were detected on the genome of *Hydrogenovibrio* already in 2006 [37], but H₂ oxidation was experimentally verified only ~10 years later [5, 38]. *Hydrogenovibrio* species have been shown to possess [NiFe]-hydrogenases of group 1 (*hyaAB*) and group 2b (*hupUV*) [5, 14, 38]. Hydrogenase-related genes were upregulated in cells grown on H₂ relative to those on S₂O₃²⁻ (Fig. 4). Unexpectedly, for cells grown with Fe(II), some hydrogenase-related genes were upregulated relative to those from H₂ amended incubations.

Hydrogenouibrio species rely on the Sox enzyme system and the sulfide:quinone reductase (Sqr) for $S_2O_3^{2-}$ oxidation [14, 37]. Strain 104 has an incomplete Sox pathway, soxB, soxD, and sqr genes are missing. The incomplete pathway results in the intraor extracellular accumulation of ZVS [39, 40]. Genes soxZ, soxC, soxY, and soxX were upregulated in the $S_2O_3^{2-}$ vs. H₂ comparison, whereas soxA was slightly downregulated (Fig. 4). When the strain was grown with Fe(II) and H₂, generally sox genes were downregulated and genes needed for assimilatory sulfate reduction (cysD and cysN, APS reductase, cysI, cysG) were upregulated, showing that the strain used sulfate, the sole S-source in these media, for assimilation.



Fig. 4. Transcription of selected genes related to metabolic sulfur, iron, or hydrogen oxidation in Hydrogenovibrio. Each gene is differentially regulated in at least one of the three comparisons. An extended heatmap can be found in Fig. S4.

Transcriptomic shifts of genes related to autotrophic CO_2 -fixation

Transcriptomic data indicate that strain 104 features multiple RubisCO enzymes, which may allow to facilitate CO_2 fixation at a variety of CO_2 and O_2 exposures, as has been shown for other *Hydrogenovibrio* strains before [41]. A total of three RubisCOs types were found to be transcribed in *Hydrogenovibrio* strain 104, namely (i) a carboxysomal form IA (IAc), (ii) a non-carboxysomal form IA (IAq), and (iii) a form II RubisCO. Both, form IAc and

IAq RubisCO structural genes (*cbbLS*) were markedly upregulated when Hydrogenovibrio strain 104 grew with $S_2O_3^{2-}$ (Fig. 4), which coincides with the enhanced growth observed under this treatment (Fig. 3). RubisCO form II is compared to RubisCO form I the faster catalyst, which, however, comes at the expense of a lower specificity towards CO_2 and results in lower tolerances to oxygen [42]. Using Hydrogenovibrio marinus it was demonstrated that the three RubisCO operons are regulated in response to CO_2 concentration, with elevated CO_2 levels (<2%) promoting CbbM synthesis [43]. The down regulation of the form II RubisCO structural gene *cbbM* in the $S_2O_3^{2-}$ treatment, performed under fully oxic conditions and atmospheric CO₂ levels is therefore plausible, as is its significant upregulation in the low-O₂ but high-CO₂ treatments, i.e., cultures grown with H₂ and FeCl₂ in a 20:80 CO₂:N₂ atmosphere with 1% O₂. However, the non-carboxysomal form IAq is significantly downregulated when comparing H₂ and FeCl₂ treatments, although the starting O₂ and CO₂ levels are identical in both experiments, suggesting that the O₂ and CO₂ levels are not the only factor determining the expression of RubisCO, but that available electron donors are directly or indirectly also of relevance.

The carboxysomal RubisCO form IAc of strain 104 is arranged in a carboxysome operon together with the carboxysome shell proteins and a carboxysomal carbonic anhydrase. Thus, *Hydrogenovibrio* strain 104, like other *Hydrogenovibrio* species, possesses a carbon concentrating mechanism (CCM) that facilitates the active uptake of dissolved inorganic carbon to generate up to 100-fold increased intracellular versus extracellular concentrations [44, 45]. Therefore it is reasonable that the genes related to CCM (carbonic anhydrases and carboxysomes) were highly upregulated when strain 104 grew with $S_2O_3^{2-}$ under atmospheric i.e., comparatively low CO_2 concentrations, versus FeCl₂- and H₂treated cultures growing under high CO_2 levels, i.e., with a $CO_2:N_2$ headspace of 20:80 (Fig. 4). Additional information on RubisCO associated genes (*cbbQO* and *lysR*) can be found in the SI.

Discussion

We describe three newly isolated strains of Hydrogenovibrio from hydrothermal vents of the IR, able to grow autotrophically by Fe(II), H₂, and $S_2O_3^{2-}$ oxidation. This study highlights the metabolic flexibility of Hydrogenovibrio providing a competitive advantage over other organisms to thrive in various vent environments.

All fluids from which the enrichments originate contained H₂S (<10-60 μ M) and Fe(II) (1.3-45 μ M) [17]. High H₂ concentrations have been measured in fluids of the Kairei vent field, located on the CIR, and previously H₂ oxidizers have been isolated or enriched from the IR [14, 17, 46]. Given that in the environment they are exposed to various electron donors, it is beneficial if microorganisms can switch between different metabolisms or simultaneously use them. On the ZVI plates that were used for isolation of the strains, Fe(II) and H₂ is available simultaneously as H₂ is produced from the reduction of H₂O by ZVI during the hydrogen evolution reaction (HER) [47], for more information see SI. The production of H₂ per ZVI oxidized during HER is quite significant with an overall stoichiometry between 1:1 to 4:3 [47]. That the strains can also grow on FeCl₂ and in gradient tubes with FeS in the bottom plug shows that they do not depend on H₂ for growth but can grow autotrophically on Fe(II) alone. It is yet to be investigated which electron donor they prefer or if they can make use of them simultaneously, as it was shown previously for the Fe(II)- and H₂-oxidizing Ghiorsea bivora [48], which had a growth benefit when both electron donors were present simultaneously. When strain 104 was growing on FeCl₂ we found that hydrogenases were amongst the genes with the highest log2-fold change compared to cells grown on $S_2O_3^{2-}$. This is an indication that the strain may be using both metabolisms simultaneously in the environment. ZVI plates, which were used for isolation, would provide an advantage for organisms being able to use H₂ and Fe(II) simultaneously and therefore likely specifically select for organisms with this ability.

Most microaerophilic Fe(II)-oxidizers are described to produce twisted stalks or sheaths as characteristic Fe(III)-mineral structures [49-51]. This is different for our strains, they produce "corallike" structures consisting of bulbous Fe(III)-minerals with \sim 1-6 μ m in diameter (Fig. 1). These structures appear very different from Fe(III) that precipitated in abiotic controls and are similar to what is reported for Ghiorsea bivora [48], a Zetaproteobacterium that is also capable of Fe(II) and H₂ oxidation, and Hydrogenovibrio sp. SC-1, the only other known Hydrogenovibrio capable of Fe(II) oxidation [15]. For Ghiorsea it was shown that it produces soluble exometabolites. These exometabolites are responsible for the Fe(III)-mineral formation outside the cell, that is different from abiotic mineral formation [52]. We could not find any known genes for Fe(II) oxidation in the transcriptome, therefore we conclude that, as previously suggested for SC-1 [15], the organism uses another Fe(II) oxidation pathway as indicated by Fe(III)minerals being different from those generated by other Fe(II)oxidizers. The genome of SC-1 was also sequenced and no known genes for Fe oxidation can be found in the genome [53].

When the strain 104 was grown on FeCl₂, hydrogenases were upregulated compared to S_2O_3 [2]. Five of thirteen observed hydrogenase-related genes were significantly higher expressed with FeCl₂ than in H₂, whereas two were significantly less expressed (Fig. 4). However, the latter two are other hydrogenases than the ones important for H₂ oxidation as seen from the comparison of H₂ and $S_2O_3^{2-}$, as they are also downregulated in this comparison.

It could be that hydrogenases are upregulated with FeCl₂ as the strain potentially uses Fe(II) and H₂ simultaneously. However, under the conditions that the Fe(II) culture was grown for the transcriptomic analyses, there is no H₂ available. Another reason for the upregulation could be that hydrogenases themselves are somehow involved in Fe(II) oxidation. This hypothesis would require further research though. Another interesting aspect is the upregulation of TonB in the culture grown on Fe(II). One possibility for the upregulation of genes related to siderophores in our strain could be that they are involved in the formation of the Fe(III) mineral structures, similar to the soluble exometabolites that are involved in Fe(III) mineral formation in Ghiorsea [52]. Siderophores have been found to be abundant and diverse in hydrothermal plumes [54]. They are important in keeping Fe in solution and thereby impact cycling and transport of iron in the ocean [55]. Whereas TonB can also be involved in the transport of carbohydrates [56], especially in venting regions where carbohydrates are abundant, such as the Guaymas Basin [57]. A study that investigated microbial genes involved in Fe uptake in hydrothermal plumes in the Guaymas Basin found that TonB was amongst the most abundant genes in the whole plume metatranscriptome [58] and the authors link that to Fe transport. That our strain has the potential to be involved in siderophore production/export highlights the environmental importance that Fe-oxidizing microbes such as Hydrogenovibrio might have.

Thermodynamically, thiosulfate is with -762 kJ/mol S₂O₃²⁻ oxidized the most favorable of the investigated electron donors [59]. Hydrogen yields -237 kJ/mol H₂ [60] and microaerophilic Fe(II) oxidation only about -90 kJ/ mol Fe oxidized [61]. Results from the 14 C-HCO₃⁻ incorporation experiments show that the strain can grow autotrophically with all three tested electron donors. C-fixation rates with S₂O₃²⁻ were much higher compared to H₂ and Fe(II), consistent with much higher cell growth and the high upregulation of RubisCO with S₂O₃²⁻. Based on our measurements for C-fixation and concurrent substrate utilization, for each mole of C-fixed ca. 16, 23, and 35 moles of substrate had

to be oxidized for $S_2O_3^{2-}$, H_2 , and Fe(II), respectively. The ratios of substrate oxidized per C-fixed in other autotrophic organisms range from 2:1 to 5:1 for aerobic sulfide oxidizers [62-65], 10:1 for ammonium oxidizers [66, 67], and 25:1 to 80:1 for nitrite oxidizers [68]. Considering that 4 mol electrons are needed for each mol of C-fixation, this means that \sim 7, 9, and 11% of the electrons transferred from the substrate went into C-fixation (for C-reduction) and the remaining electrons were used for energy generation. This agrees relatively well with previous reports that ~10-20% of the electrons are used for biomass synthesis [63, 69-71]. The values for oxidized Fe(II) per fixed CO₂ that we found for Fe(II)-oxidizing conditions are slightly lower than what was estimated previously for other neutrophilic microaerophilic Fe(II) oxidizers, for which values of 43 to 70 mol oxidized Fe(II) per mol of fixed CO₂ were suggested [72, 73]. The rates of CO₂-fixation were much higher for $S_2O_3^{2-}$ compared to FeCl₂ and H₂, likewise the increase in cell numbers was much faster. It will have to be elucidated in the future, which is the favored electron donor in the environment where the strains were isolated from.

In the environmental samples used for isolation of our new strains, we found that Hydrogenovibrio/Thiomicrospira had a relative abundance of up to 6% [17]. For the fluid from which strain 104 was isolated we counted 3.44×10^6 cells ml⁻¹. This means that we potentially have up to 2.064×10^5 Hydrogenovibrio cells per ml of hydrothermal fluid. Assuming that all these cells consume either Fe(II), H₂, or $S_2O_3^{2-}$ at the rate measured for the isolates, we could find rates of C-fixation of 1.75×10^{-8} , 6.04×10^{-8} , and 4.8 \times 10⁻⁶ mmol C ml⁻¹ h⁻¹, respectively. Unfortunately, we do not have flow rates for the vents from which the samples originate. Based on literature, we assume rates for low-temperature diffuse vents between 116–17580 l h^{-1} [74, 75]. Based on these numbers and the rates we measured with the cultures, between 71-10777, 159–24109, and 6282–952183 μ mol Fe(II), H₂, and S₂O₃²⁻ could become oxidized per vent-site per hour, respectively. For Cfixation we calculated that during Fe(II), H_2 , and $S_2O_3^{2-}$ oxidation by Hydrogenovibrio between 2–308, 7–1061, and 557–84420 μ mol C could be fixed per vent site and hour, respectively. These rates are potential rates that can be reached by a pure culture under ideal conditions. Thus, they are likely overestimates of what can be reached in the environment and many assumptions go into this calculation. The details of the calculation can be found in the SI (Table S5). Comparisons of the cell-specific ¹⁴C fixation rates from our strain (Table 2) and rates determined in hydrothermal fluids, which were in the range of 0.0001–0.1 fmol cell⁻¹ h⁻¹ [31, 32], show that our rates are high. However, in another study ¹⁴C fixation rates with Fe(II) in hydrothermal fluids have been in the same range [6] $(4 \times 10^{-7} \text{ to } 20 \times 10^{-7} \text{ mmol C ml}^{-1} \text{ h}^{-1}$, Table 2). Even though our estimates are rough, the calculations demonstrate that Hydrogenovibrio potentially substantially contributes to C, S, Fe, and H₂ cycling at these vents.

The metabolic versatility of the isolated strains highlight that *Hydrogenovibrio* species do not only play an important role in S and carbon cycling, but are also important in H_2 and iron cycling. Because no known genes for Fe(II) oxidation were found, we conclude that the strain uses a yet unknown pathway for Fe(II) oxidation. The metabolic versatility makes these strains good candidates for studying new Fe oxidation genes and mechanisms.

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Author contributions

MP planned the study and secured funding. KLM collected samples, performed enrichment, isolation, cultivation of the strains, and physiological and ¹⁴C incorporation experiments. MP and KLM evaluated data and wrote the manuscript. MA evaluated the transcriptomic data. SB quantified RubisCO activity, isolated RNA for the transcriptome sequencing and evaluated data. PD evaluated and visualized transcriptomic data. CHS and JS performed SEM and XRD analysis and evaluated the data. AS performed thiosulfate measurements and evaluated data. TB helped with the evaluation and discussion of transcriptomic data. SH supervised DIC measurements. JF performed transcriptome sequencing. All authors contributed to and revised the manuscript.

Supplementary material

Supplementary material is available at The ISME Journal online.

Conflicts of interest

The authors declare no competing interests.

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Data availability

Sequence data for the transcriptomes reported in this publication have been submitted to the European Nucleotide Archive. They are publicly available under accession PRJEB71696. The 16S rRNA gene sequences of all three strains were deposited under GenBank accession numbers PP083208, PP083209, and PP083210.

References

- Jannasch HW, Mottl MJ. Geomicrobiology of deep-sea hydrothermal vents. Science 1985;29:717–25.
- Gartman A, Yücel M, Madison AS et al. Sulfide oxidation across diffuse flow zones of hydrothermal vents. Aquat Geochem 2011;17:583–601. https://doi.org/10.1007/s10498-011-9136-1
- Beinart RA, Gartman A, Sanders JG et al. The uptake and excretion of partially oxidized sulfur expands the repertoire of energy resources metabolized by hydrothermal vent symbioses. Proc Biol Sci 2015;282:20142811. https://doi.org/10.1098/rspb.2014.2811
- McCollom TM, Shock EL. Geochemical constraints on chemolithoautotrophic metabolism by microorganisms in seafloor hydrothermal systems. *Geochim Cosmochim Acta* 1997;**61**:4375–91. https://doi.org/10.1016/S0016-7037(97) 00241-X
- Hansen M, Perner M. A novel hydrogen oxidizer amidst the sulfur-oxidizing Thiomicrospira lineage. ISME J 2015;9:696–707. https://doi.org/10.1038/ismej.2014.173

- 6. Böhnke S, Sass K, Gonnella G et al. Parameters governing the community structure and element turnover in Kermadec volcanic ash and hydrothermal fluids as monitored by inorganic electron donor consumption, autotrophic CO₂ fixation and 16S tags of the transcriptome in incubation experiments. Front Microbiol 2019;10:2296. https://doi.org/10.3389/fmicb.2019.02296
- Ruby EG, Jannasch HW. Physiological characteristics of Thiomicrospira sp. strain L-12 isolated from deep-sea hydrothermal vents. J Bacteriol 1982;149:161–5. https://doi.org/10.1128/ jb.149.1.161-165.1982
- Jannasch HW, Wirsen CO, Nelson DC et al. Thiomicrospira crunogena sp. nov., a colorless, sulfur-oxidizing bacterium from a deepsea hydrothermal vent. Int J Syst Bacteriol 1985;35:422–4. https:// doi.org/10.1099/00207713-35-4-422
- Cao H, Wang Y, Lee OO et al. Microbial sulfur cycle in two hydrothermal chimneys on the southwest Indian ridge. MBio 2014;5:10–1128. https://doi.org/10.1128/mBio.00980-13
- Ruby EG, Wirsen CO, Jannasch HW. Chemolithotrophic sulfuroxidizing bacteria from the Galapagos rift hydrothermal vents. *Appl Environ Microbiol* 1981;42:317–24. https://doi.org/10.1128/ aem.42.2.317-324.1981
- Brinkhoff T, Muyzer G. Increased species diversity and extended habitat range of sulfur-oxidizing Thiomicrospira spp. Appl Environ Microbiol 1997;63:3789–96. https://doi.org/10.1128/ aem.63.10.3789-3796.1997
- Takai K, Nealson KH, Horikoshi K. Hydrogenimonas thermophila gen. Nov., sp. nov., a novel thermophilic, hydrogen-oxidizing chemolithoautotroph within the ε-Proteobacteria, isolated from a black smoker in a central Indian ridge hydrothermal field. Int J Syst Evol Microbiol 2004;54:25–32. https://doi.org/10.1099/ ijs.0.02787-0
- Boden R, Scott KM, Williams J et al. An evaluation of Thiomicrospira, Hydrogenovibrio and Thioalkalimicrobium: reclassification of four species of Thiomicrospira to each Thiomicrorhabdus gen. Nov. and Hydrogenovibrio, and reclassification of all four species of Thioalkalimicrobium to Thiomicrospira. Int J Syst Evol Microbiol 2017;67:1140–51. https://doi.org/10.1099/ijsem.0.001855
- Jiang L, Lyu J, Shao Z. Sulfur metabolism of Hydrogenovibrio thermophilus strain S5 and its adaptations to deep-sea hydrothermal vent environment. Front Microbiol 2017;8:1–12. https://doi. org/10.3389/fmicb.2017.02513
- Barco RA, Hoffman CL, Ramírez GA et al. In-situ incubation of iron-sulfur mineral reveals a diverse chemolithoautotrophic community and a new biogeochemical role for *Thiomicrospira*. Environ Microbiol 2017;19:1322–37. https://doi. org/10.1111/1462-2920.13666
- Brazelton WJ, Baross JA. Metagenomic comparison of two Thiomicrospira lineages inhabiting contrasting deep-sea hydrothermal environments. PLoS One 2010;5:e13530. https://doi.org/10.1371/ journal.pone.0013530
- Adam-Beyer N, Laufer-Meiser K, Fuchs S et al. Microbial ecosystem assessment and hydrogen oxidation potential of newly discovered vent systems from the central and south-east Indian ridge. Front Microbiol 2023;14:1173613. https://doi.org/10.3389/ fmicb.2023.1173613
- Laufer K, Nordhoff M, Røy H et al. Coexistence of microaerophilic, nitrate-reducing, and phototrophic Fe(II) oxidizers and Fe(III) reducers in coastal marine sediment. Appl Environ Microbiol 2016;82:1433–47. https://doi.org/10.1128/AEM.03527-15
- Lane DJ. 16S/23S rRNA sequencing. In: Stackebrandt E.G., Goodfellow M. (eds.), Nucleic Acid Techniques Inbacterial Systematics. Hoboken: John Wiley and Sons, 1991, 115–75.

- Kumar S, Stecher G, Li M et al. MEGA X: molecular evolutionary genetics analysis across computing platforms. Mol Biol Evol 2018;35:1547–9. https://doi.org/10.1093/molbev/msy096
- Thompson JD, Higgins DG, Gibson TJ. CLUSTAL W: improving the sensitivity of progressive multiple sequence alignment through sequence weighting, position-specific gap penalties and weight matrix choice. Nucleic Acids Res 1994;22:4673–80. https://doi. org/10.1093/nar/22.22.4673
- Emerson D, Moyer C. Isolation and characterization of novel iron-oxidizing bacteria that grow at circumneutral pH. Environ Microbiol 1997;63:4784–92. https://doi.org/10.1128/aem.63.12. 4784-4792.1997
- Emerson D, Floyd MM. Enrichment and isolation of ironoxidizing bacteria at neutral pH. Methods Enzymol 2005;397: 112-23. https://doi.org/10.1016/S0076-6879(05)97006-7
- Stookey LL. Ferrozine-a new spectrophotometric reagent for iron. Anal Chem 1970;42:779–81. https://doi.org/10.1021/ac60289 a016
- Steudel R, Holdt G, Göbel T et al. Chromatographic separation of higher polythionates SnO (n = 3...22) and their detection in cultures of Thiobacillus ferroxidans; molecular composition of bacterial sulfur secretions. Angew Chemie Int Ed English 1987;26: 151–3. https://doi.org/10.1002/anie.198701511
- Schippers A, Jørgensen BB. Oxidation of pyrite and iron sulfide by manganese dioxide in marine sediments. *Geochim Cos*mochim Acta 2001;65:915–22. https://doi.org/10.1016/S0016-7037 (00)00589-5
- Laufer K, Røy H, Jørgensen BB et al. Evidence for the existence of autotrophic nitrate-reducing Fe(II)-oxidizing bacteria in marine coastal sediment. Appl Env Microbiol 2016;82:6120–31. https://doi. org/10.1128/AEM.01570-16
- Böhnke S, Perner M. A function-based screen for seeking RubisCO active clones from metagenomes: novel enzymes influencing RubisCO activity. ISME J 2015;9:735–45. https://doi. org/10.1038/ismej.2014.163
- Jakob R, Saenger W. Reversed-phase ion-pair chromatographic separation of ribulose 1,5-bisphosphate from 3phosphoglycerate and its application as a new enzyme assay for RuBP carboxylase/oxygenase. FEBS Lett 1985;183:111–4. https:// doi.org/10.1016/0014-5793(85)80965-0
- Garber AI, Nealson KH, Okamoto A et al. FeGenie: a comprehensive tool for the identification of iron genes and iron gene neighborhoods in genome and metagenome assemblies. Front Microbiol 2020;11:499513. https://doi.org/10.3389/fmicb.2020.00037
- Perner M, Hentscher M, Rychlik N et al. Driving forces behind the biotope structures in two low-temperature hydrothermal venting sites on the southern mid-Atlantic ridge. Environ Microbiol Rep 2011;3:727–37. https://doi.org/10.1111/j.1758-2229.2011. 00291.x
- Perner M, Hansen M, Seifert R et al. Linking geology, fluid chemistry, and microbial activity of basalt-and ultramafic-hosted deep-sea hydrothermal vent environments. *Geobiology* 2013;11: 340–55. https://doi.org/10.1111/gbi.12039
- Toyoda K, Ishii M, Arai H. Function of three RuBisCO enzymes under different CO₂ conditions in Hydrogenovibrio marinus. JBiosci Bioeng 2018;**126**:730–5. https://doi.org/10.1016/j. jbiosc.2018.06.005
- Moeck GS, Coulton JW. TonB-dependent iron acquisition: mechanisms of siderophore-mediated active transport. Mol Microbiol 1998;28:675–81. https://doi.org/10.1046/j.1365-2958. 1998.00817.x

- Ferguson AD, Deisenhofer J. TonB-dependent receptors structural perspectives. Biochim Biophys Acta Biomembr 2002; 1565:318–32. https://doi.org/10.1016/S0005-2736(02)00578-3
- Ringel MT, Brüser T. The biosynthesis of pyoverdines. Microb. Cell 2018;5:424–37. https://doi.org/10.15698/mic2018.10.649
- Scott KM, Sievert SM, Abril FN et al. The genome of deep-sea vent chemolithoautotroph Thiomicrospira crunogena XCL-2. PLoS Biol 2006;4:2196–212. https://doi.org/10.1371/journal.pbio.0040383
- Hansen M, Perner M. Hydrogenase gene distribution and H₂ consumption ability within the Thiomicrospira lineage. Front Microbiol 2016;**7**:99. https://doi.org/10.3389/fmicb.2016.00099
- Frigaard NU, Bryant DA. Genomic insights into the sulfur metabolism of phototrophic green sulfur bacteria. In: Hell R., Dahl C., Knaff D. et al. (eds.), Sulfur Metabolism in Phototrophic Organisms. Dordrecht: Springer, 2008, 337–55 10.1007/978-1-4020-6863-8_17.
- Gregersen LH, Bryant DA, Frigaard NU. Mechanisms and evolution of oxidative sulfur metabolism in green sulfur bacteria. Front Microbiol 2011;2:116. https://doi.org/10.3389/ fmicb.2011.00116
- 41. Scott KM, Williams J, Porter CM et al. Genomes of ubiquitous marine and hypersaline Hydrogenovibrio, Thiomicrorhabdus and Thiomicrospira spp. encode a diversity of mechanisms to sustain chemolithoautotrophy in heterogeneous environments. Environ Microbiol 2018;20:2686–708. https://doi. org/10.1111/1462-2920.14090
- Erb TJ, Zarzycki J. A short history of RubisCO: the rise and fall (?) of Nature's predominant CO₂ fixing enzyme. *Curr Opin Biotechnol* 2018;**49**:100–7. https://doi.org/10.1016/j.copbio.2017.07.017
- Yoichi Y, Koichi T, Hiroyuki A et al. CO₂-responsive expression and gene organization of three ribulose-1,5-bisphosphate carboxylase/oxygenase enzymes and carboxysomes in Hydrogenovibrio marinus strain MH-110. J Bacteriol 2004;186:5685–91. https://doi.org/10.1128/JB.186.17.5685-5691.2004
- 44. Scott KM, Leonard JM, Boden R et al. Diversity in CO₂concentrating mechanisms among chemolithoautotrophs from the genera Hydrogenovibrio, Thiomicrorhabdus, and Thiomicrospira, ubiquitous in sulfidic habitats worldwide. Appl Environ Microbiol 2019;85:e02096–18. https://doi.org/10.1128/AEM.02096-18
- Dobrinski KP, Longo DL, Scott KM. The carbon-concentrating mechanism of the hydrothermal vent chemolithoautotroph Thiomicrospira crunogena. J Bacteriol 2005;187:5761–6. https://doi. org/10.1128/JB.187.16.5761-5766.2005
- Adam N, Han Y, Laufer-Meiser K et al. Deltaproteobacterium strain KaireiS1, a mesophilic, hydrogen-oxidizing and sulfatereducing bacterium from an inactive deep-sea hydrothermal chimney. Front Microbiol 2021;12:686276. https://doi.org/10.3389/ fmicb.2021.686276
- Qin H, Guan X, Bandstra JZ et al. Modeling the kinetics of hydrogen formation by zerovalent iron: effects of sulfidation on micro- and nano-scale particles. Environ Sci Technol 2018;52: 13887–96. https://doi.org/10.1021/acs.est.8b04436
- Mori JF, Scott JJ, Hager KW et al. Physiological and ecological implications of an iron- or hydrogen-oxidizing member of the Zetaproteobacteria, *Ghiorsea bivora*, gen. Nov., sp. nov. ISME J 2017;**11**:2624. https://doi.org/10.1038/ismej.2017.132
- Emerson D, Rentz JA, Lilburn TG et al. A novel lineage of proteobacteria involved in formation of marine Fe-oxidizing microbial mat communities. PLoS One 2007;2:e667. https://doi. org/10.1371/journal.pone.0000667
- McAllister SM, Moore RM, Gartman A et al. The Fe(II)-oxidizing Zetaproteobacteria: historical, ecological and genomic perspectives. FEMS Microbiol Ecol 2019;95:fiz015. https://doi.org/10.1093/ femsec/fiz015

- Hribovšek P, Olesin Denny E, Dahle H et al. Putative novel hydrogen-and iron-oxidizing sheath-producing Zetaproteobacteria thrive at the Fåvne deep-sea hydrothermal vent field. Msyst 2023;8:e00543–23. https://doi.org/10.1128/msystems.00543-23
- Baker IR, Matzen SL, Schuler CJ et al. Aerobic iron-oxidizing bacteria secrete metabolites that markedly impede abiotic iron oxidation. PNAS nexus 2023;2:pgad421. https://doi.org/10.1093/ pnasnexus/pgad421
- Neely C, Bou Khalil C, Cervantes A et al. Genome sequence of Hydrogenovibrio sp. strain SC-1, a chemolithoautotrophic sulfur and iron oxidizer. Genome Announc 2018;6:4–5. https://doi. org/10.1128/genomeA.01581-17
- Hoffman CL, Monreal PJ, Albers JB et al. Microbial strong organic ligand production is tightly coupled to iron in hydrothermal plumes. bioRxiv 2023; 2023.01.05.522639. https:// doi.org/10.1101/2023.01.05.522639
- Gledhill M, Buck KN. The organic complexation of iron in the marine environment: a review. Front Microbiol 2012;3:69. https:// doi.org/10.3389/fmicb.2012.00069
- Noinaj N, Guillier M, Barnard TJ et al. TonB-dependent transporters: regulation, structure, and function. Ann Rev Microbiol 2010;64:43–60. https://doi.org/10.1146/annurev. micro.112408.134247
- Ziervogel K, Arnosti C. Substantial carbohydrate hydrolase activities in the water column of the Guaymas Basin (gulf of California). Front Mar Sci 2020;6:815. https://doi.org/10.3389/ fmars.2019.00815
- Li M, Toner BM, Baker BJ et al. Microbial iron uptake as a mechanism for dispersing iron from deep-sea hydrothermal vents. Nat Commun 2014;5:3192. https://doi.org/10.1038/ncomms4192
- Amend JP, Shock EL. Energetics of overall metabolic reactions of thermophilic and hyperthermophilic archaea and bacteria. FEMS Microbiol Rev 2001;25:175–243. https://doi.org/10.1111/ j.1574-6976.2001.tb00576.x
- Fuchs G. Allgemeine Mikrobiologie, 11th edn. Stuttgart: Georg Thieme Verlag, 2007, 10.1055/b-002-44938.
- Emerson D, Fleming EJ, McBeth JM. Iron-oxidizing bacteria: an environmental and genomic perspective. Ann Rev Microbiol 2010;64:561–83. https://doi.org/10.1146/annurev.micro.112408. 134208
- 62. Boschker HTS, Vasquez-Cardenas D, Bolhuis H *et al.* Chemoautotrophic carbon fixation rates and active bacterial communities in intertidal marine sediments. *PLoS One* 2014;**9**:e101443. https://doi.org/10.1371/journal.pone.0101443
- Jørgensen BB, Nelson DC. Sulfide oxidation in marine sediments: geochemistry meets microbiology. Geol Soc Am Spec Pap 2004;379: 63–81. https://doi.org/10.1130/0-8137-2379-5.63
- 64. Howarth RW, Jørgensen BB. Formation of ³⁵S-labelled elemental sulfur and pyrite in coastal marine sediments (Limfjorden and Kysing Fjord, Denmark) during short-term ³⁵SO₄²⁻ reduction measurements. *Geochim Cosmochim Acta* 1984;**48**:1807–18. https://doi.org/10.1016/0016-7037(84)90034-6
- Thomsen U, Kristensen E. Dynamics of ΣCO₂ in a surficial sandy marine sediment: the role of chemoautotrophy. Aquat Microb Ecol 1997;12:165–76. https://doi.org/10.3354/ame012165
- Wuchter C, Abbas B, Coolen MJ et al. Archaeal nitrification in the ocean. Proc Natl Acad Sci 2006;103:12317–22. https://doi. org/10.1073/pnas.0600756103
- 67. Berg C, Listmann L, Vandieken V *et al*. Chemoautotrophic growth of ammonia-oxidizing Thaumarchaeota enriched from a pelagic redox gradient in the Baltic Sea. Front Microbiol 2014;**5**:786.
- Billen G. Evaluation of nitrifying activity in sediments by dark ¹⁴C-bicarbonate incorporation. Water Res 1976;10:51–7. https:// doi.org/10.1016/0043-1354(76)90157-3

- Kelly DP. Biochemistry of the chemolithotrophic oxidation of inorganic Sulphur. Philos Trans R Soc Lond Biol Sci 1982;298:499–528. https://doi.org/10.1098/rstb.1982. 0094
- Timmer-ten HA. Cell yield and bioenergetics of Thiomicrospira denitrificans compared with Thiobacillus denitrificans. Antonie Van Leeuwenhoek 1981;47:231–43. https://doi.org/10.1007/ BF00403394
- Nelson DC, Jørgensen BB, Revsbech NP. Growth pattern and yield of a chemoautotrophic Beggiatoa sp. in oxygen-sulfide microgradients. Appl Env Microbiol 1986;52:225–33. https://doi. org/10.1128/aem.52.2.225-233.1986
- Neubauer SC, Emerson D, Megonigal JP. Life at the energetic edge: kinetics of circumneutral iron oxidation by lithotrophic iron-oxidizing bacteria isolated from the wetland-plant rhizosphere. Appl Env Microbiol 2002;68:3988–95. https://doi. org/10.1128/AEM.68.8.3988-3995.2002

- Sobolev D, Roden EE. Characterization of a neutrophilic, chemolithoautotrophic Fe(II)-oxidizing β-Proteobacterium from freshwater wetland sediments. *Geomicrobiol J* 2004;**21**:1–10. https://doi.org/10.1080/01490450490253310
- Sarrazin J, Rodier P, Tivey MK et al. A dual sensor device to estimate fluid flow velocity at diffuse hydrothermal vents. *Deep* Sea Res Part I Oceanogr Res Pap 2009;56:2065–74. https://doi. org/10.1016/j.dsr.2009.06.008
- Wankel SD, Germanovich LN, Lilley MD et al. Influence of subsurface biosphere on geochemical fluxes from diffuse hydrothermal fluids. Nat Geosci 2011;4:461–8. https://doi.org/10.1038/ ngeo1183
- Takai K, Campbell BJ, Cary SC et al. Enzymatic and genetic characterization of carbon and energy metabolisms by deepsea hydrothermal chemolithoautotrophic isolates of Epsilonproteobacteria. Appl Environ Microbiol 2005;**71**:7310–20. https://doi. org/10.1128/AEM.71.11.7310-7320.2005