

Marine Biotoxins: Determination of Spirolide Profiles in Phytoplankton by LC/MS/MS

1. Introduction
2. Search for unknowns
3. Mass spectral characterization
4. Phenotypic profiling of *Alexandrium ostenfeldii* strains
5. Pitfalls

Ecological Chemistry: Allelochemical Effects of Protists

Allelochemistry:

Interaction of *biologically active components eliciting specific responses* in target organisms.

These highly specific allelochemical compounds are typically *secondary metabolites* and should be distinguished from low molecular weight inorganic and organic nutrients and complex but poorly defined dissolved organic matter (DOM) that may be utilized as growth substrates by protists.

There are both *stimulatory* and *inhibitory* functions to be exploited via production of allelochemicals by protists. Among the putative functions of allelochemicals, their use as agents of chemical defence is most often invoked.

Ecological Chemistry: Allelochemical Effects of Protists

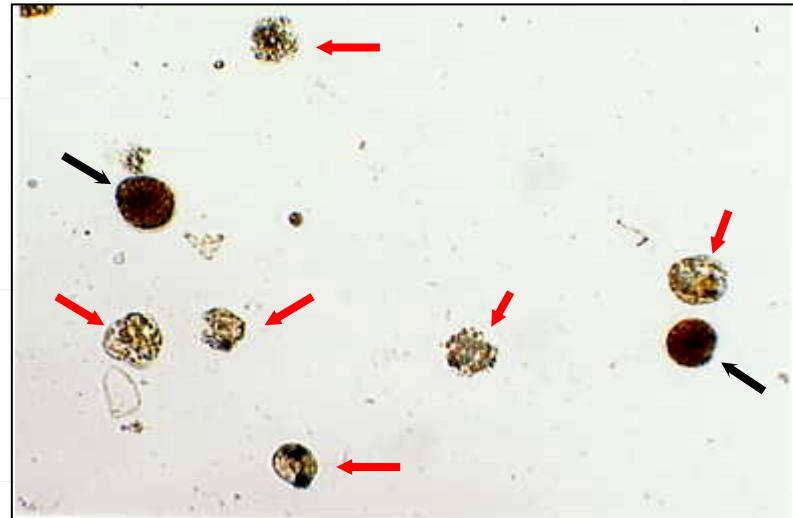
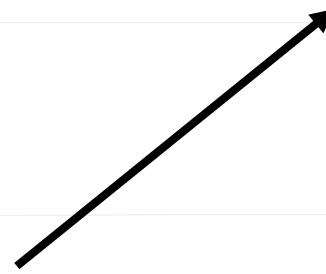
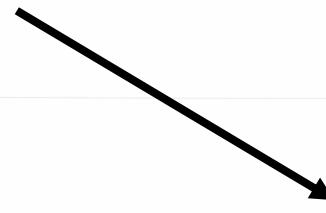


Oxyrrhis marina



Alexandrium ostenfeldii

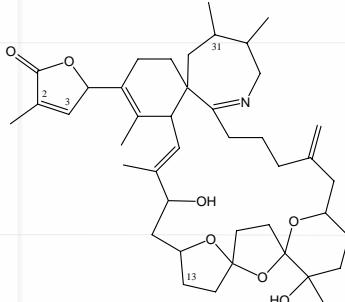
Photos: U. Tillmann



Lytic effect of *Alexandrium*, here the example of *Oxyrrhis marina* (heterotrophic dinoflagellate).

Black arrows: *Alexandrium*;
Red arrows: remains of *Oxyrrhis*

Ecological Chemistry: Allelochemical Effects of Protists

Organism	Effect	chemical interaction	ecological function
<i>Alexandrium ostenfeldii</i>	Accumulation in marine food webs, poisoning of vertebrates and humans	Spirolides  The chemical structure of Spirolide A is shown, featuring a complex polycyclic system with multiple rings, hydroxyl groups (OH), and a terminal alkene group. It includes numbered carbons (2, 3, 13, 31) and a nitrogen atom.	?
	Lysis of other protists	?	Defense against predators and/or competitors

The Discovery of Novel „Fast Acting Toxins“

1991-1992

During routine monitoring of shellfish aquaculture sites in Nova Scotia, extracts of the digestive glands of blue mussels (*Mytilus edulis*) from Ship Harbour and sea scallops (*Placopecten magellanicus*) (Graves Shoal) elicited a *unique toxic response* in the DSP mouse bioassay.....

Coincident consumer complaints of *mild illness* after shellfish consumption

“Fast Acting Toxicity”: Lipophilic extracts (DSP toxins)

Symptomology:

not PSP/DSP(!)

strong convulsions

tail whirling

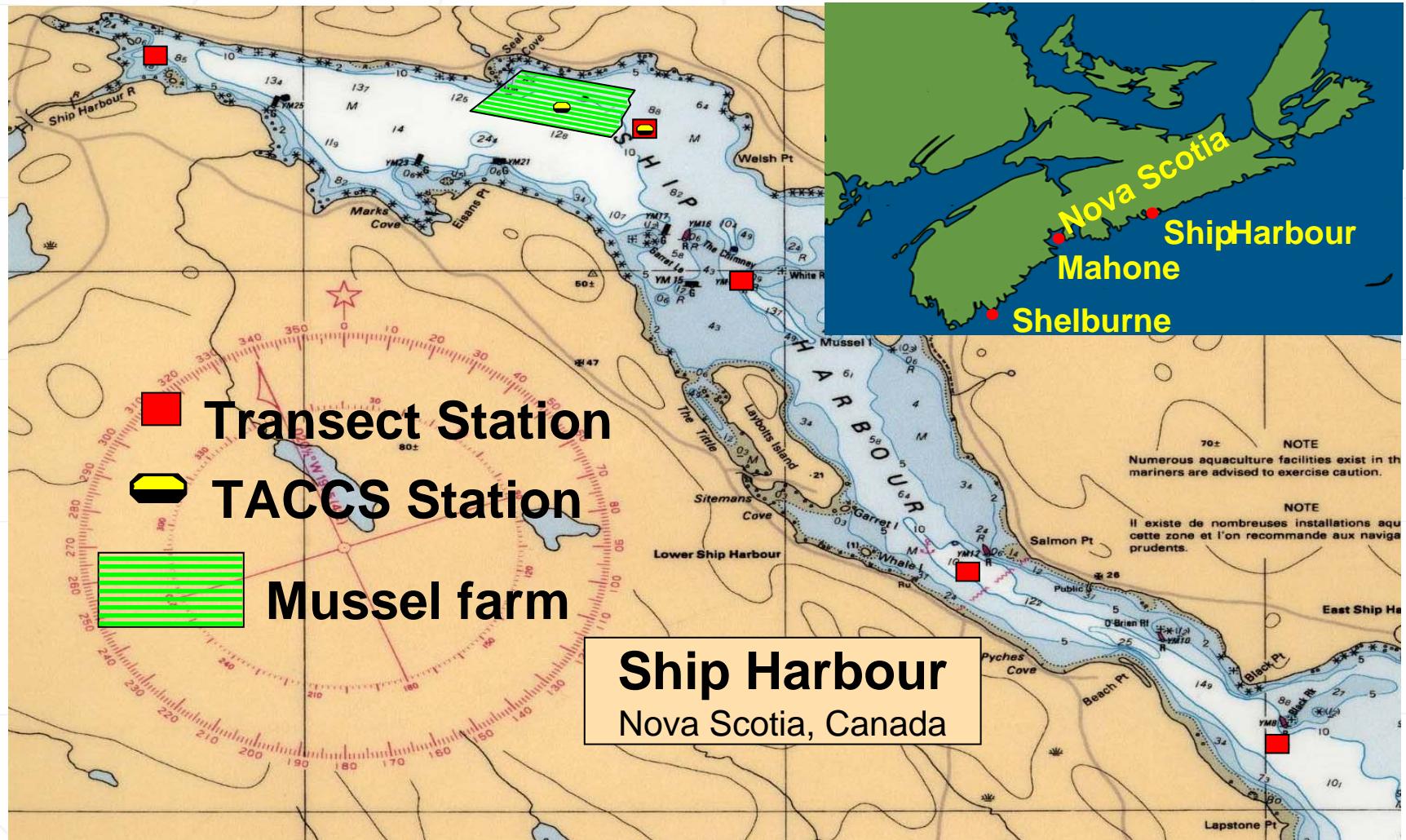
body arching

rapid death (min)

High I.P. Toxicity



“Fast Acting Toxicity”: Occurrence



“Fast Acting Toxicity”: Occurrence

Limfjorden,
Denmark

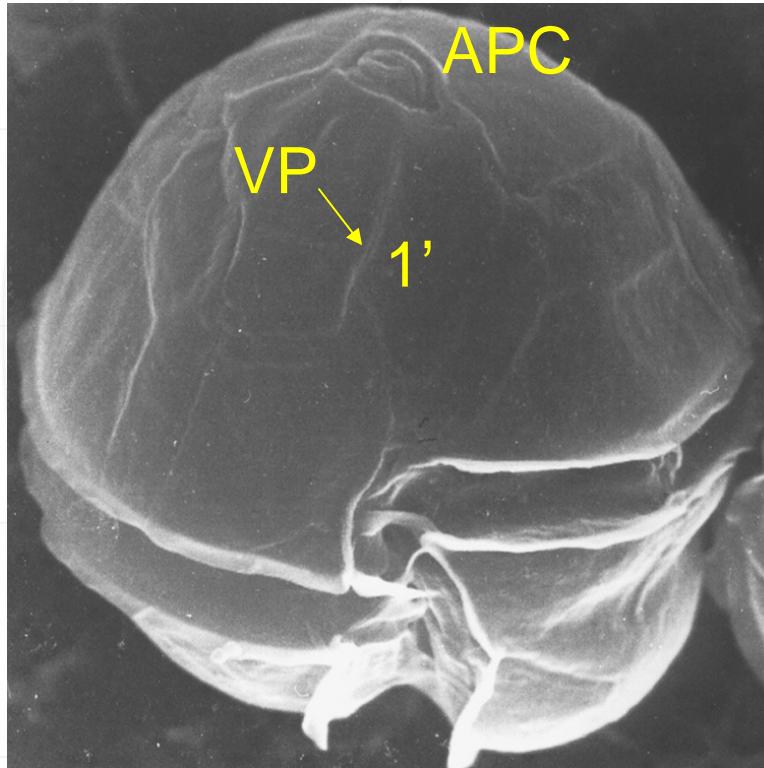
Later also found in:

Norway,

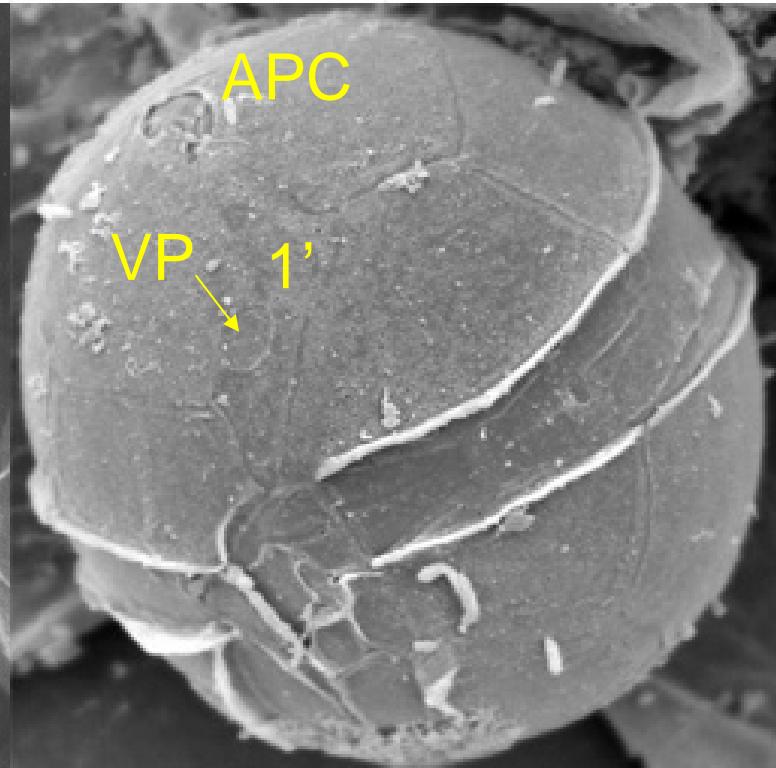
Scotland



Scanning electron micrographs of vegetative cells



Alexandrium tamarense



Alexandrium ostenfeldii

Confirmation of spirolides in cultured isolates from Nova Scotia



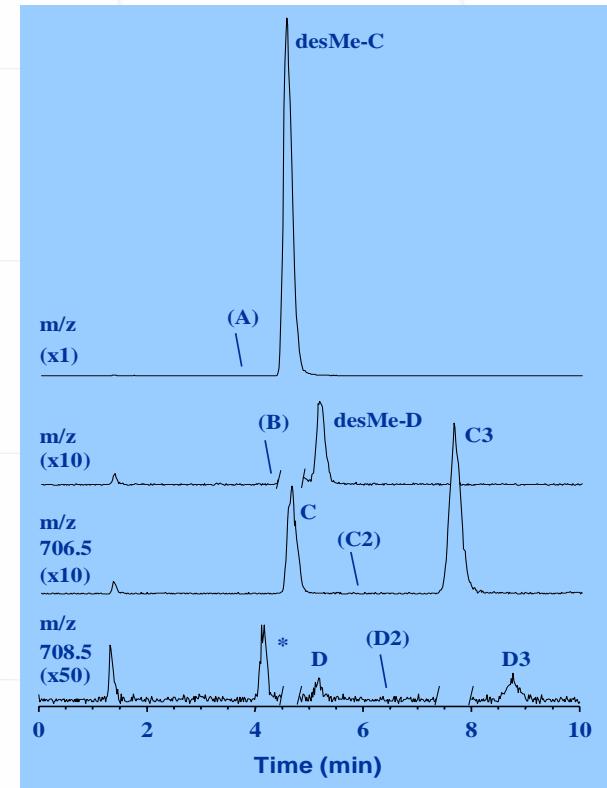
Alexandrium tamarense

Produces PSP
toxins, but no
spirolides



Alexandrium ostenfeldii

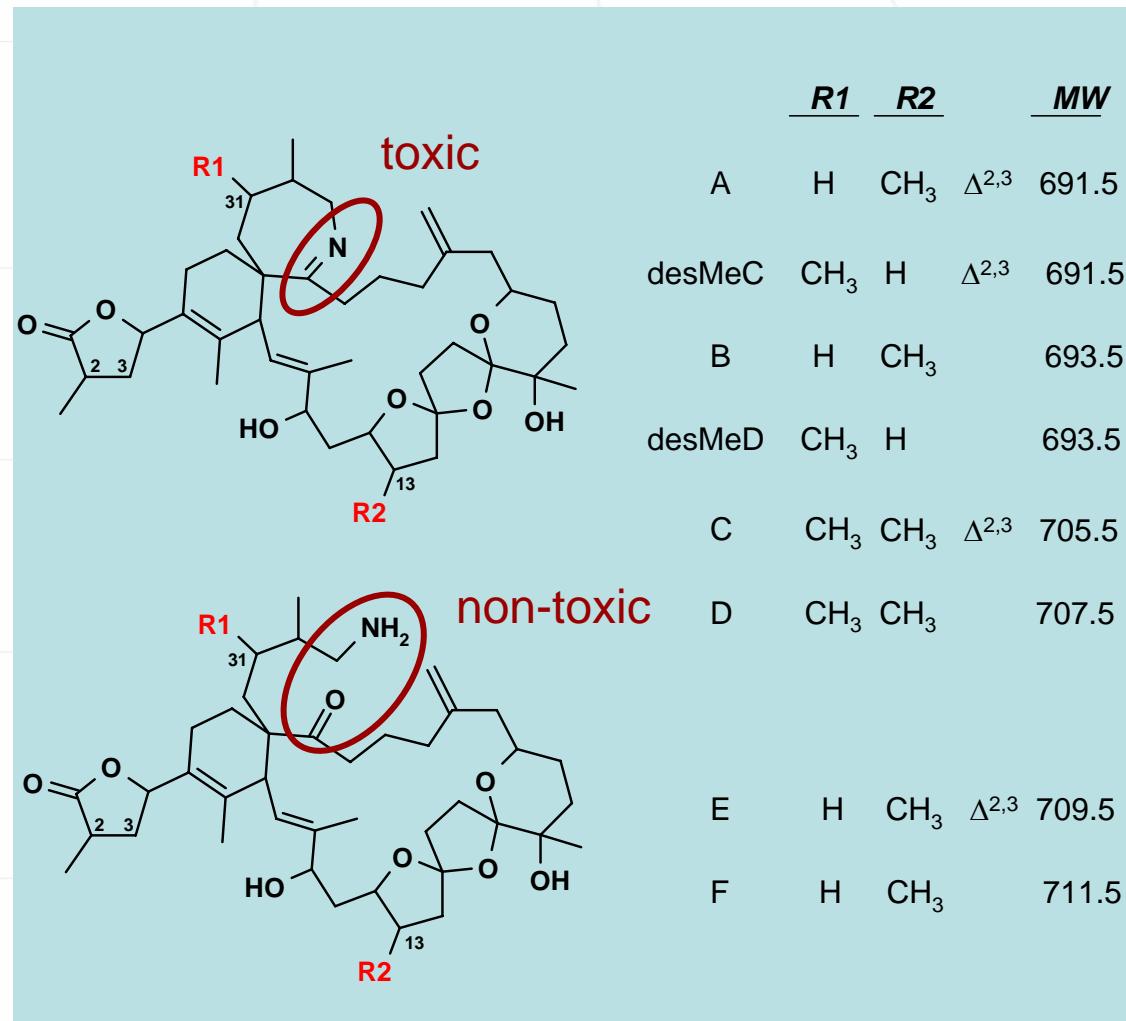
Produces
spirolides,
but no PSP
toxins



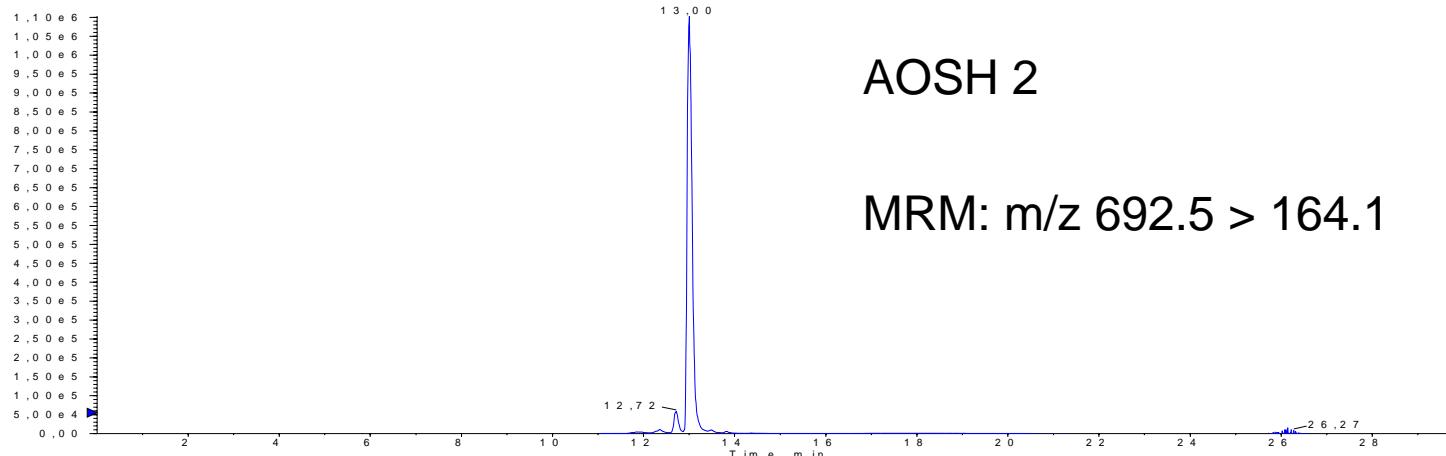
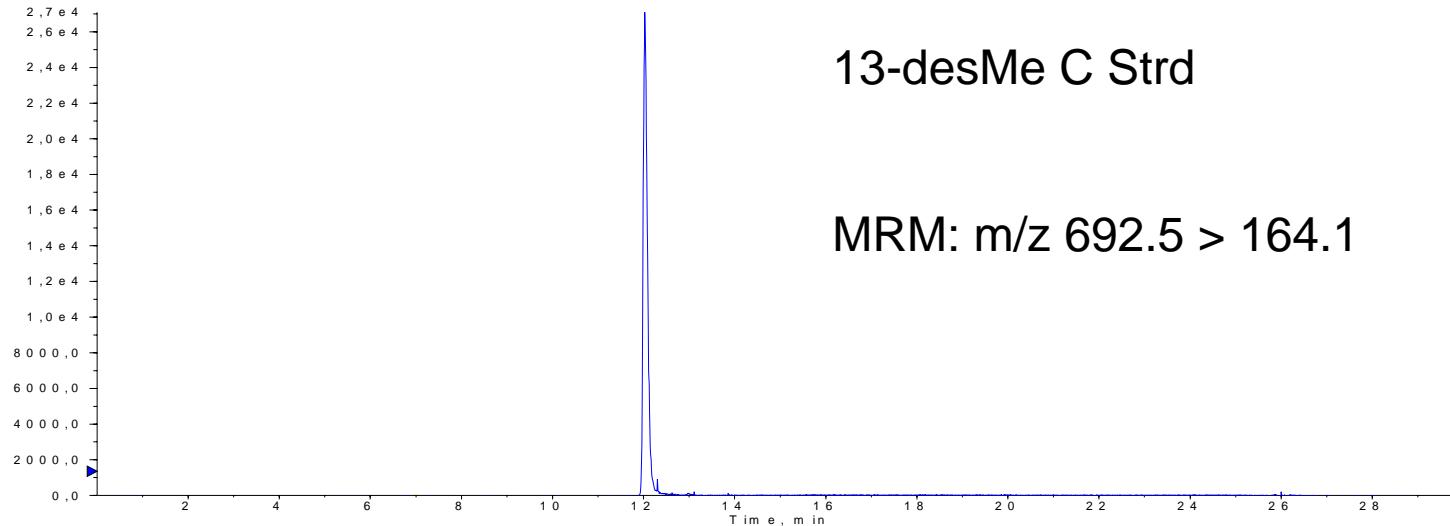
Cause of “Fast Acting Toxicity”

Novel compounds identified as “spirolides”

- macrocyclic imines
- structural similarity to pinnatoxin & gymnodimine
- pharmacologically active/inactive forms

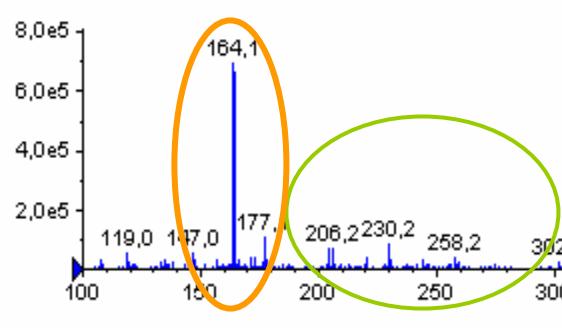


Identical mass transitions – different retention times

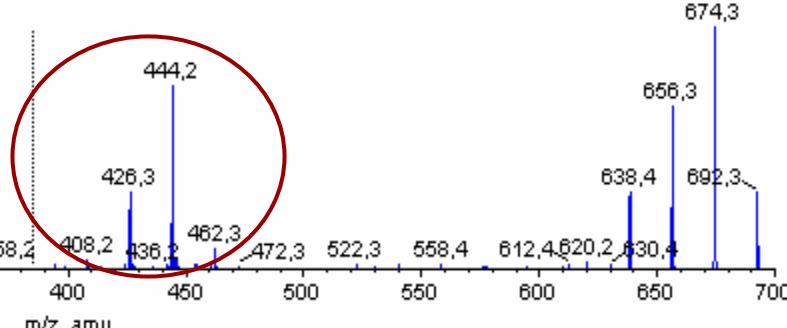


MS/MS spectra of m/z 692.5

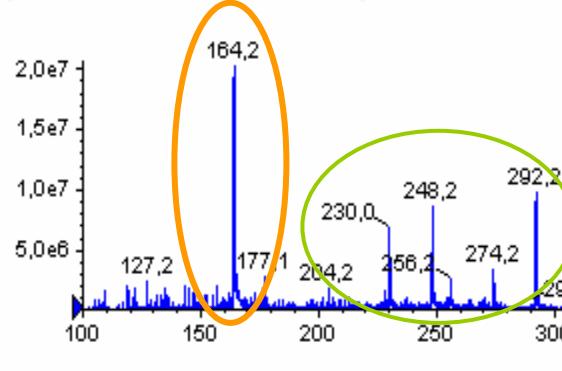
13-desMe C Strd:



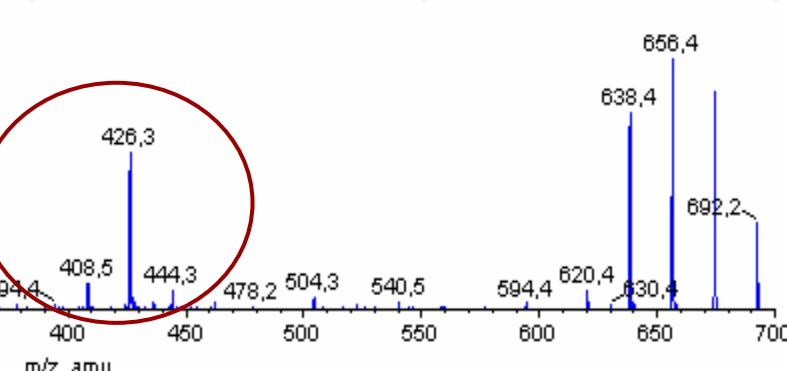
EPI m/z 692.5



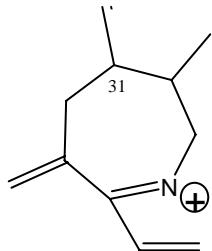
AOSH 2:



EPI m/z 692.5



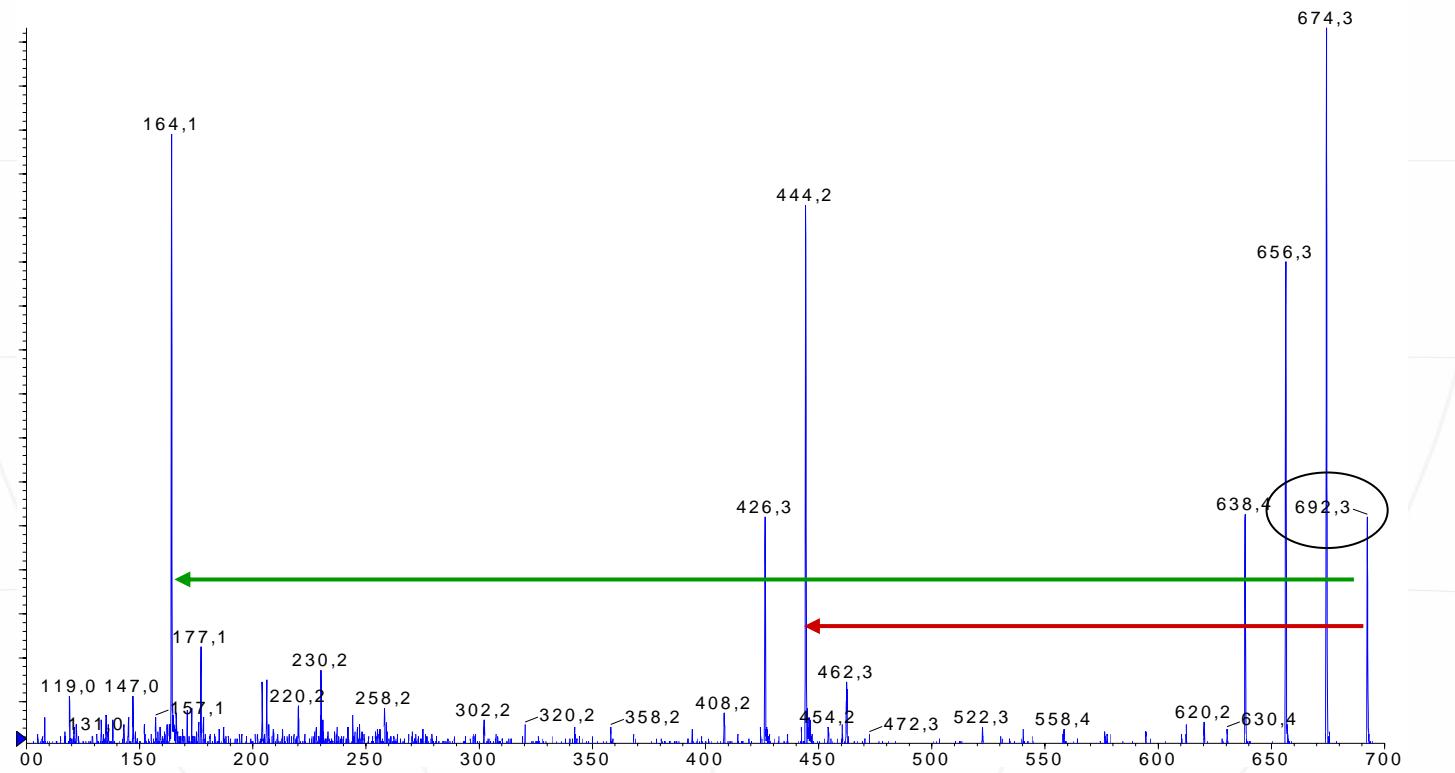
Characteristic spirolide fragment



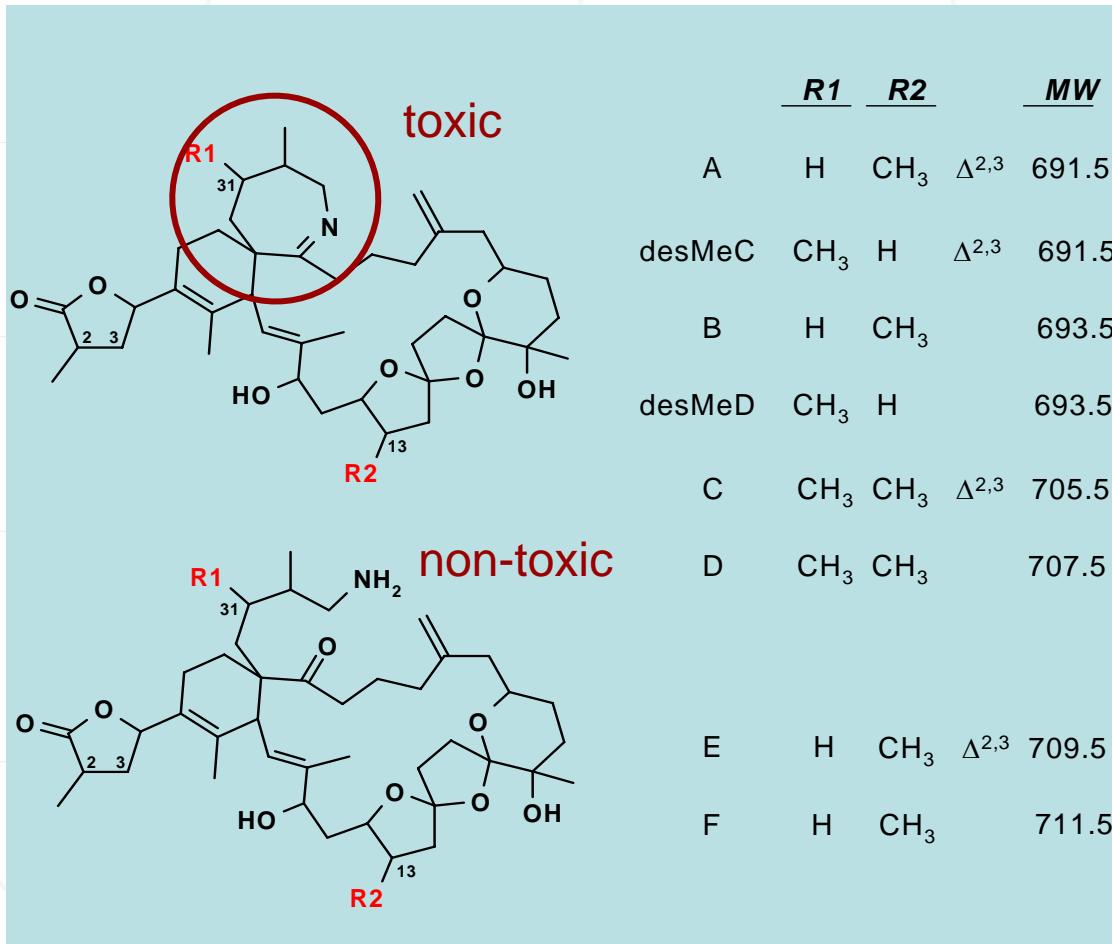
$M+H = 692.5$

$m/z = 444.3$

$m/z = 164.1$



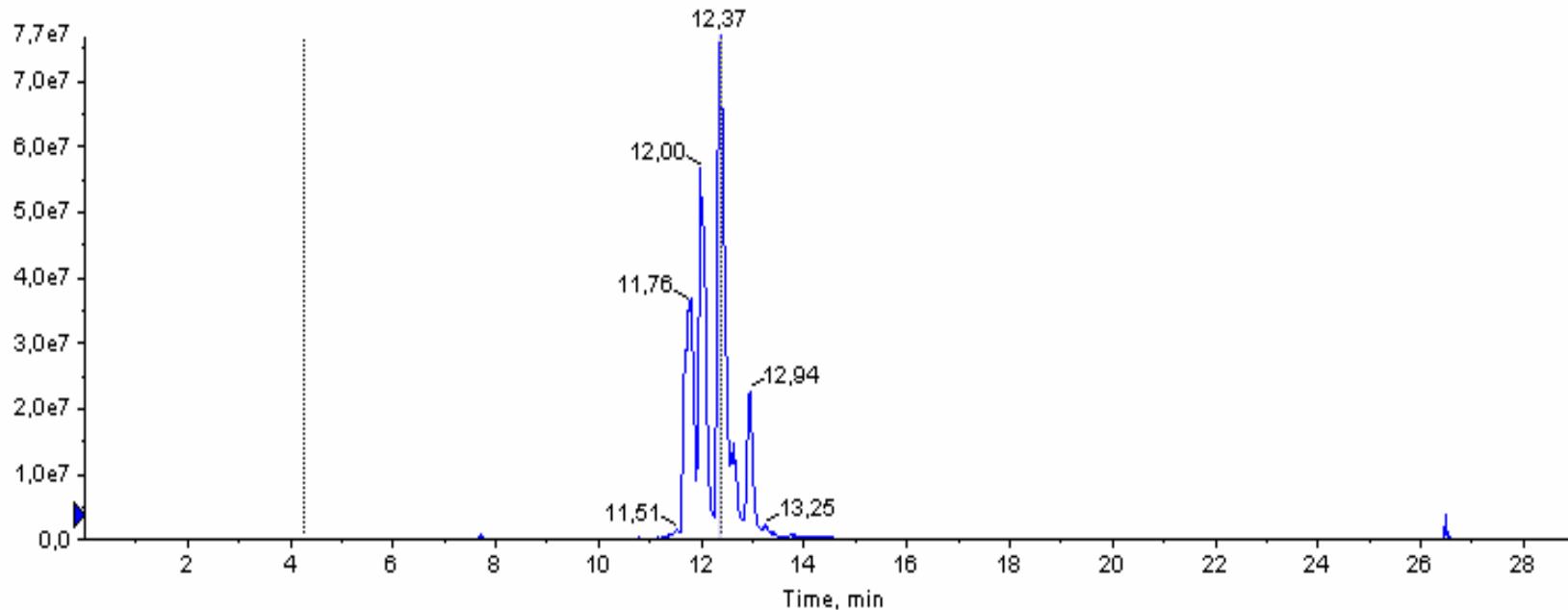
Cyclic imino moiety accounts for toxicity **and** forms a characteristic spirolide fragment



Using the characteristic fragment for the detection of unknown spiroketides

AOSH 2:

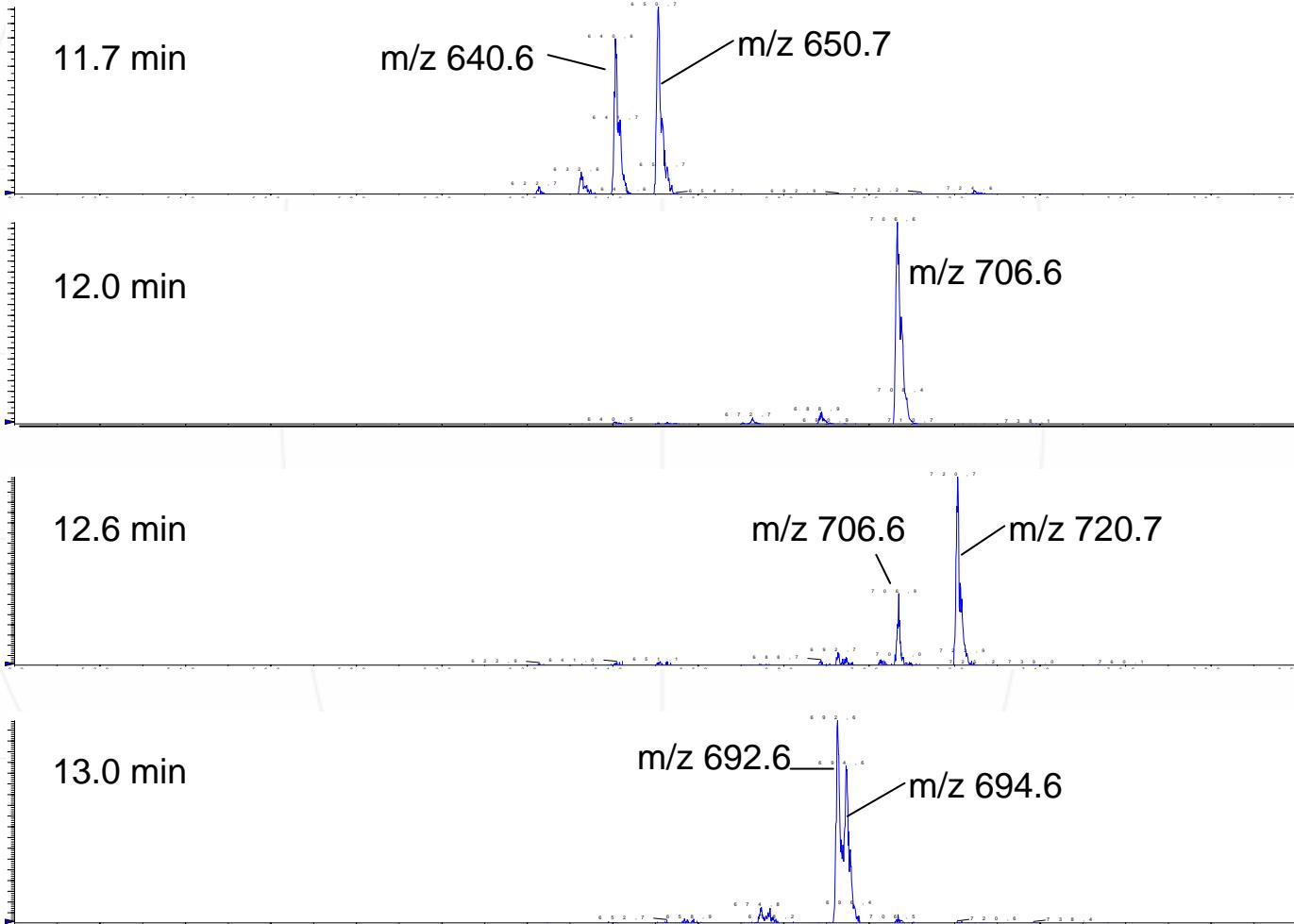
Precursor m/z 164.1



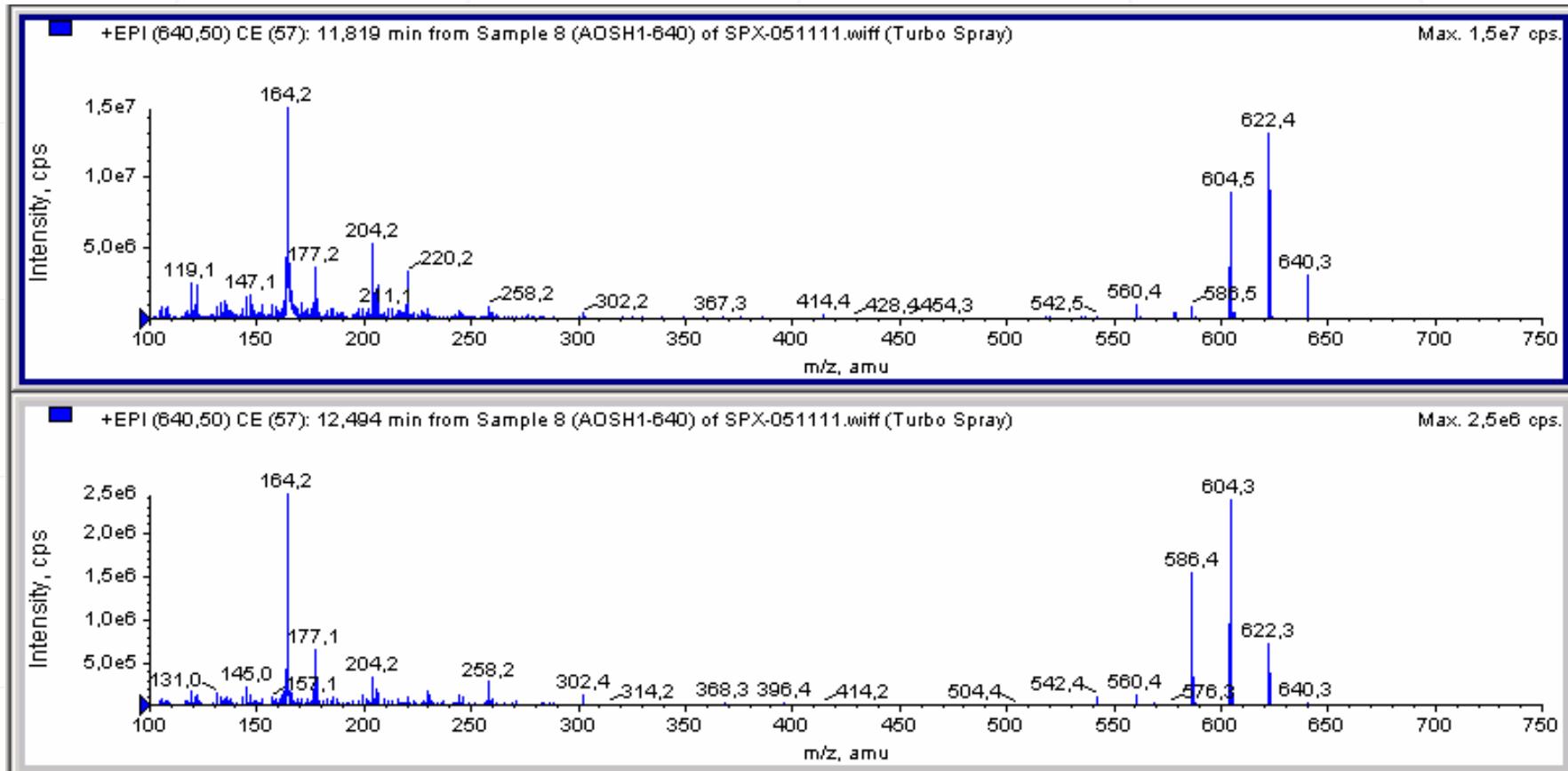
Spirolide masses

AOSH 2:

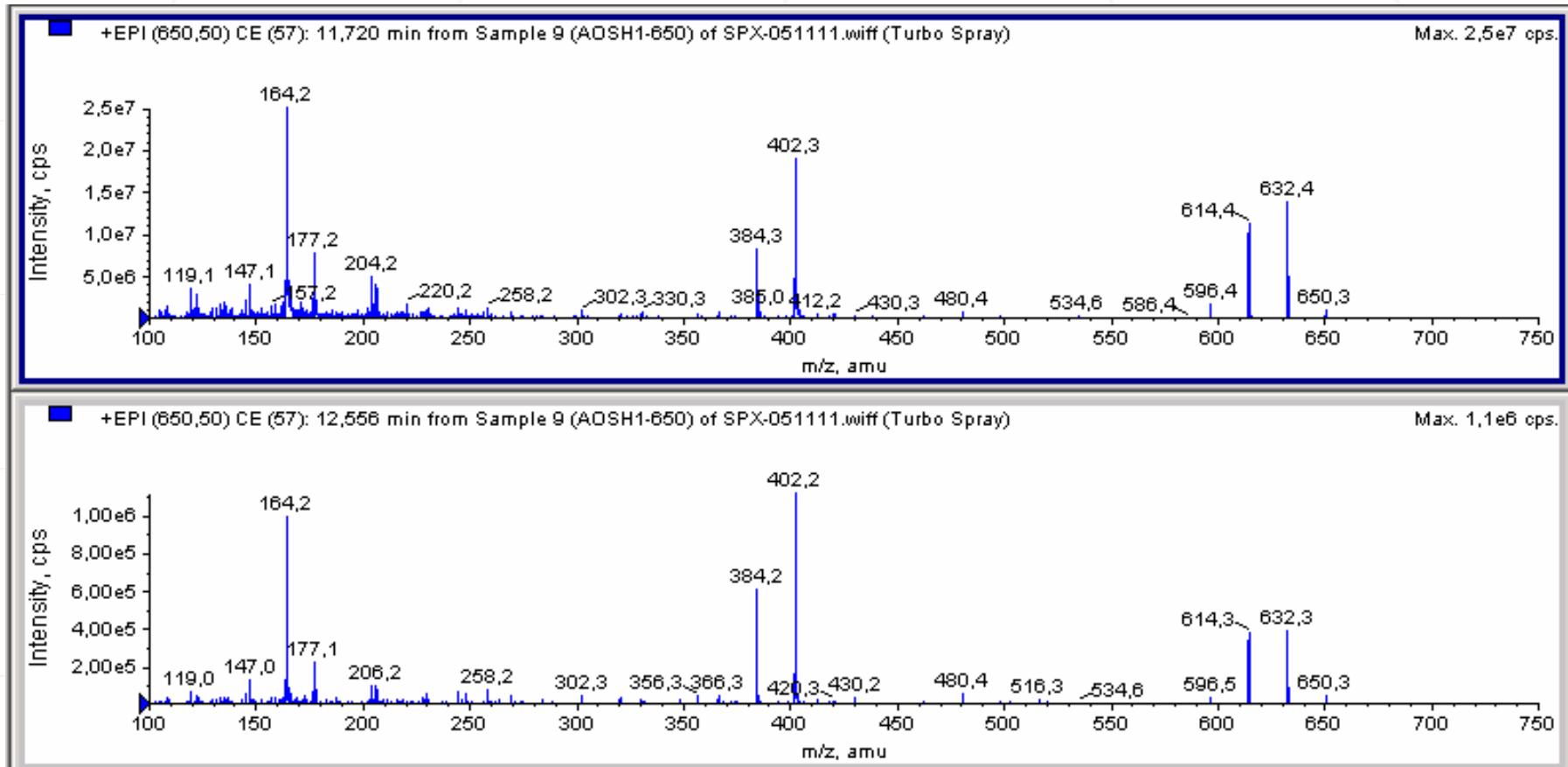
Precursor m/z 164.1



AOSH 2: m/z = 640.5

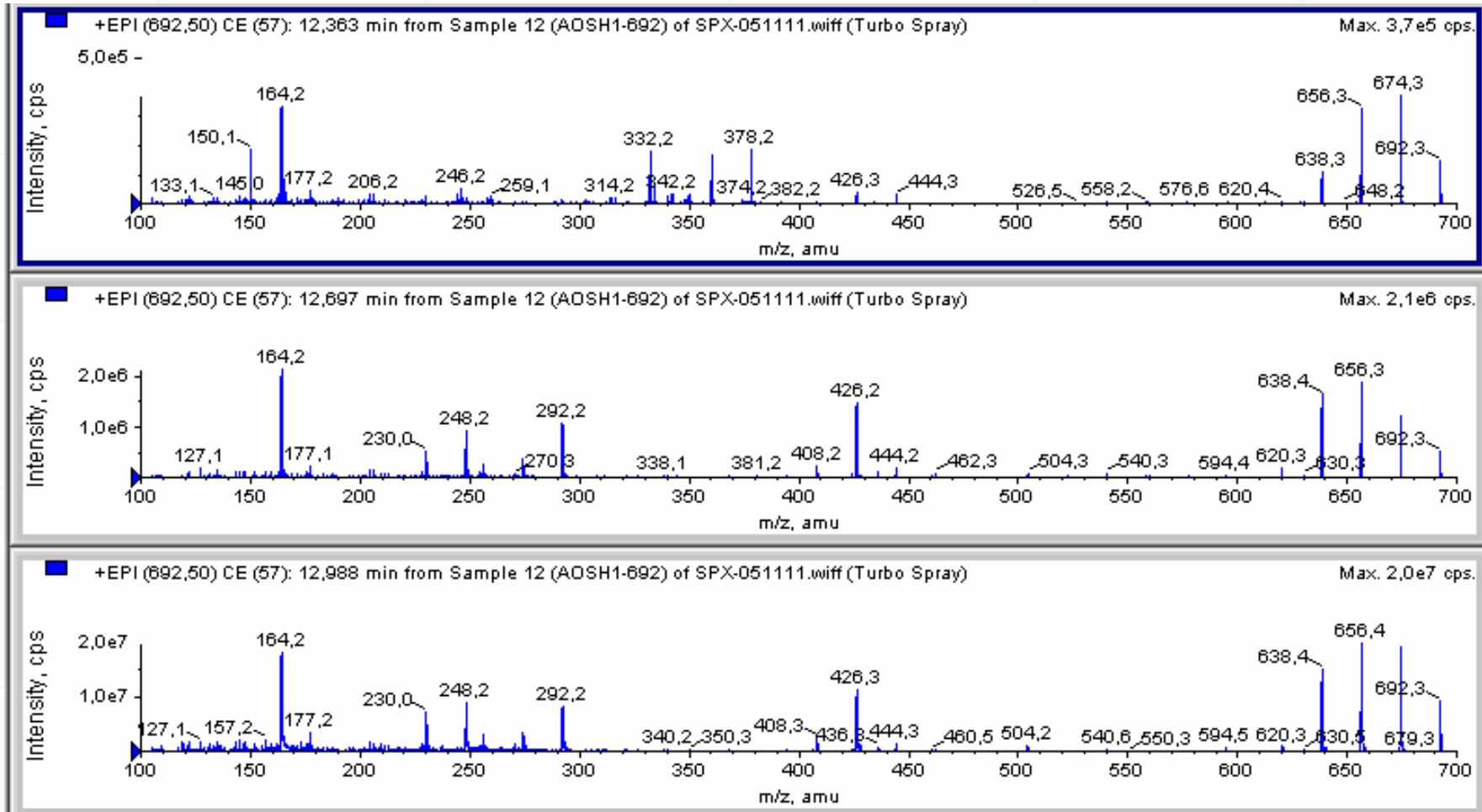


AOSH 2: m/z = 650.5

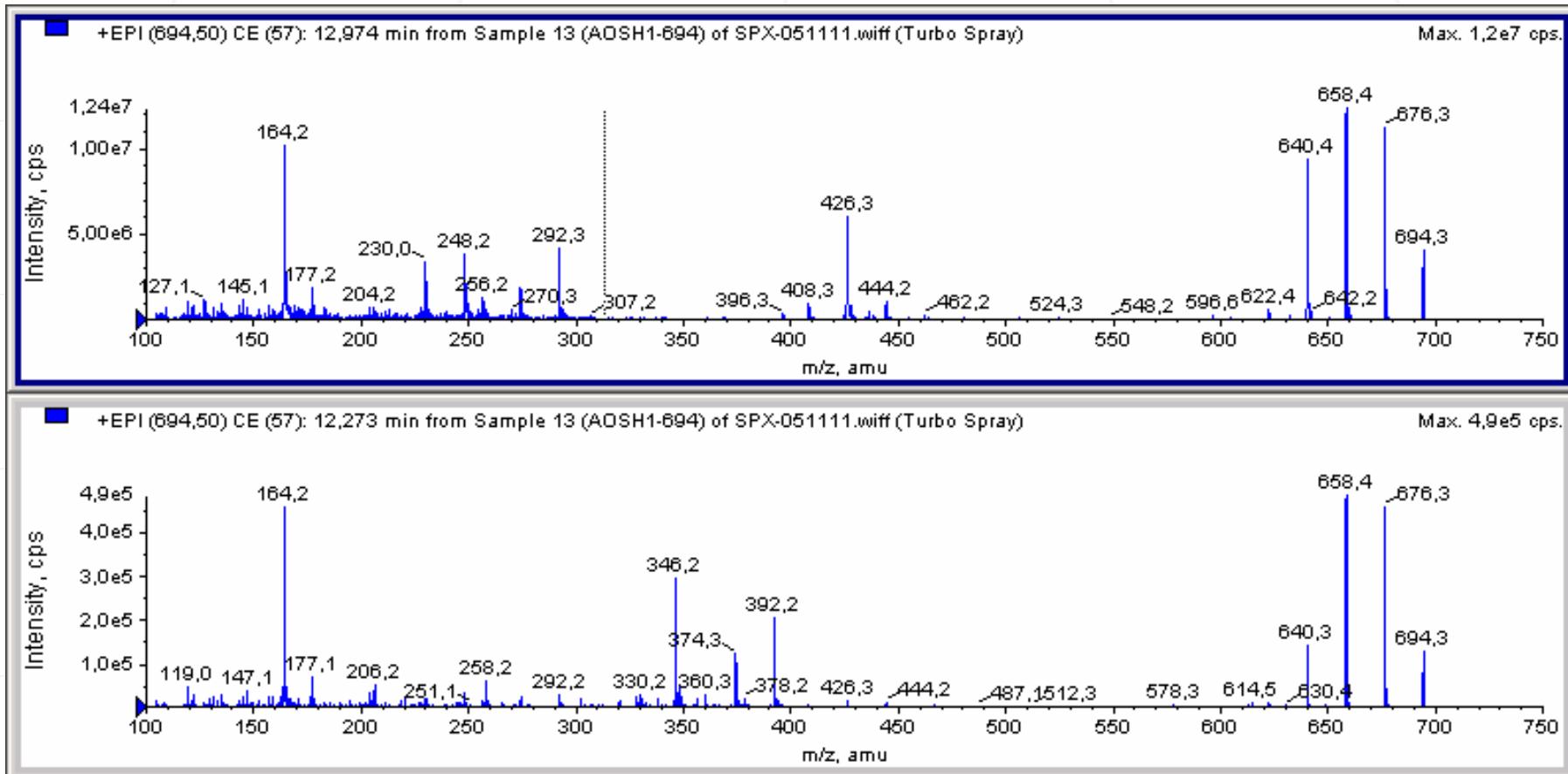


3. Mass spectral characterization

AOSH 2: m/z = 692.5

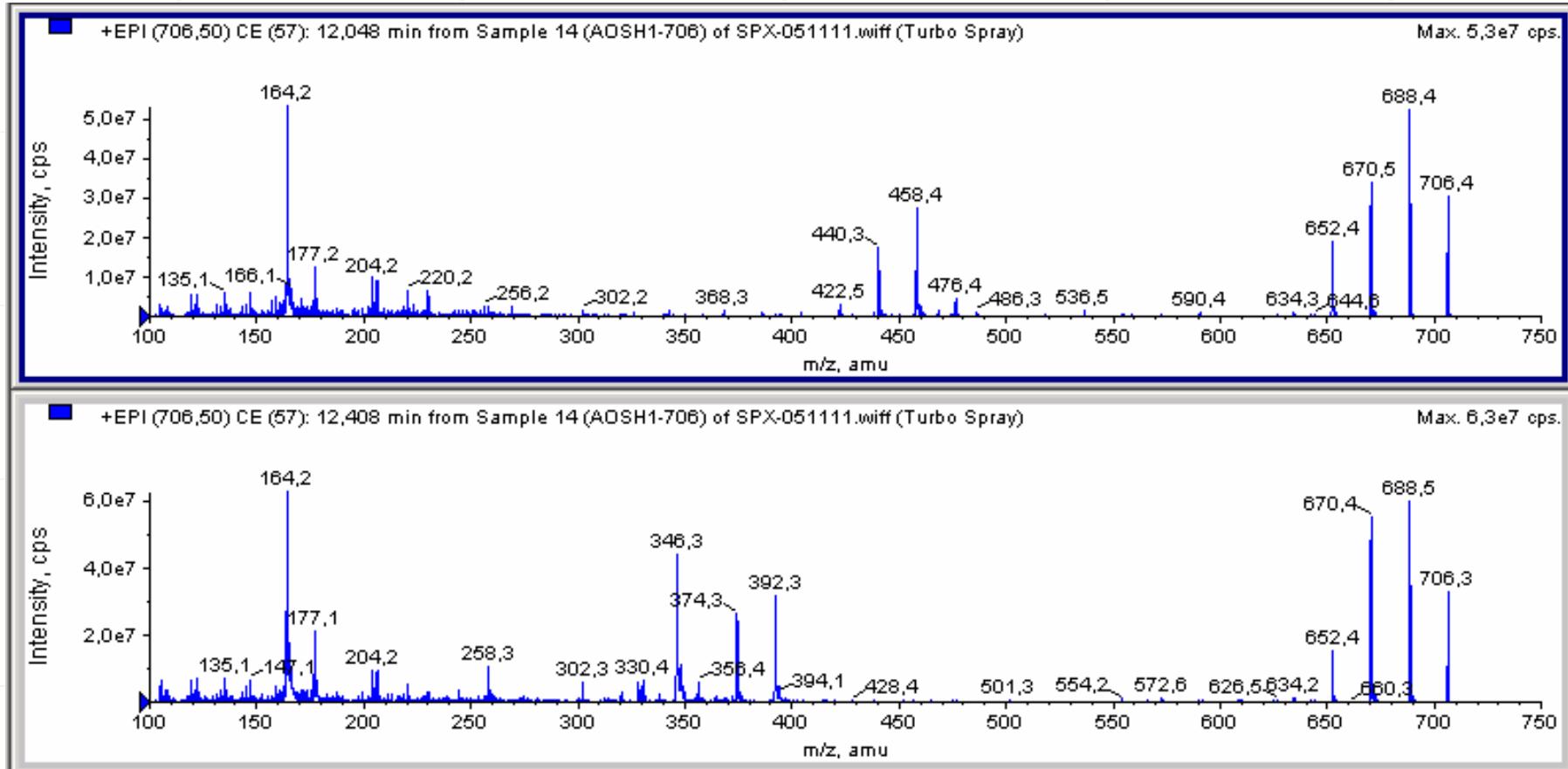


AOSH 2: m/z = 694.5

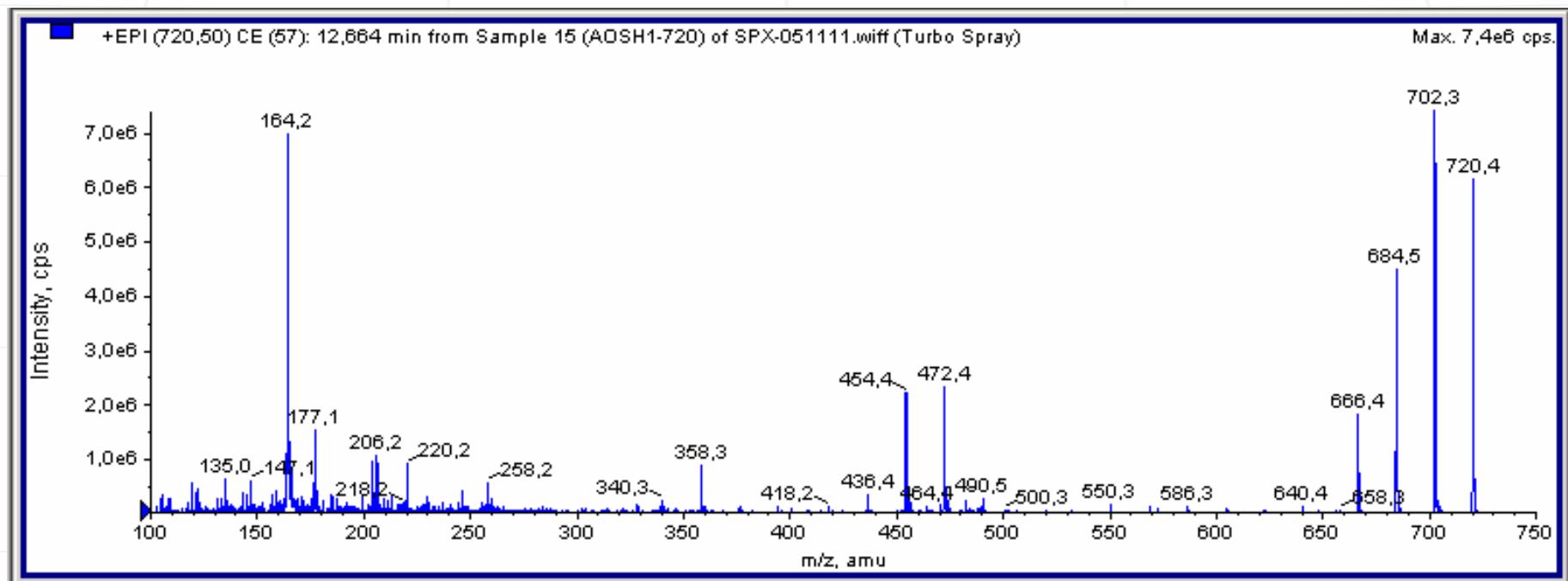


3. Mass spectral characterization

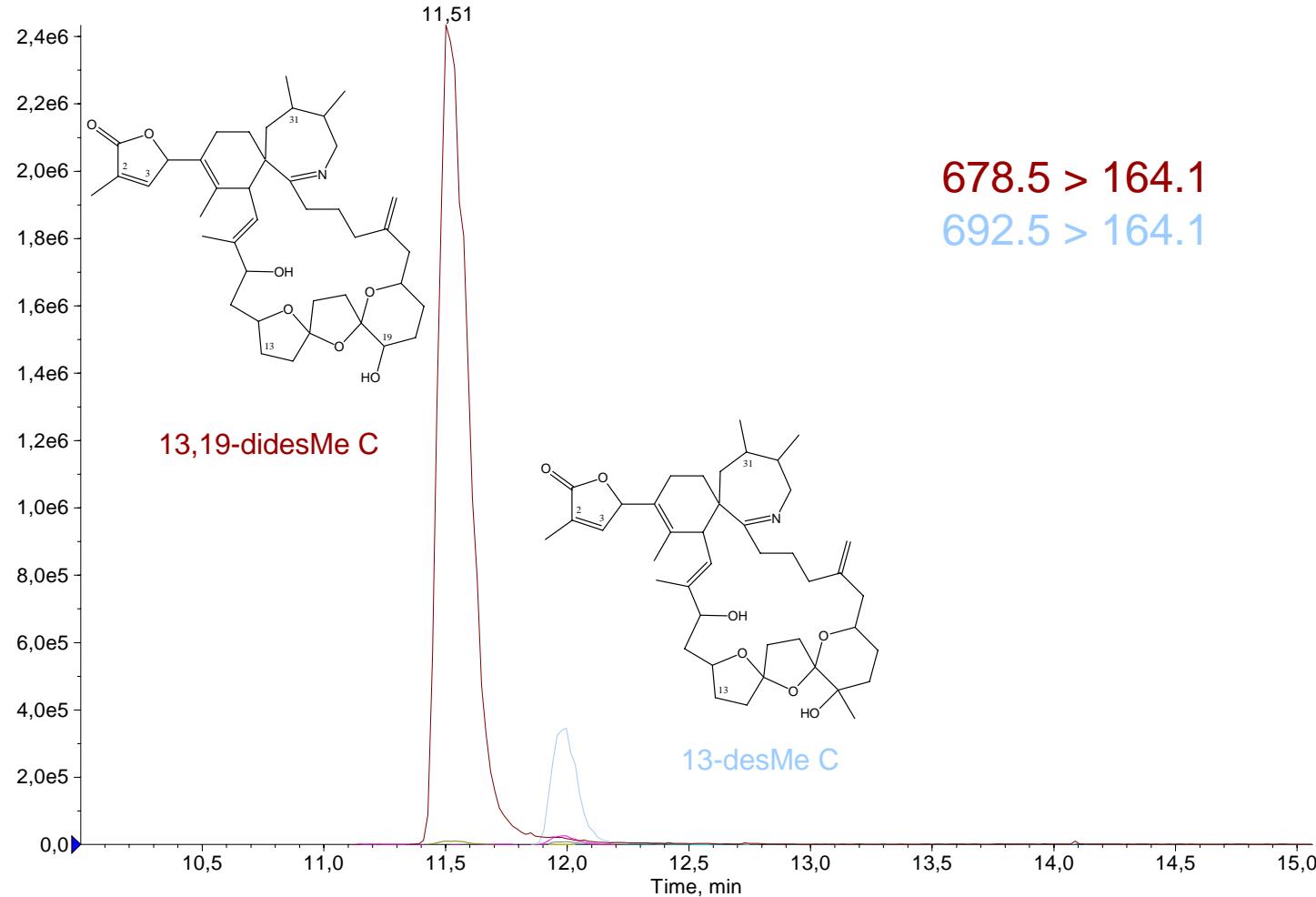
AOSH 2: m/z = 706.5



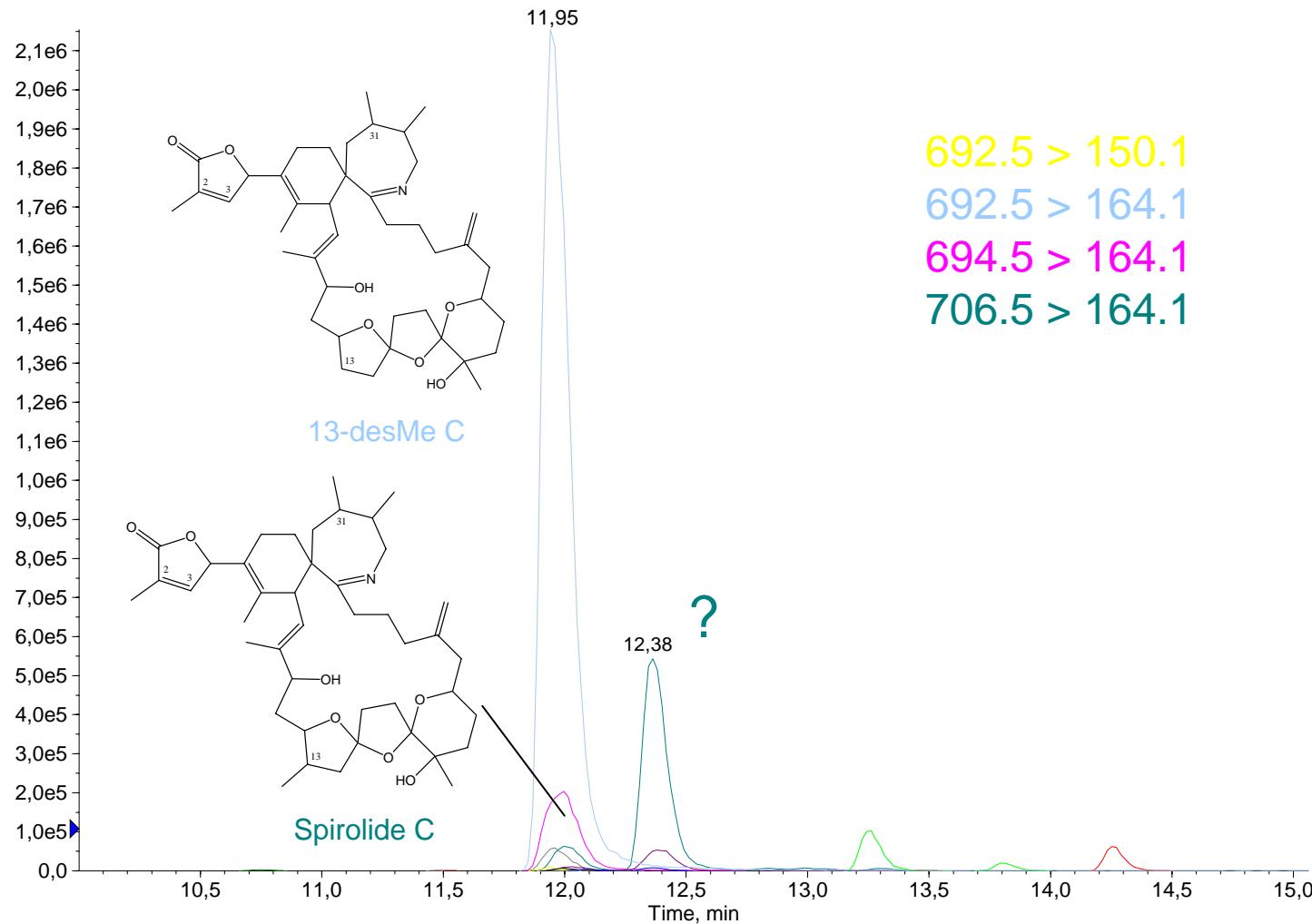
AOSH 2: m/z = 720.5



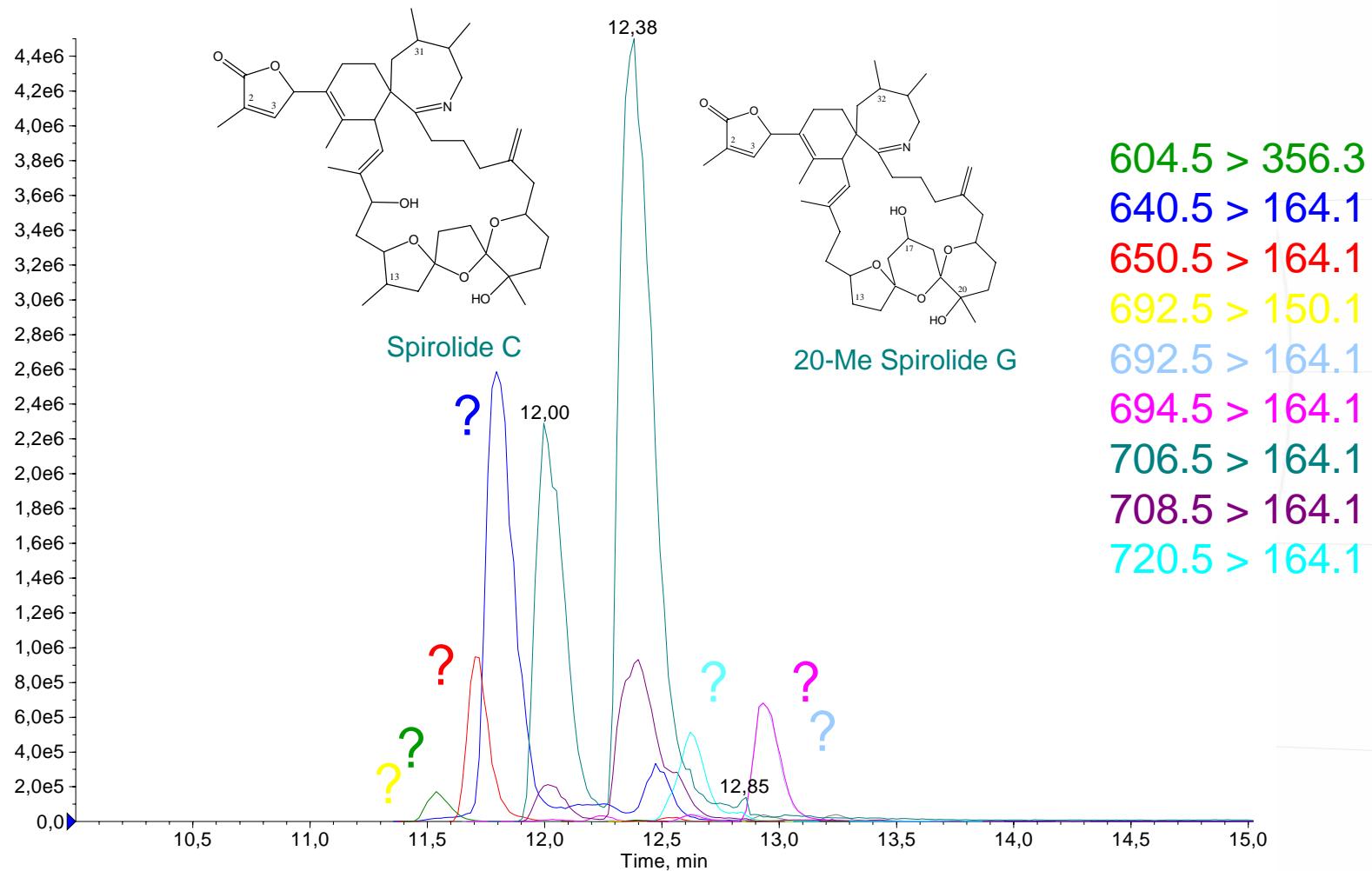
CCMP 1773, Denmark



AOSH 1, Canada

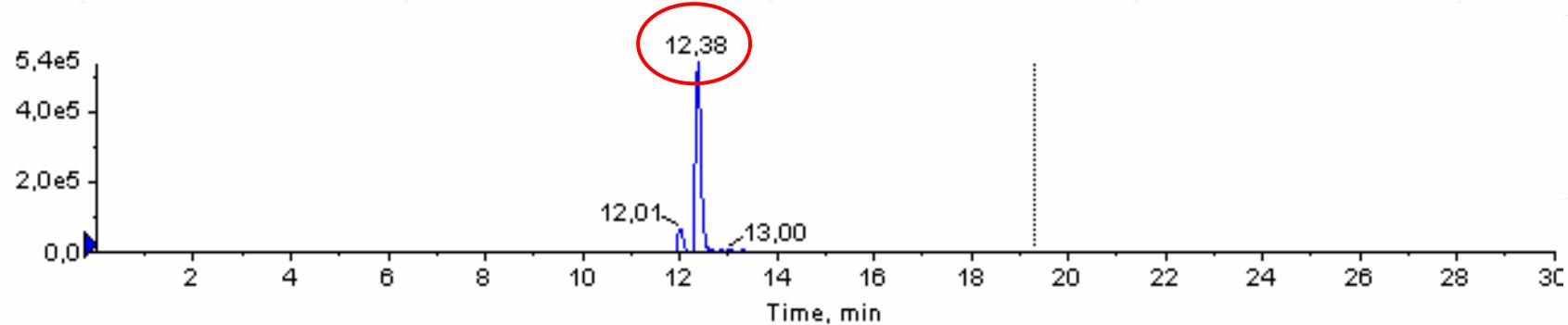


AOSH 2, Canada

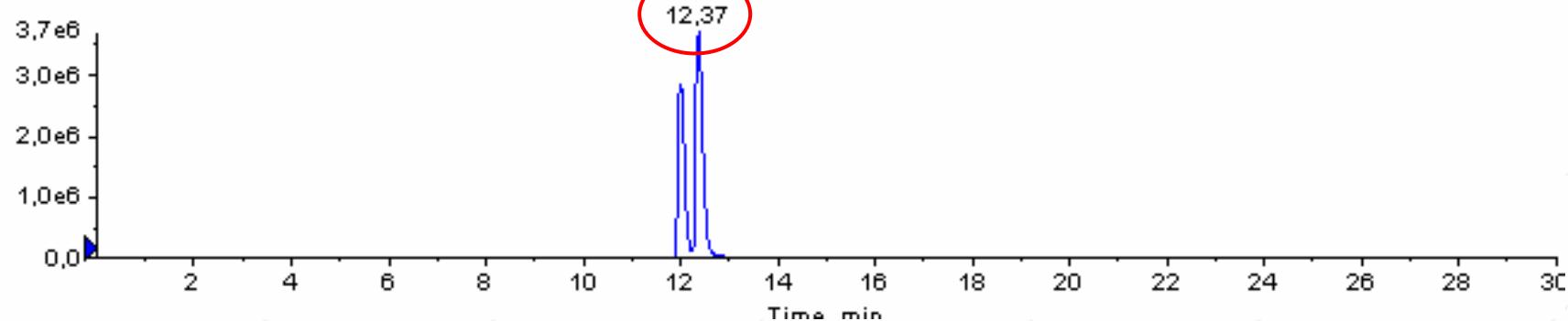


Identical retention time and mass transition,

AOSH 1: MRM 706.5 > 164.1

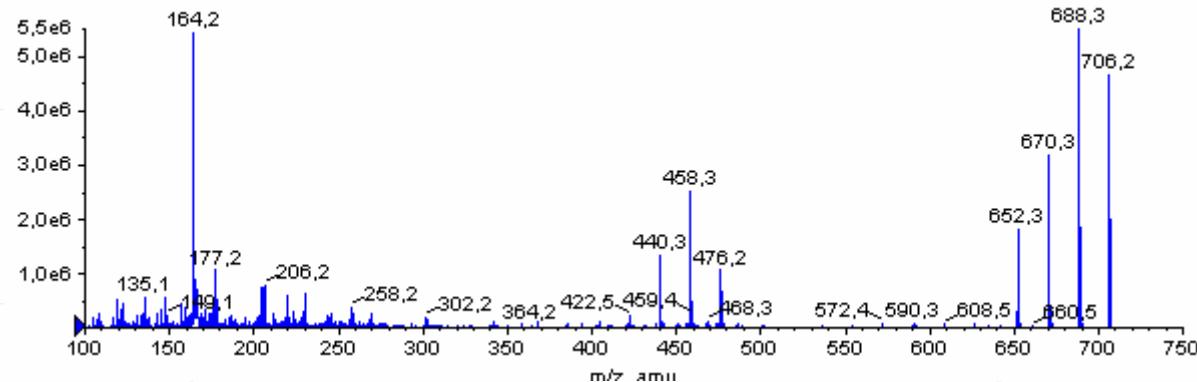


AOSH 2: MRM 706.5 > 164.1

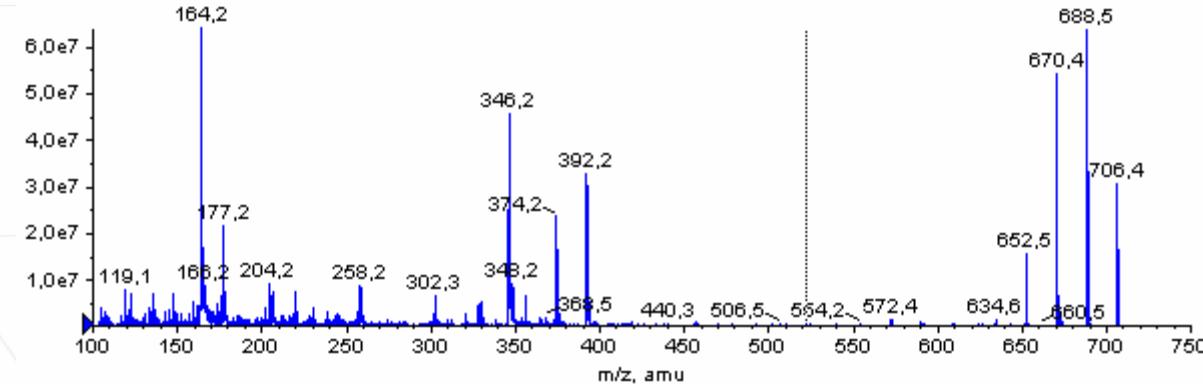


but different mass spectra

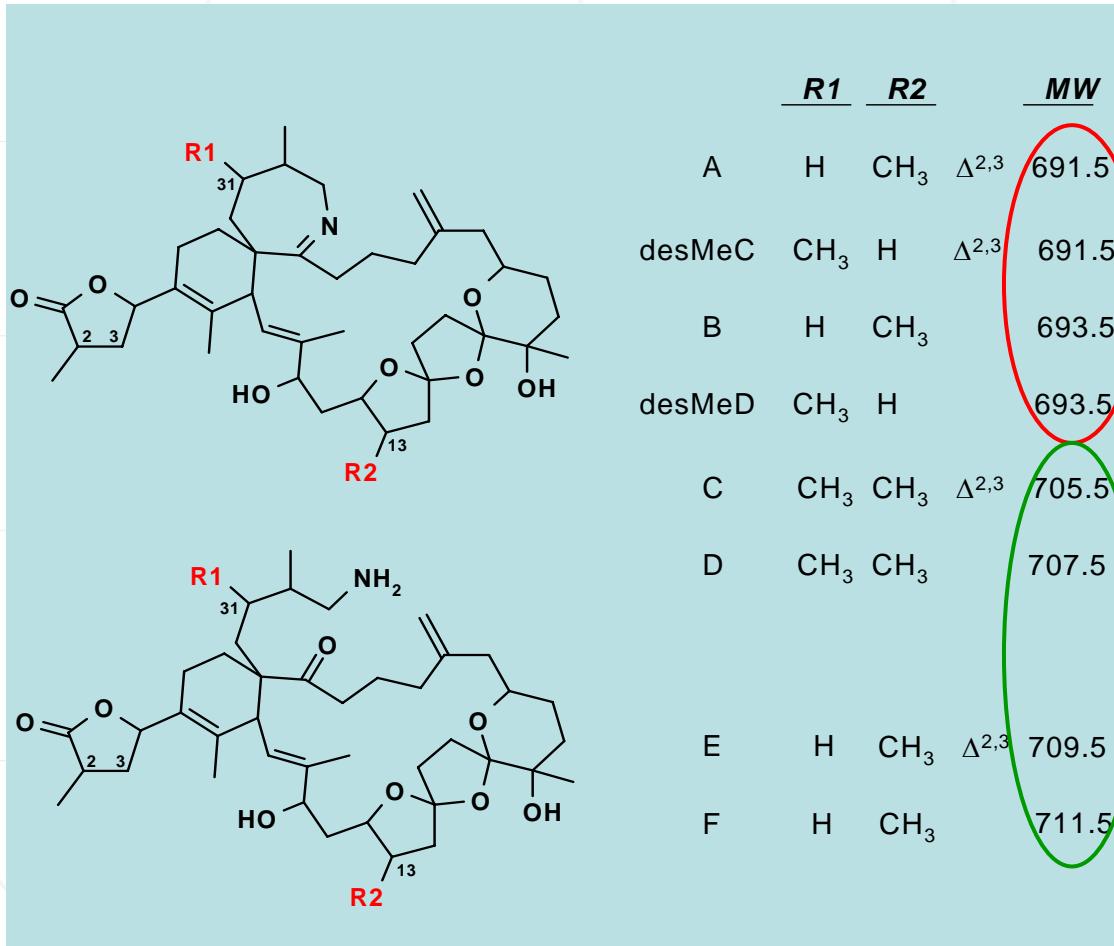
AOSH 1: EPI m/z 706.5; 12.37 min



AOSH 2: EPI m/z 706.5; 12.37 min

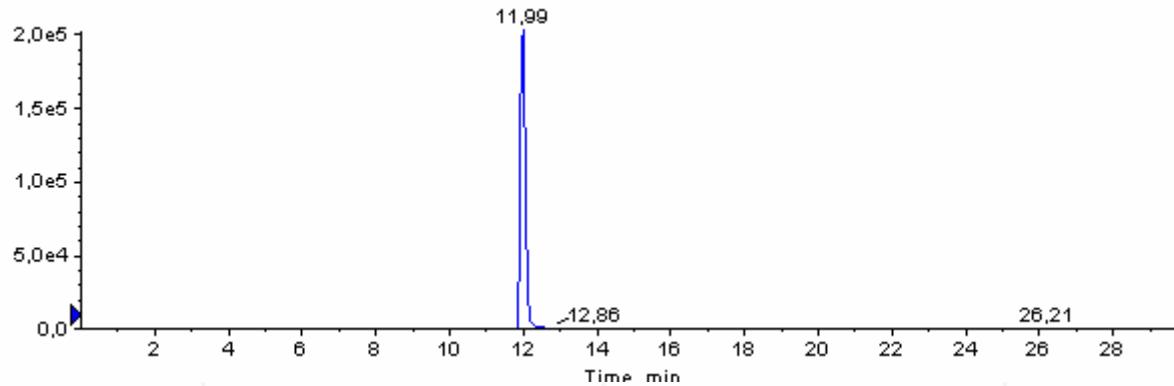


Compound or isotopic peak?

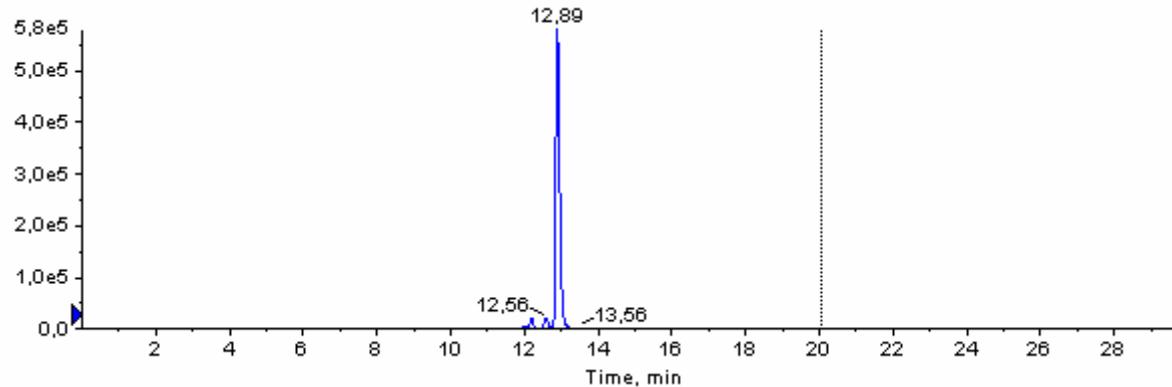


Compound or isotopic peak?

AOSH 1: MRM 694.5 > 164.1

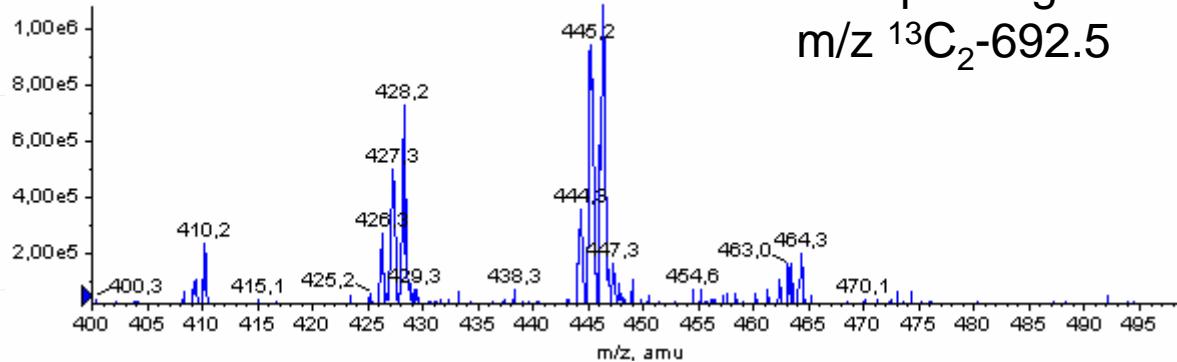


AOSH 2: MRM 694.5 > 164.1



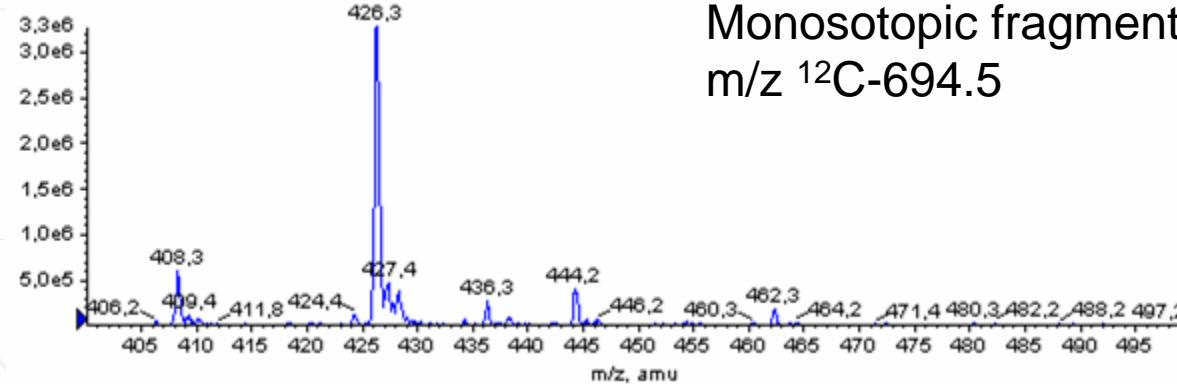
Product ion spectra reveal isotopic pattern

AOSH 1: EPI m/z 694.5; 12.0 min



Isotopic fragments of
 m/z $^{13}\text{C}_2$ -692.5

AOSH 2: EPI m/z 694.5; 12.9 min



Monosotopic fragments of
 m/z ^{12}C -694.5

Conclusions:

Triple quadrupole tandem mass spectrometry is a powerful tool for the determination and quantitation of spirocyclic imides

Unknown toxic spirocyclic imides can be detected in the precursor ion mode of the characteristic cyclo imino fragments at m/z 150 and 164, respectively

Structural information can be obtained by product ion spectra of parent ions

Co-eluting compounds with identical mass transitions can be differentiated by their product ion spectra

Product ion spectra can be used to differentiate between isotope statellites and monoisotopic peaks

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Thank You!